

Measurements and Emissions

large variety of measurement techniques for atmospheric trace are available:

- **Gas chromatography (GC, universal technique, in-situ)**
- **Optical spectroscopy (universal technique, in-situ, remote sensing)**
- **Chemiluminescence (e.g. for the detection of NO or O₃)**
- **Photoacoustic detection**
- **Electrochemical techniques**

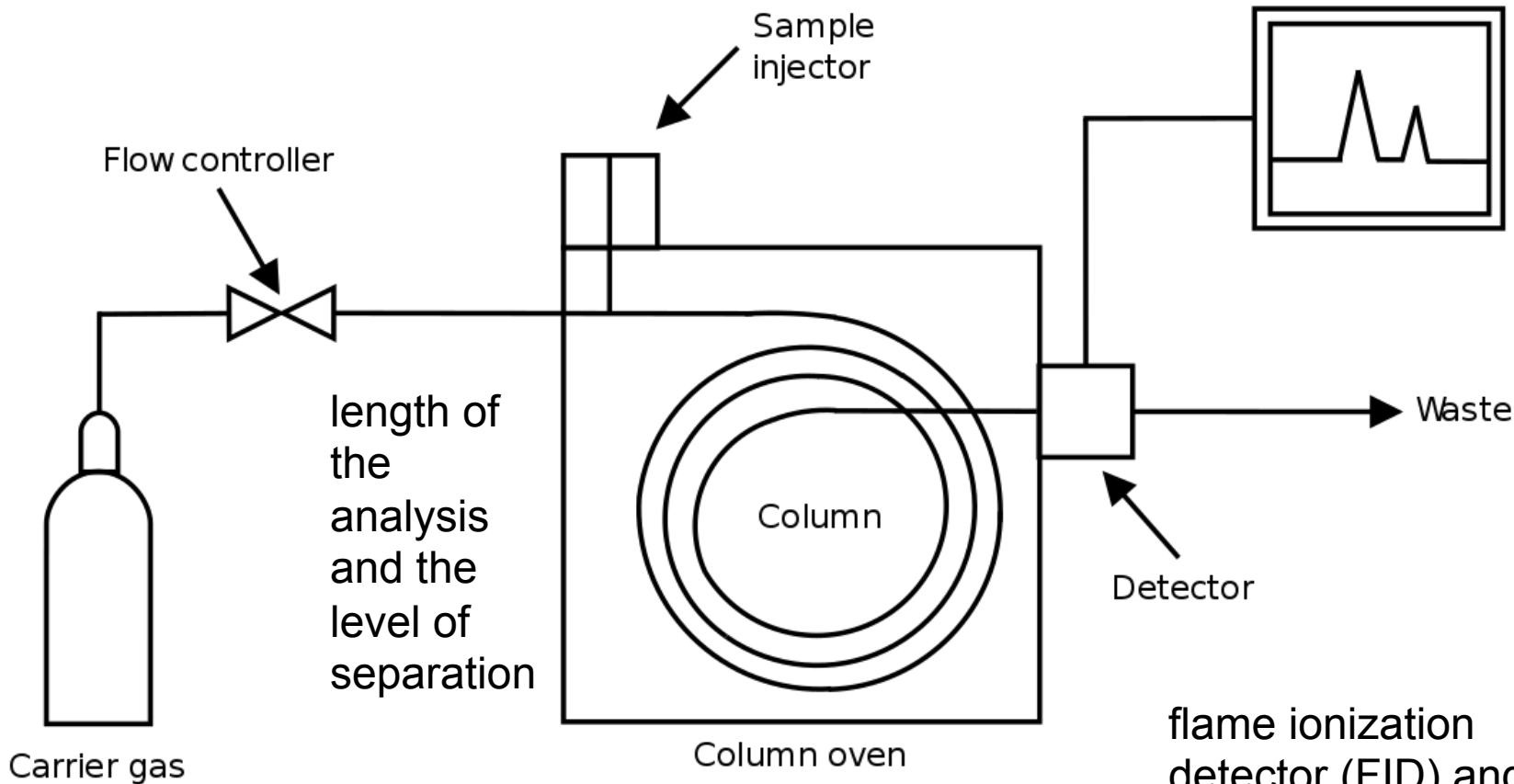
Spectroscopic techniques are a powerful variety.

These techniques are highly sensitive, very specific, universally usable, give absolute results, and have the potential for remote sensing.

Gas chromatography

separating the different components of a mixture

Retention time

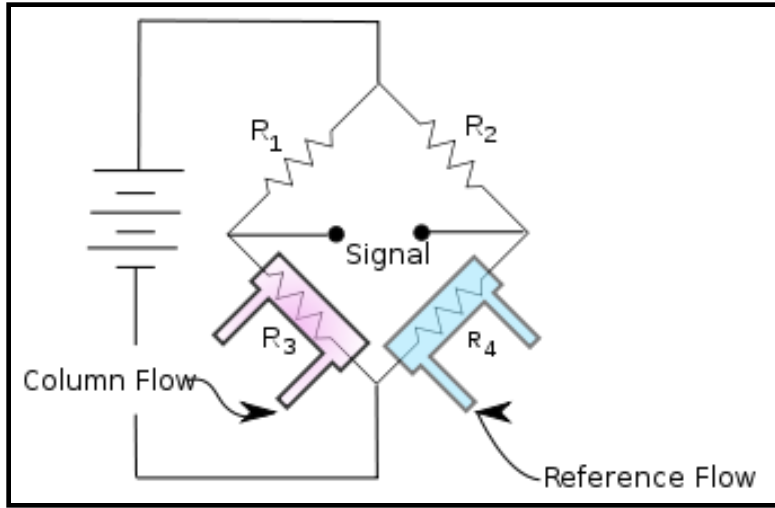


Mobile is a carrier gas, inert gas such as helium unreactive gas such as nitrogen

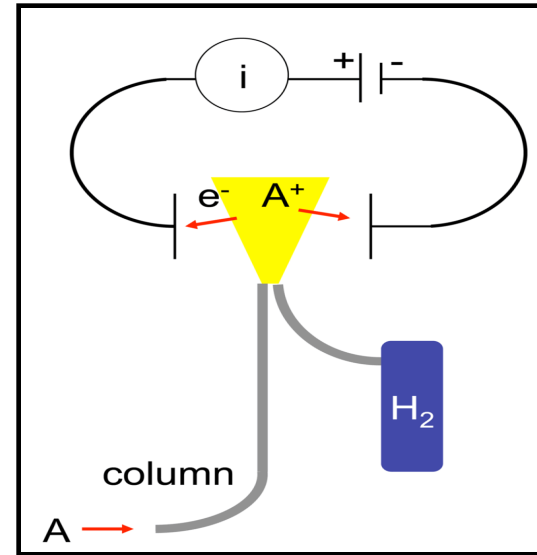
stationary phase microscopic layer of polymer on an inert solid support, inside a piece of metal tubing called a column

flame ionization detector (FID) and the thermal conductivity detector (TCD)

thermal conductivity detector (TCD)

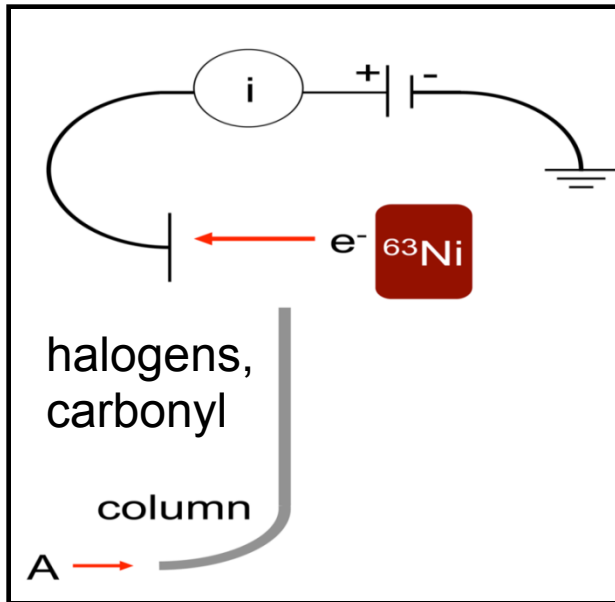


flame ionization detector (FID)



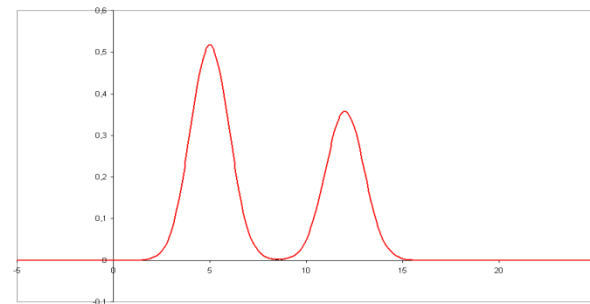
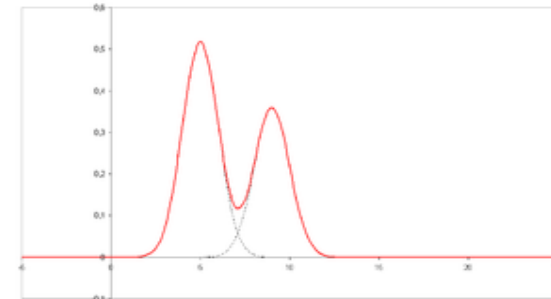
Mostly for
Hydrocarbons

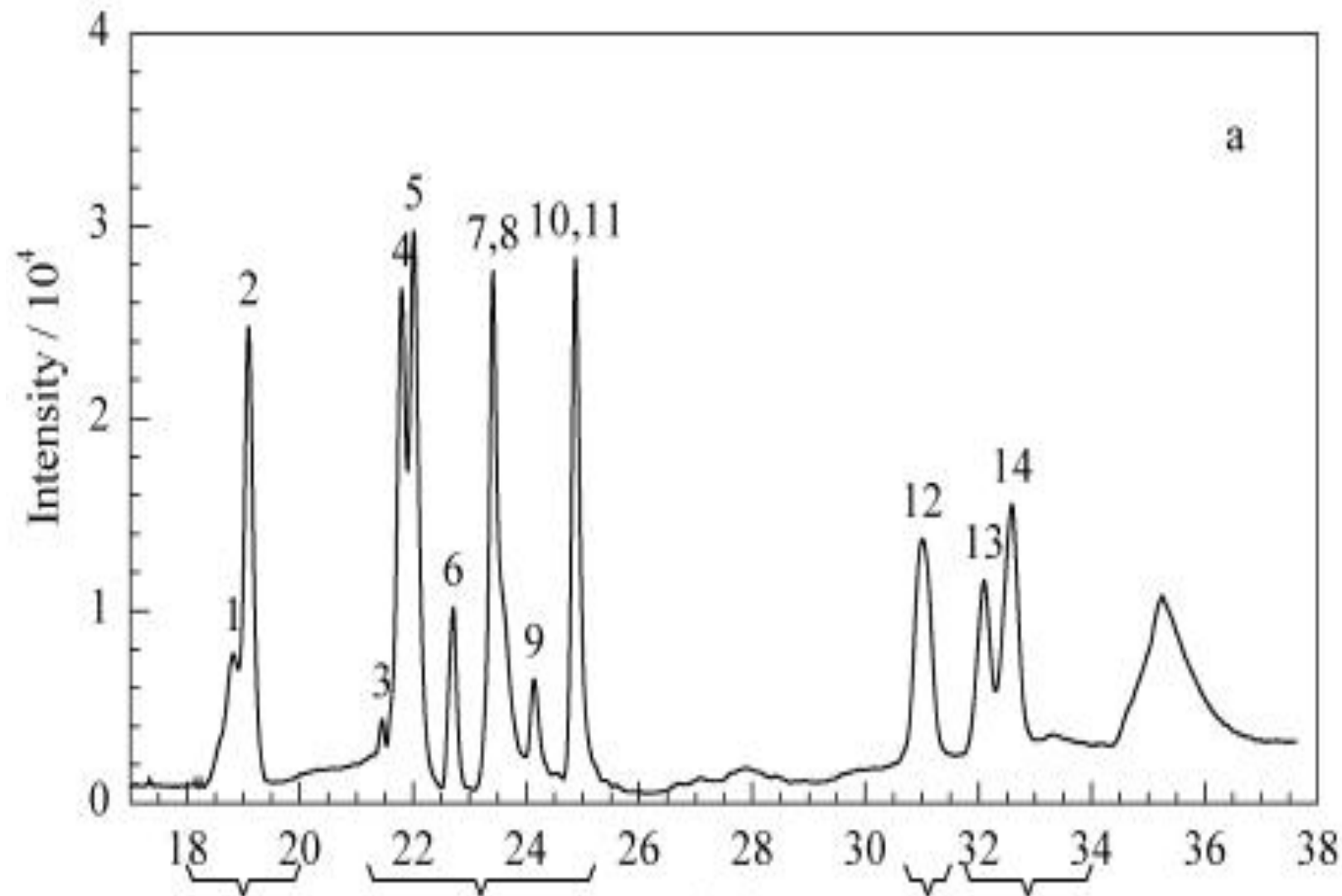
Electron capture detector (ECD)



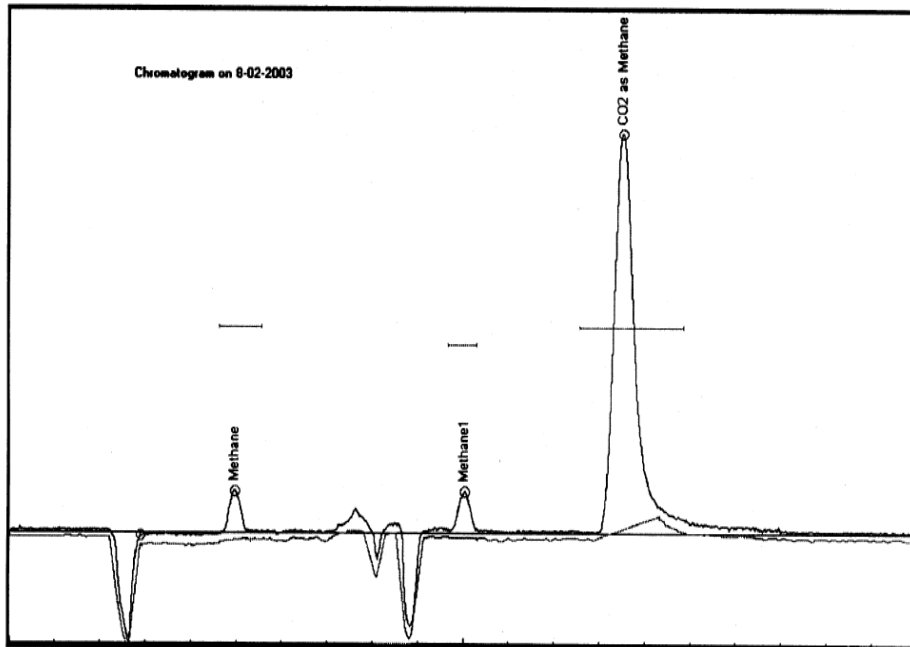
Makeup gas
 N_2
generates
background
current

Chromatogram

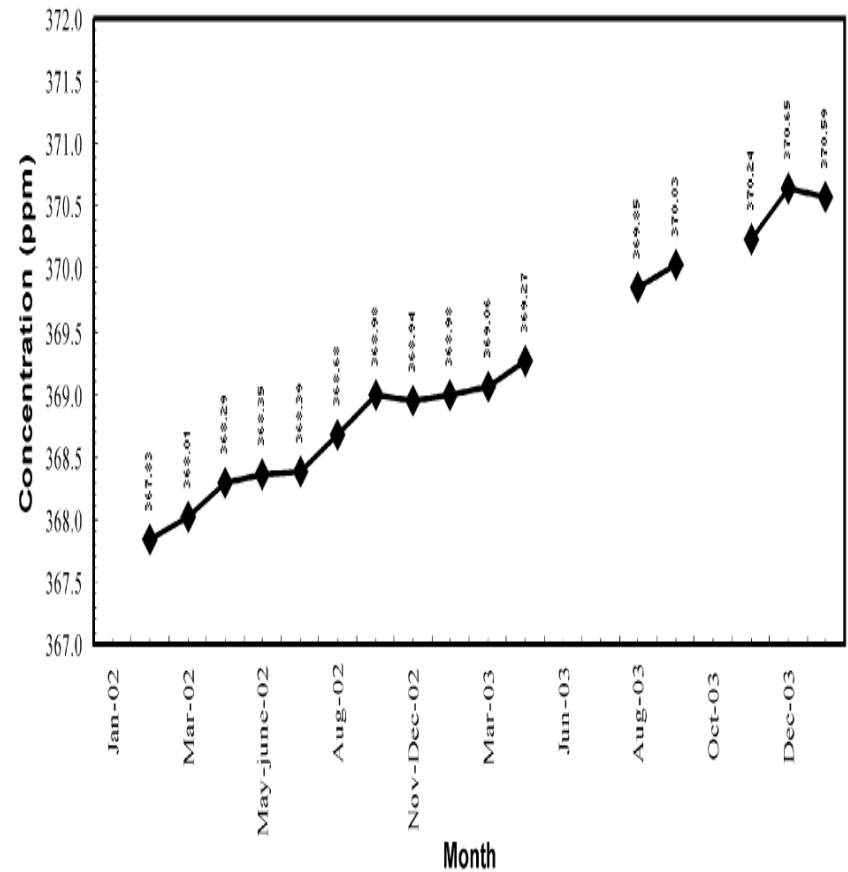




Typical Methane CO and CO₂ Chromatogram observed at Antarctica



Area of peak gives information about the concentration



Optical Measurement of Atmospheric Trace Species

Physical Process

Absorption

Emission

Thermal

Induced

Wavelength
molec. physics

UV/visible
(electronic)

IR
(vibrat.)

μ -Wave
(rotational)

Chemical

LASER

Light Path
arrangement

Single/double
Path

Multiple folded
Path

Radiation
transport

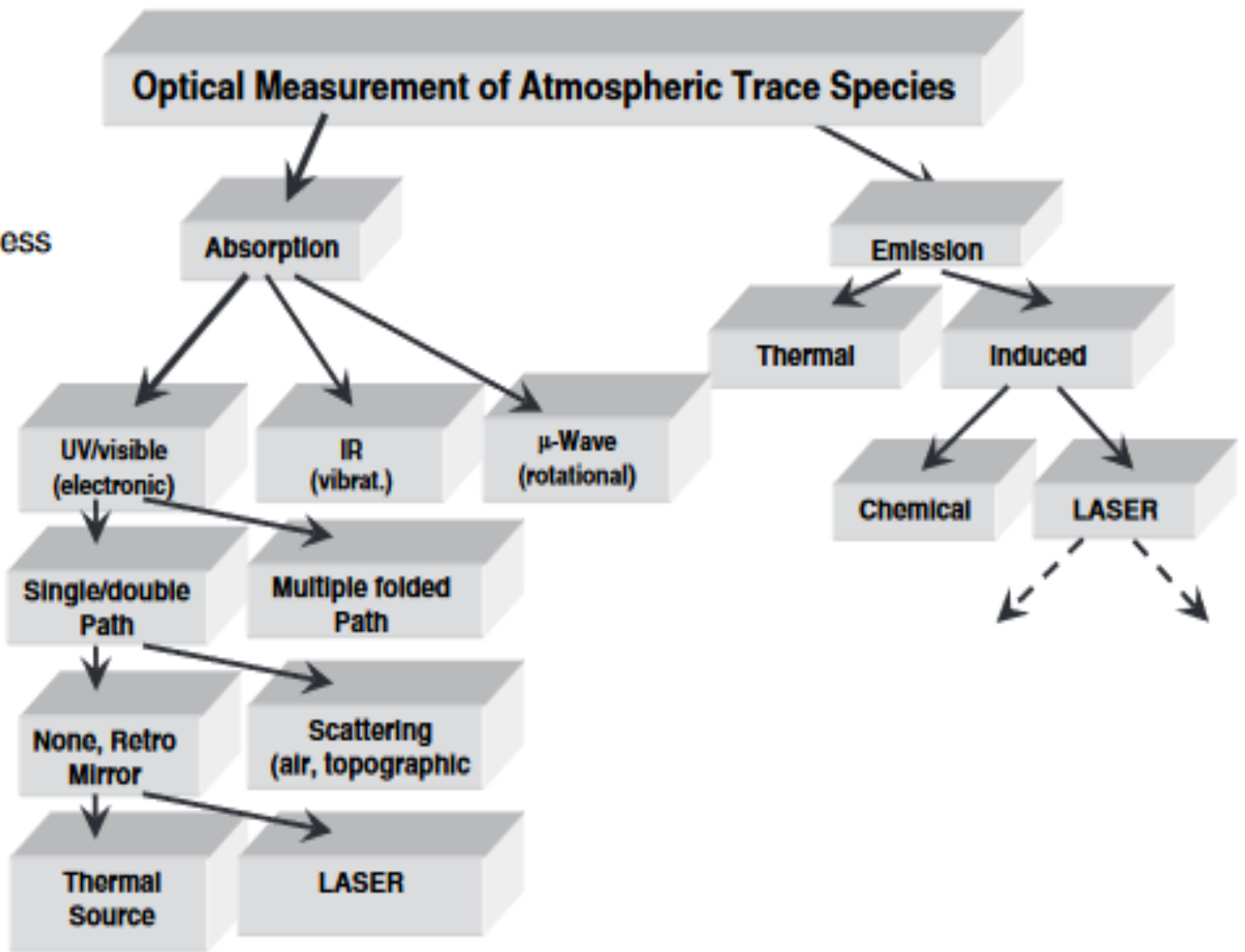
None, Retro
Mirror

Scattering
(air, topographic)

Light source

Thermal
Source

LASER

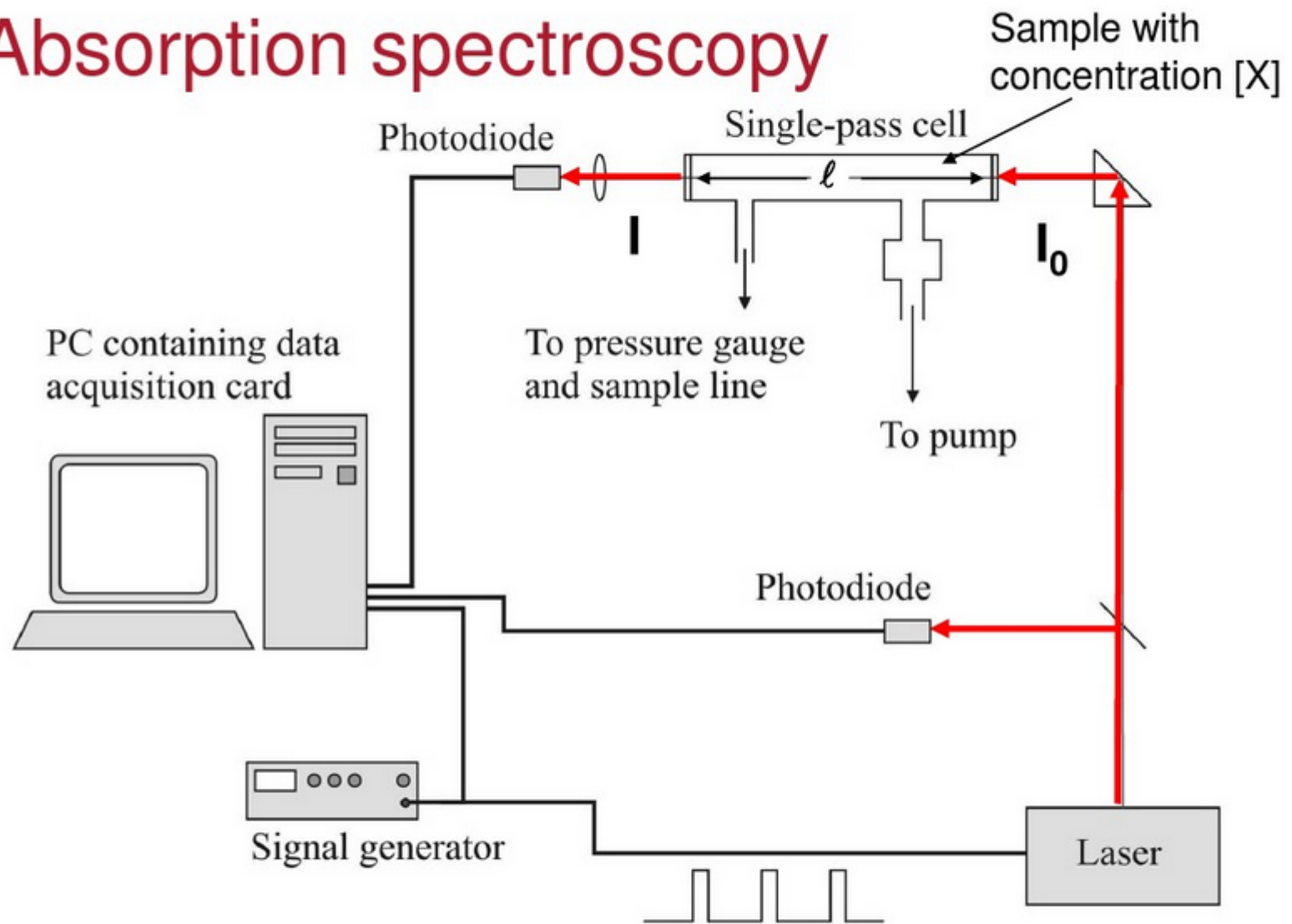


The important technical criteria of spectroscopic instruments

- (1) are the wave-length region used
- (2) the physical principle (i.e. absorption or emission spectroscopy),
- (3) the arrangement of the light path (path in the open atmosphere or enclosed – frequently folded – path)
- (4) or the type of light source used.

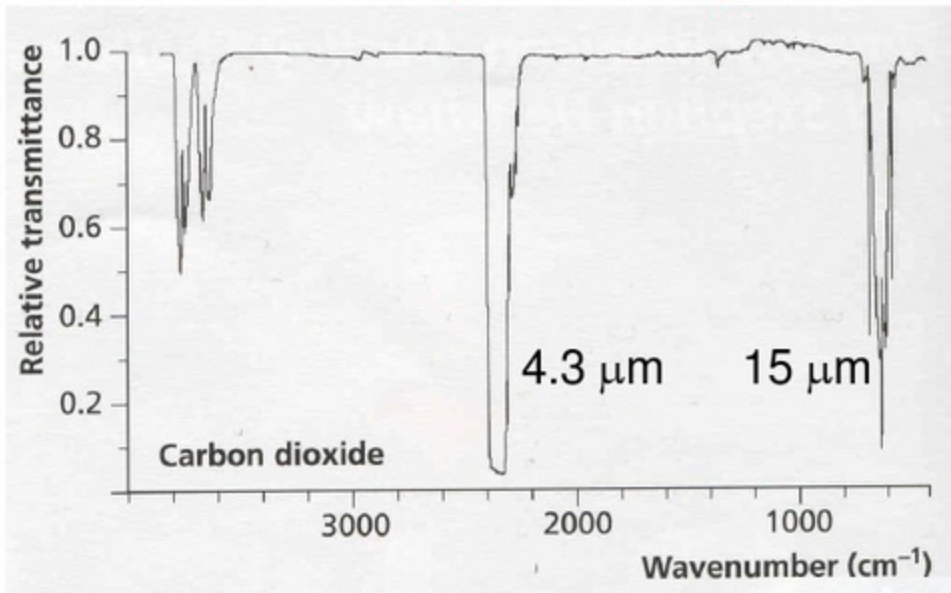
- Tunable Diode Laser Spectroscopy (TDLS)
- Cavity Enhanced Absorption Spectroscopy (CESA)
- Photo Acoustic Spectroscopy (PAS)
- Light Detection And Ranging (LIDAR)
- Differential Absorption LIDAR (DIAL)
- Laser-Induced Fluorescence (LIF)
- Cavity-Ringdown Spectroscopy (CRD)
- Cavity Attenuated Phase Shift (CAPS)
- Differential Optical Absorption Spectroscopy (DOAS)

Absorption spectroscopy



Beer-Lambert Law: $I = I_0 \exp\{-\sigma [X] l\}$

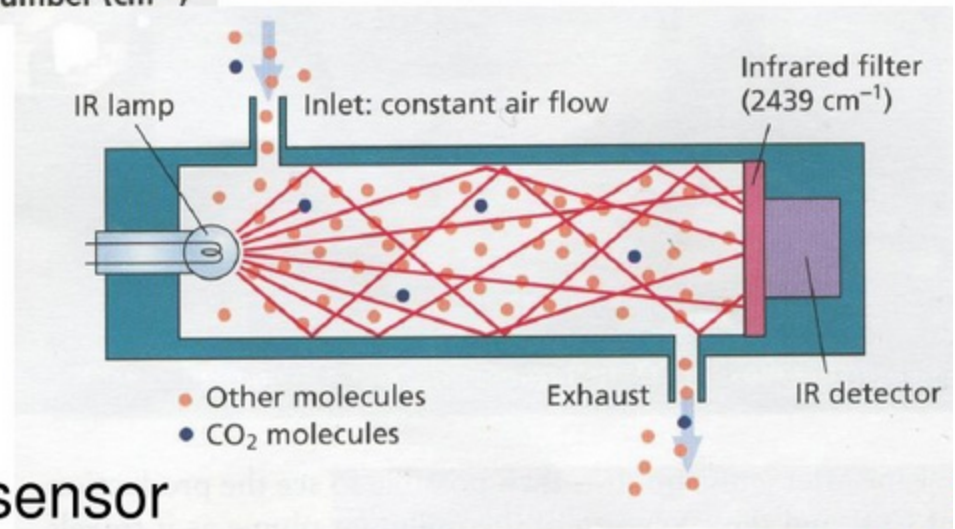
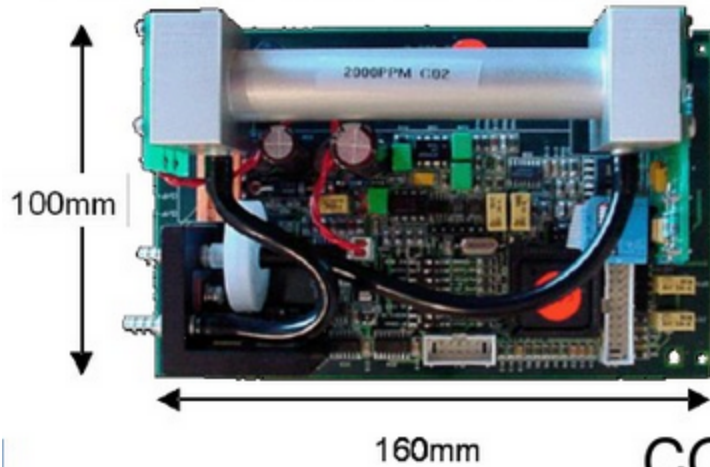
1. Absorption spectroscopy



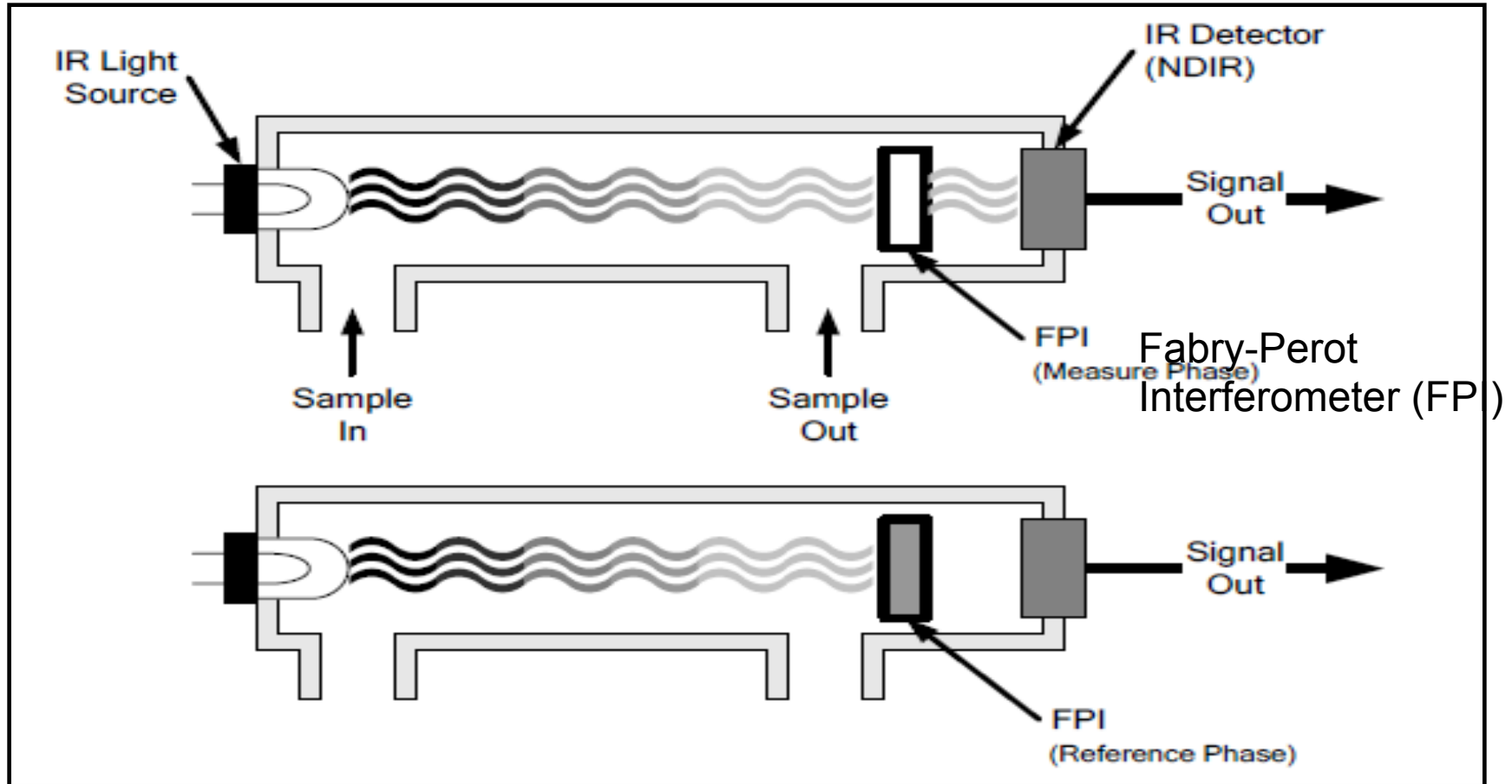
Bending
Vibration
 $\lambda = 15 \mu\text{m}$



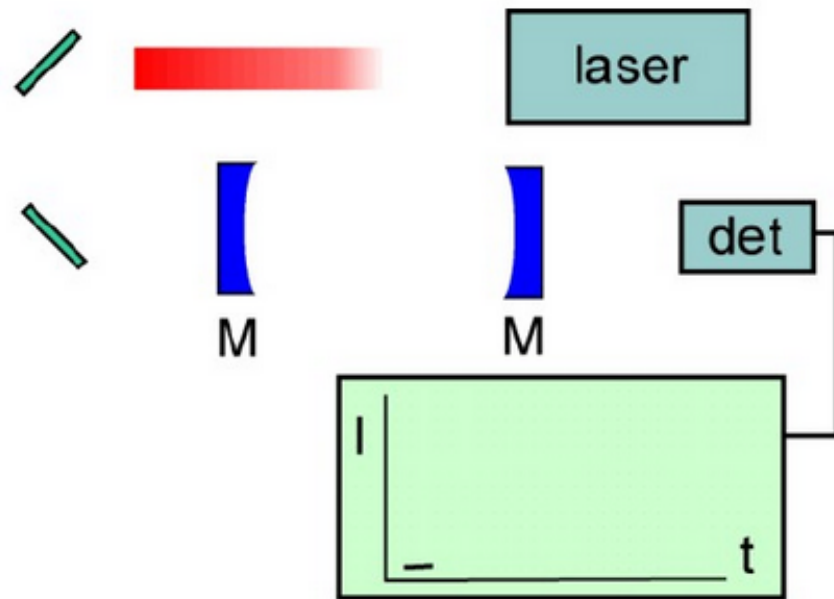
Asymmetric
Stretch
 $\lambda = 4.3 \mu\text{m}$



Infrared CO₂ analyzer

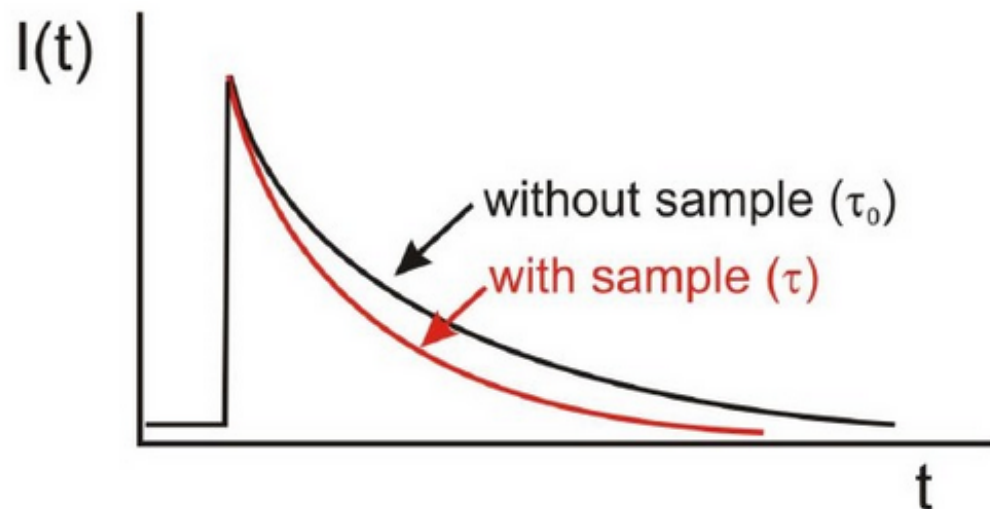
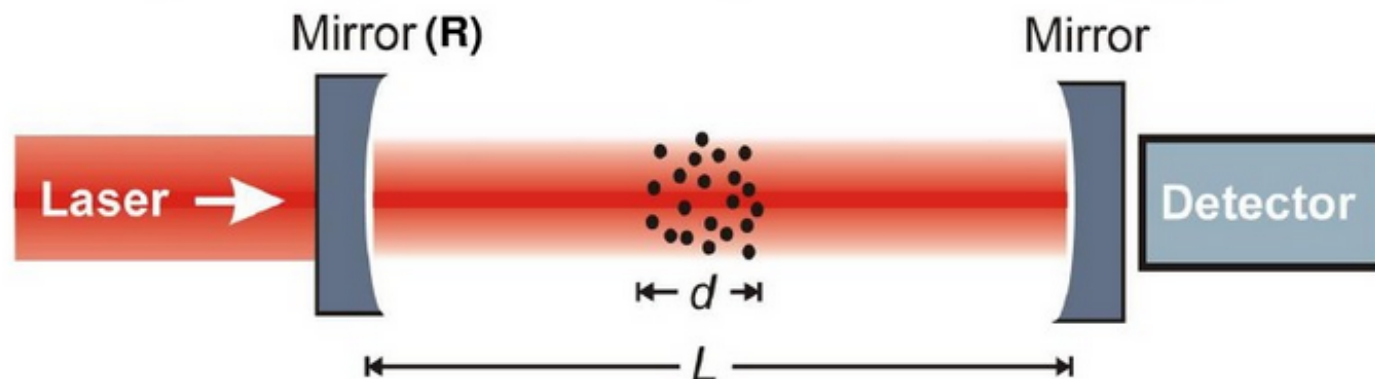


🔥 2. Cavity ring-down spectroscopy



- Trap a pulse of light between two high-reflectivity mirrors
- Place the gas sample between the mirrors (in the cavity)
- Measure the **lifetime** of the light pulse in the cavity – depends on concentration of sample

🔥 Cavity ring-down spectroscopy



For an empty cavity:

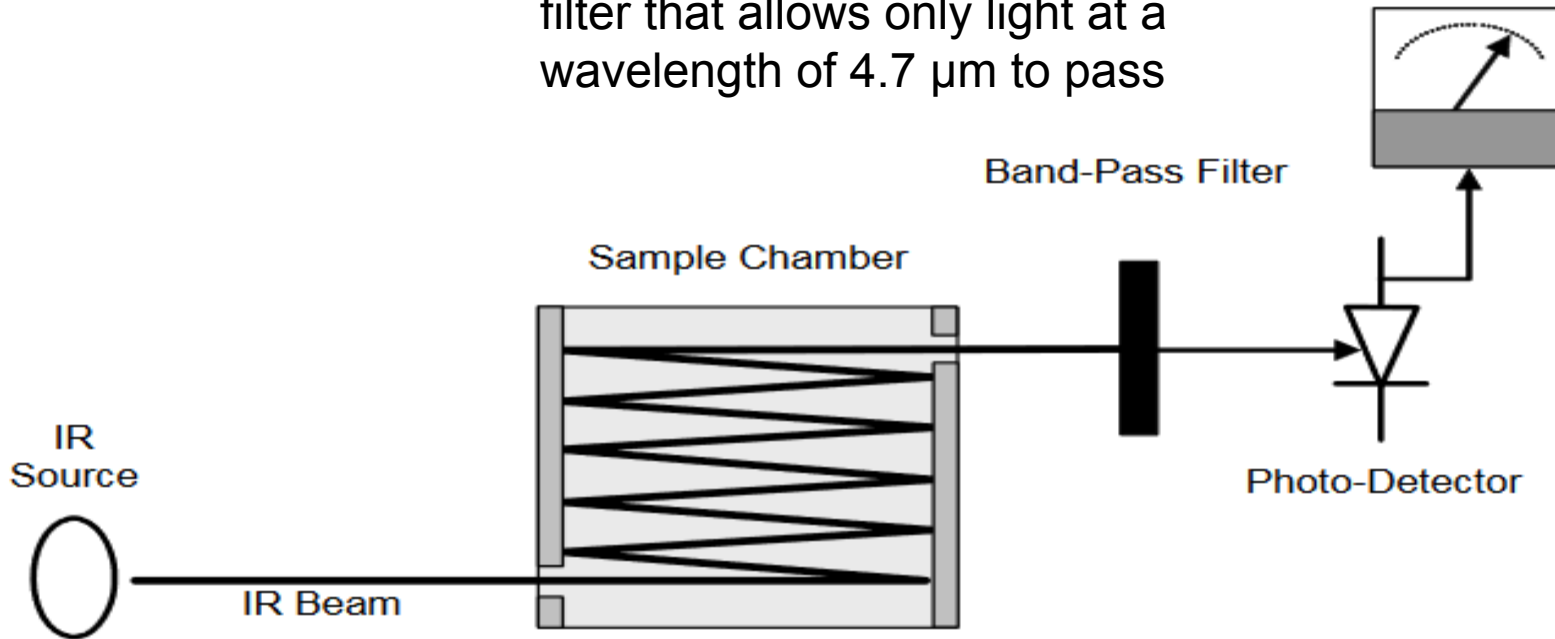
$$\tau_0 = \frac{L}{c(1-R)}$$

With an absorber:

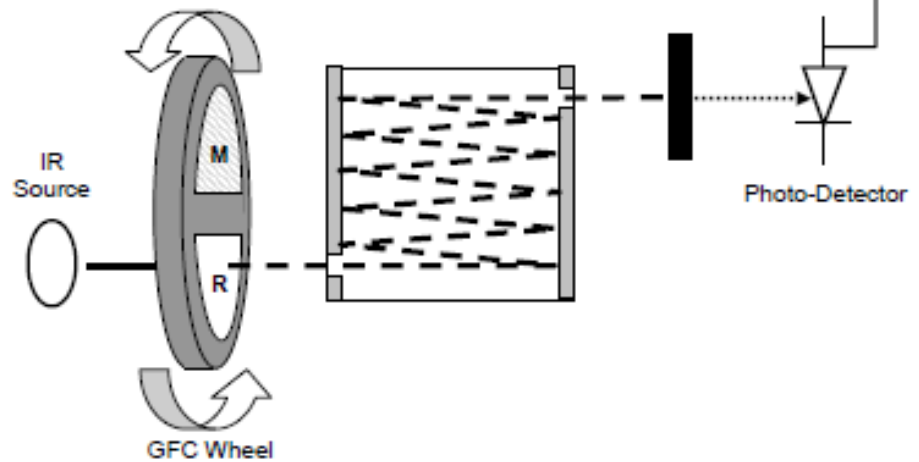
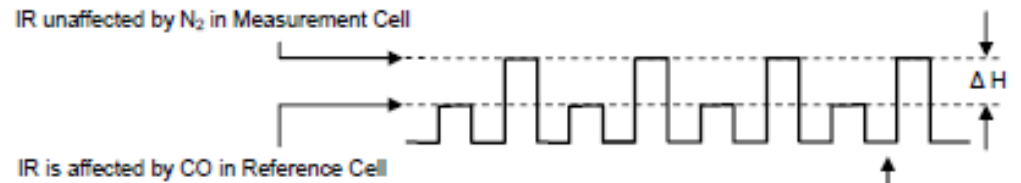
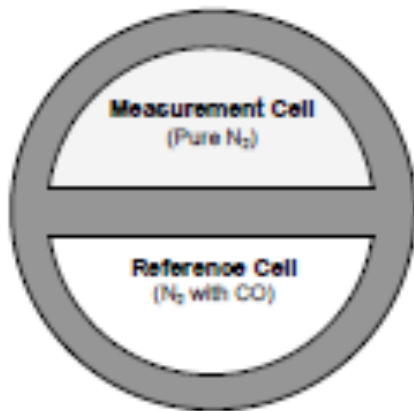
$$\frac{1}{\tau} - \frac{1}{\tau_0} = c[X]\sigma \frac{d}{L}$$

Measurement of Carbon Monoxide

beam shines through a band-pass filter that allows only light at a wavelength of $4.7\ \mu\text{m}$ to pass



high-energy heated element to generate a beam of broad-band IR light

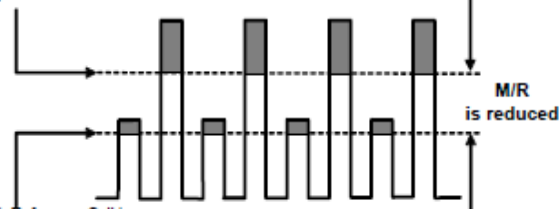


IR unaffected by N_2 in Measurement Cell of the GDC Wheel and no additional CO in the Sample Chamber

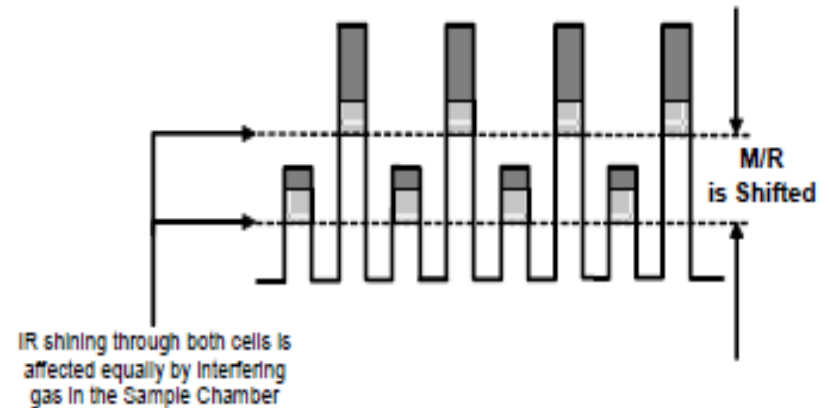


IR affected by CO in Reference Cell with no interfering gas in the Sample Chamber

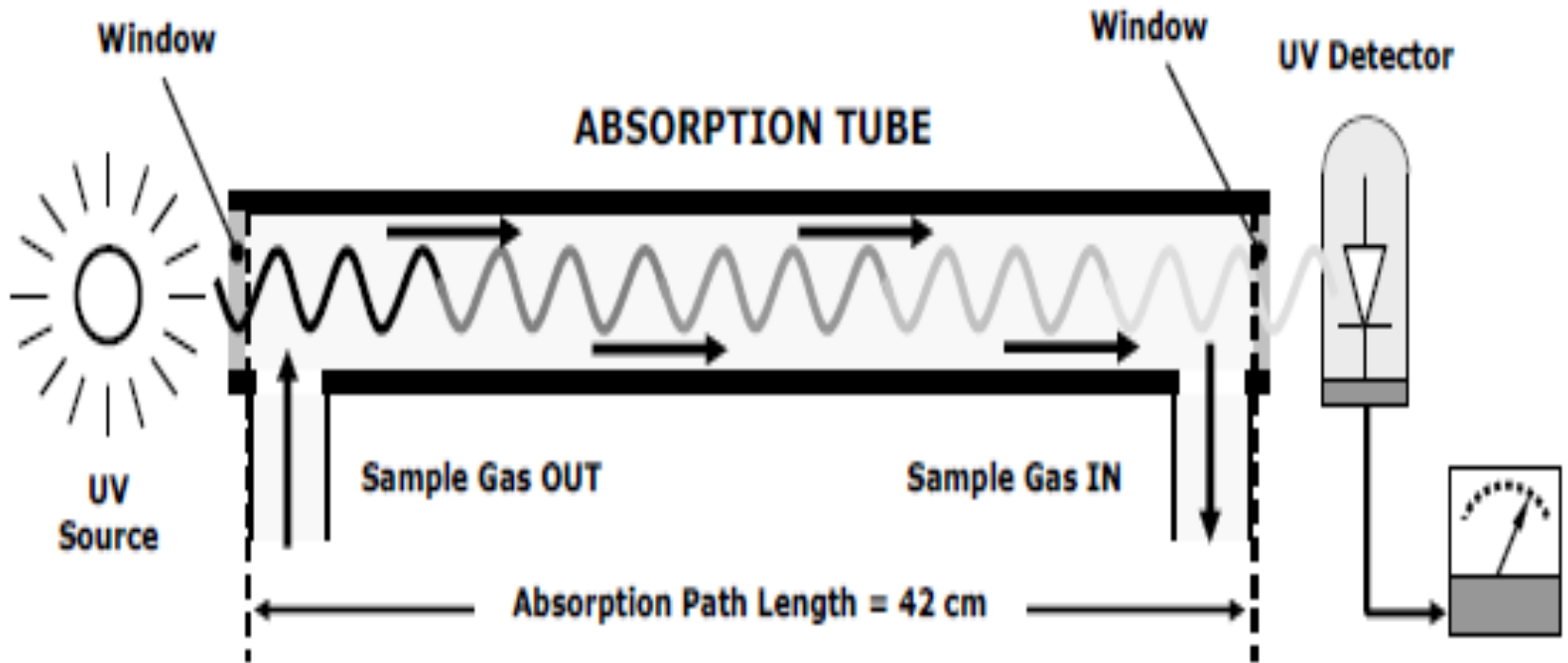
IR shining through Measurement Cell of the GDC Wheel is reduced by additional CO in the Sample Chamber



IR shining through Reference Cell is also reduced by additional CO in the Sample Chamber, but to a lesser extent

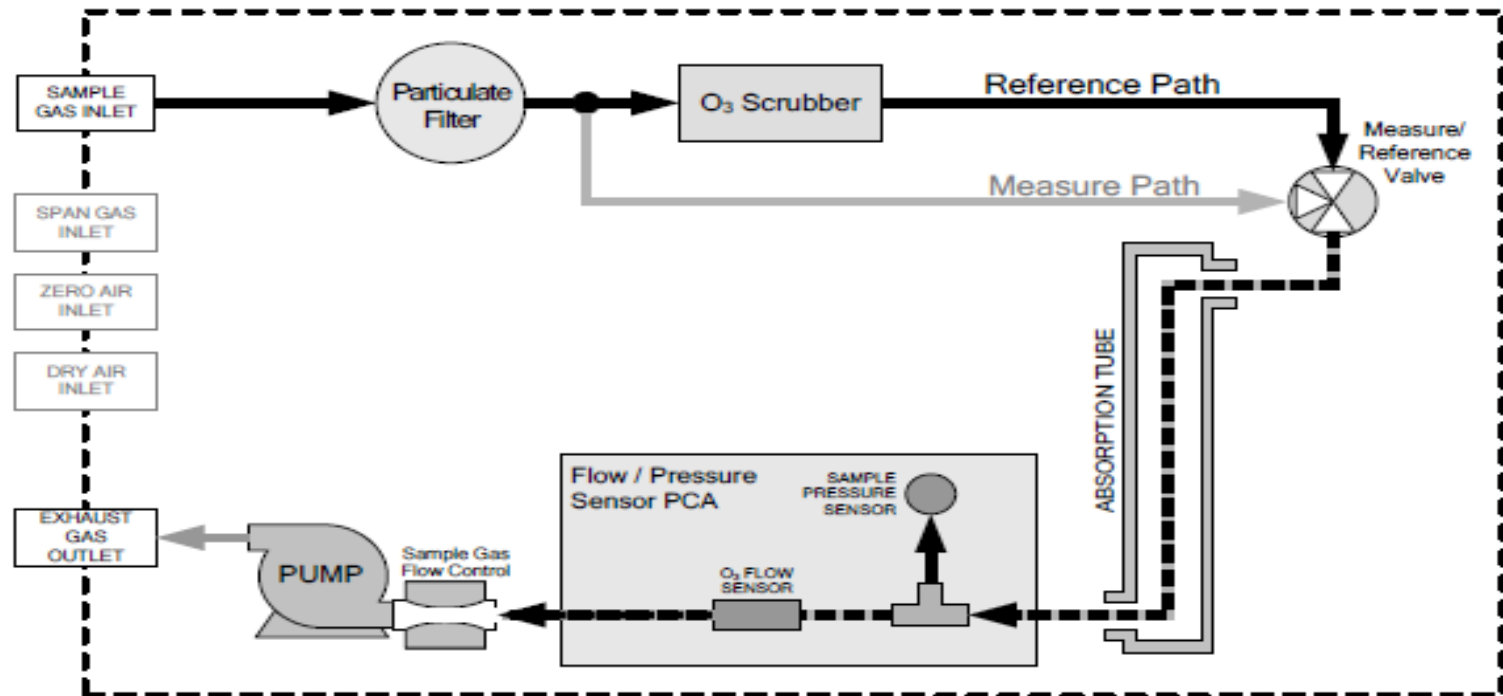


Measurement of Ozone



uses a high energy, mercury vapor lamp to generate a beam of UV light 254nm

Measurement of Ozone



The Measurement / Reference Cycle consists of:

TIME INDEX	STATUS
0 seconds	Measure/Reference Valve Opens to the Measure Path.
0 – 2 seconds	Wait Period. Ensures that the Absorption tube has been adequately flushed of any previously present gasses.
2 – 3 seconds	Analyzer measures the average UV light intensity of O ₃ bearing Sample Gas (I) during this period.
3 seconds	Measure/Reference Valve Opens to the Reference Path.
3 – 5 seconds	Wait Period. Ensures that the Absorption tube has been adequately flushed of O ₃ bearing gas.
5 – 6 seconds	Analyzer measures the average UV light intensity of Non-O ₃ bearing Sample Gas (I ₀) during this period.
CYCLE REPEAT EVERY 6 SECONDS	

The Cavity Attenuated Phase Shift (CAPS) NO₂ monitor

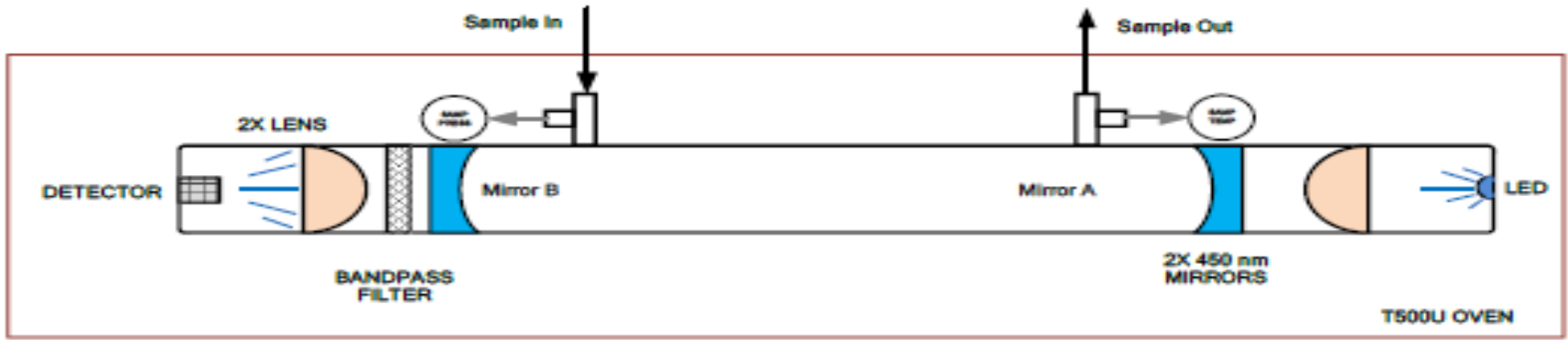
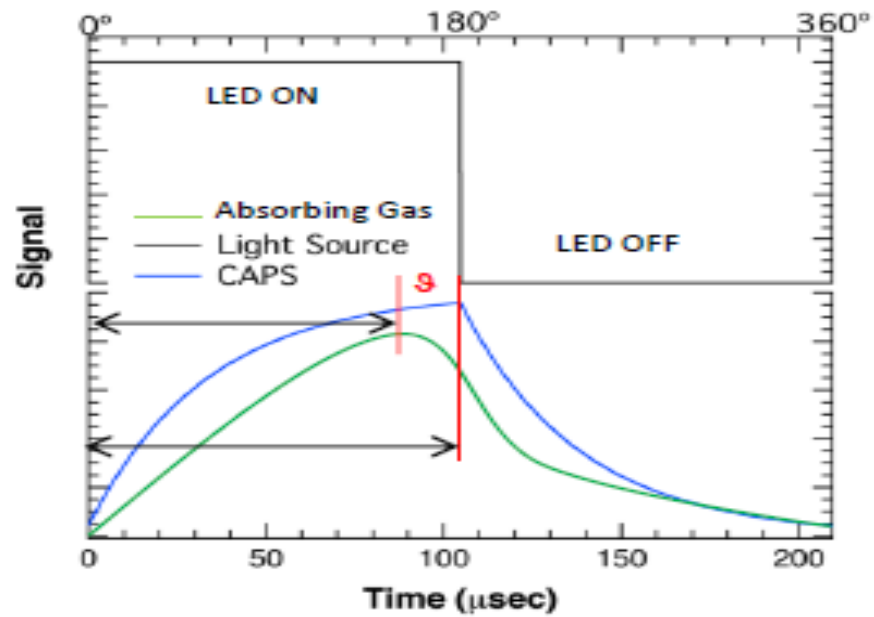
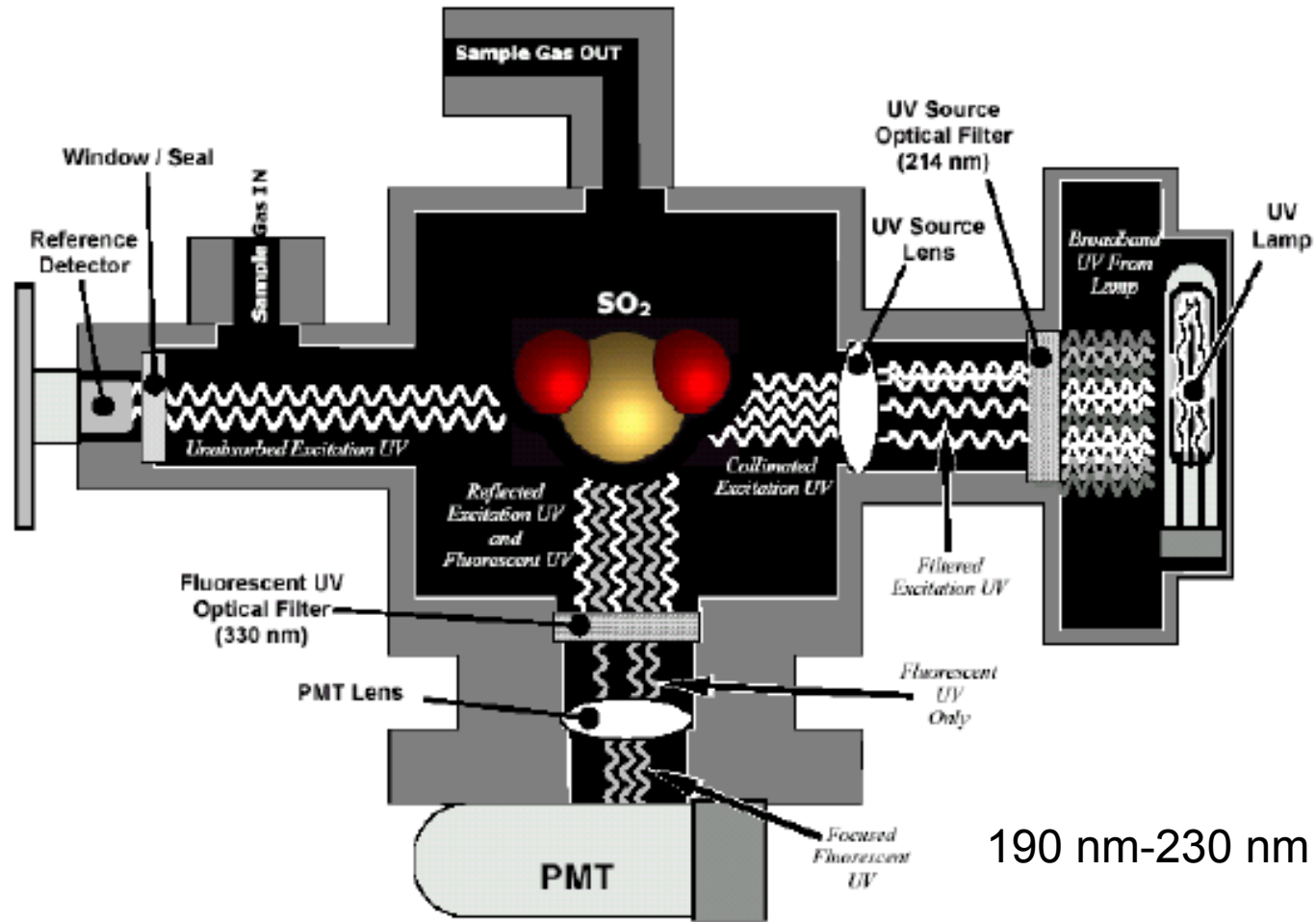


Figure 14-1: T500U Optical Absorption Cell 450 nm



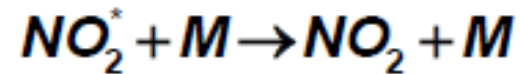
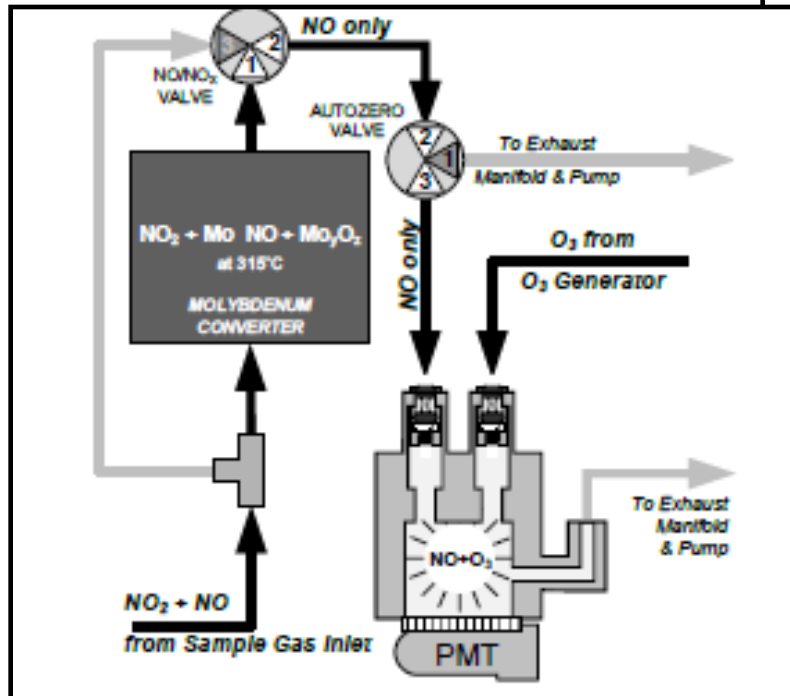
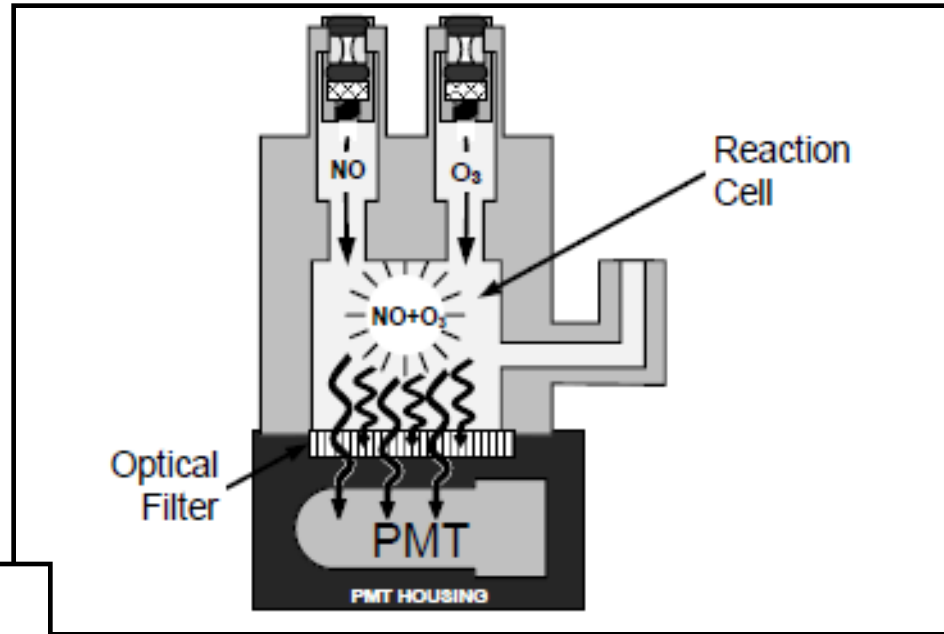
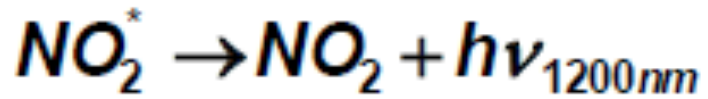
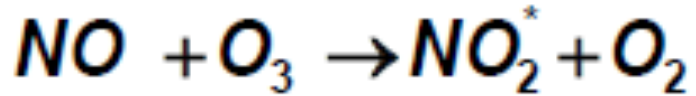
UV Fluorescence SO₂ Analyzer



190 nm-230 nm

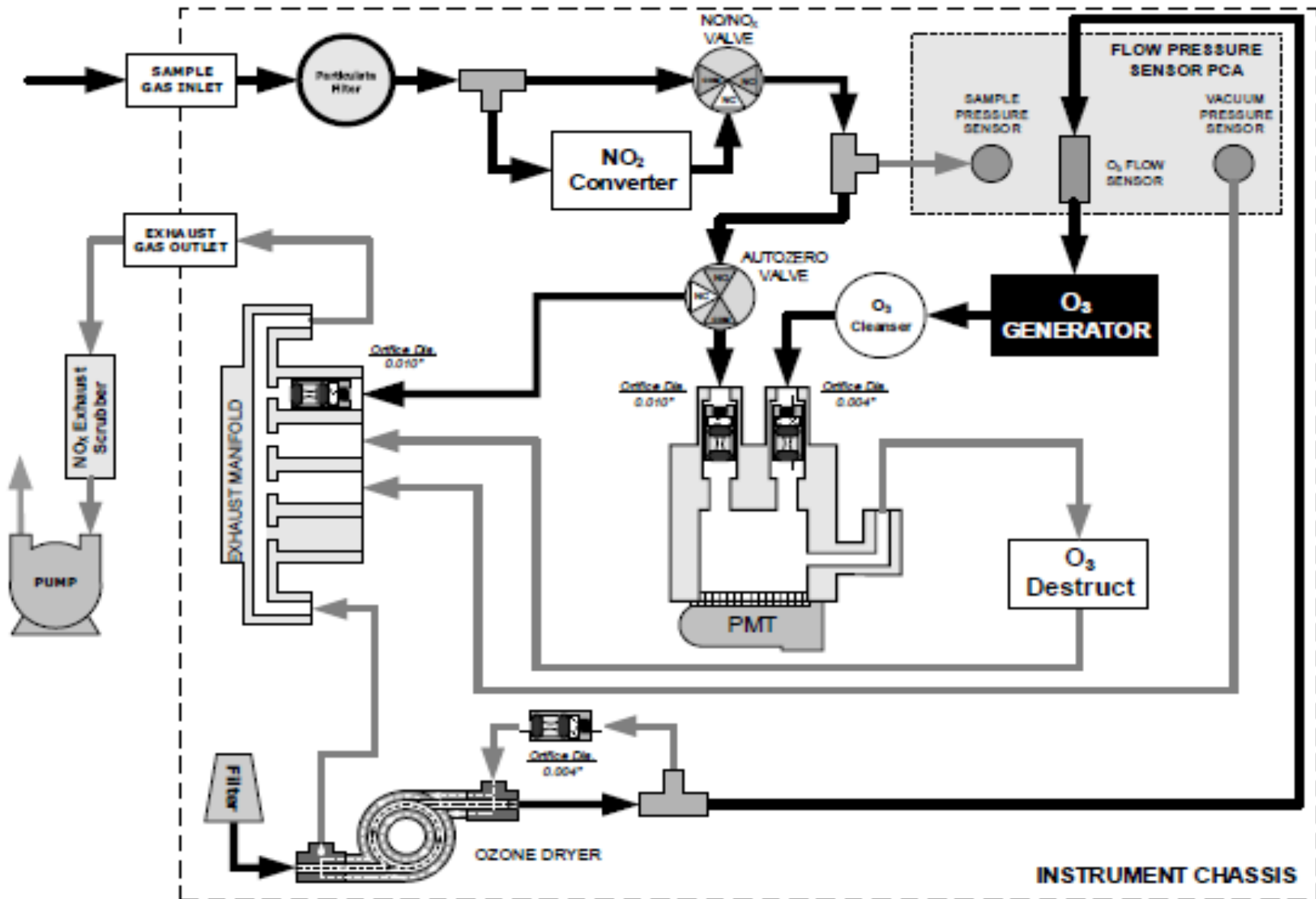
214 excitation 330 emission

CHEMILUMINESCENCE NO_x Analyzer



Less pressure (avoid collision)
increase ozone concentration

CHEMILUMINESCENCE NO_x Analyzer



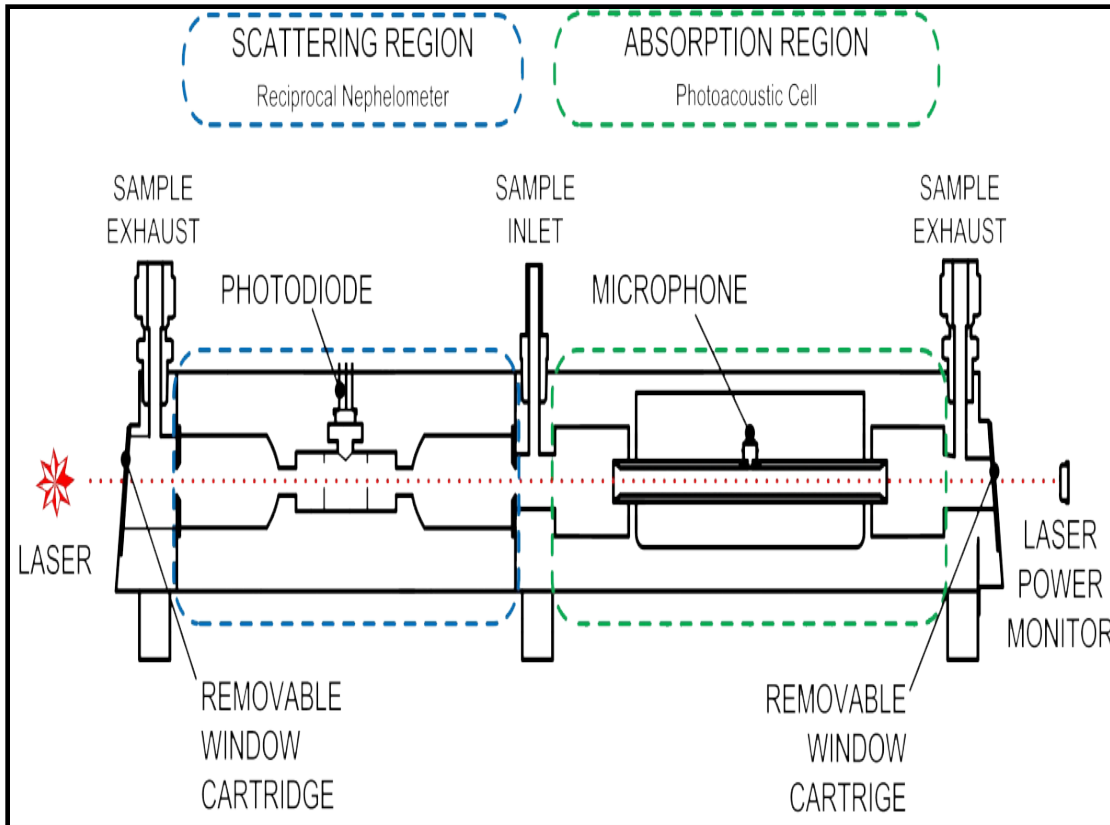
The Photoacoustic Extinctionmeter (PAX)

simultaneously measure light scattering and absorption

Nephelometer to measure the light scattering.

The absorption through photoacoustic

Absorbing particles heat up and quickly transfer heat to the surrounding air. The periodic heating produces pressure waves that can be detected with a sensitive microphone.



Red (870 nm) – absorption is highly specific for black carbon (soot) particles; scattering best for large particles

Green (532 nm) – measures in the visual range, typically what the human eye observes

Blue (405 nm) – absorption correlates to the organic, or brown carbon content; efficient scattering for fine and ultrafine particles

Microwave Radiometer



Tropospheric Profiling
of temperature, humidity and
liquid water

Water Vapour Monitoring
e.g. at astronomical sites

Now-Casting

- atmospheric stability
- severe weather

Data Assimilation
input for weather and
climate models

Boundary Layer Profiling

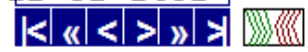
- high-resolution temperature profiles (better than balloons)
- 24/7 monitoring of temperature inversions
- fog detection
- air pollution applications

Temperature, Humidity and Liquid Profiler

21 K-band (22 to 30 GHz) channels

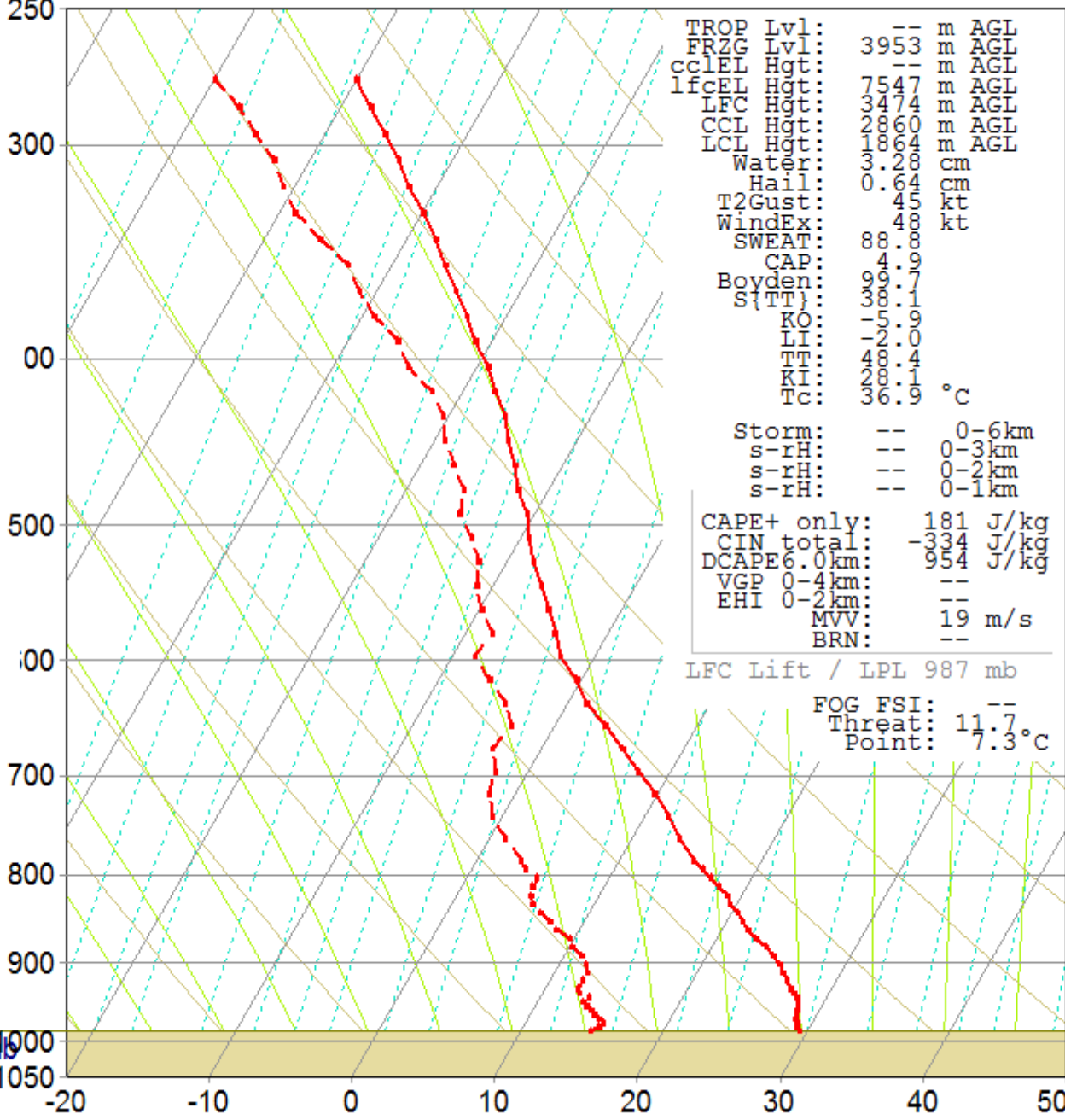
14 V-band (51 to 59 GHz) channels

that measures energy emitted at
sub-millimetre-to-centimetre
wavelengths



Parcel Data	
Pres	390.7 mb
Hgt	7.691 m
(MSL)	25.232 ft
±Std	+339 m
Temp	-21.1 °C
Td	-26.5 °C
T-Td	5.4 °C
RH	62%
PT	329.7 °K
ePT	333.8 °K
L.R.	7.6 °C/km
Tmax	55.0 °C
Wind:	
Hgt:	
(MSL)	

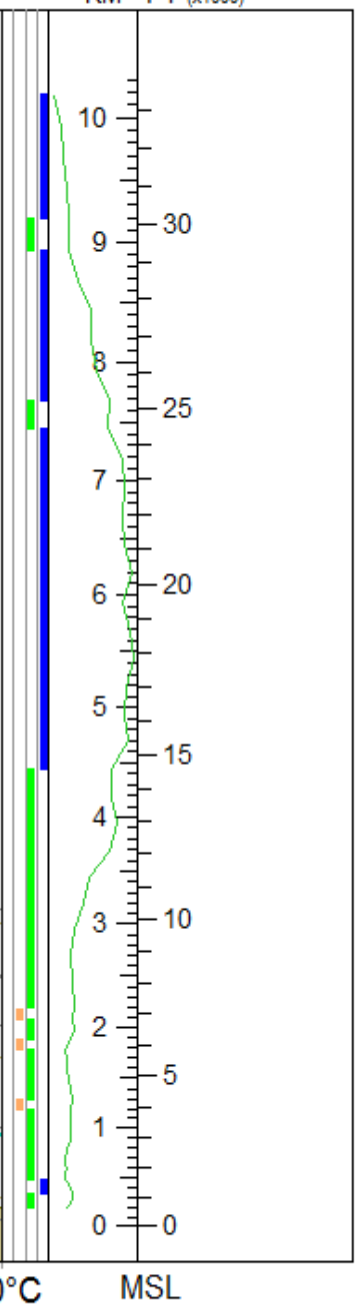
Diagram Data	
Pres	395 mb
Hgt	7.617 m
(MSL)	24.989 ft
Hgt	7.426 m
(AGL)	24.362 ft
Temp	-8.7 °C
	16.3 °F
DrvA	71.7 °C
WetA	26.1 °C
MixR	5.0 g/kg



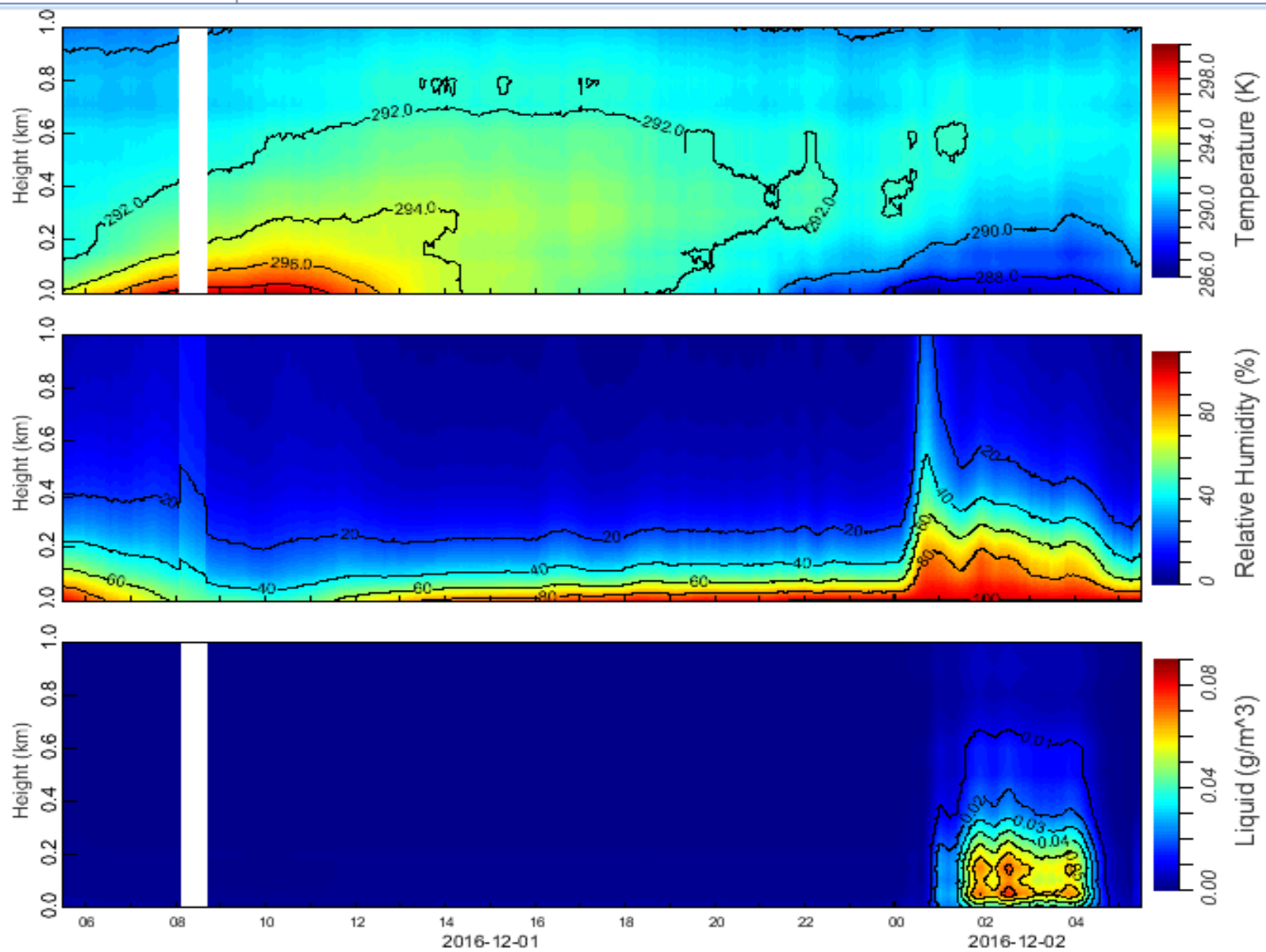
TROP Lvl:	--	m	AGL
FRZG Lvl:	395	m	AGL
CCL Lvl:	13	m	AGL
CCL Hgt:	75	m	AGL
lfc Lvl:	47	m	AGL
LFC Hgt:	34	m	AGL
CCL Hgt:	74	m	AGL
LCL Hgt:	20	m	AGL
LCL Hgt:	60	m	AGL
LCL Hgt:	40	m	AGL
Water:	0.3	cm	
Hail:	0.2	cm	
T2 Gust:	4.6	kt	
Wind Ex:	8.4	kt	
SWEAT:	88	kt	
CAP:	4.4	kt	
Boydex:	0.0	kt	
S(TT):	0.0	kt	
KO:	0.0	kt	
LI:	0.0	kt	
TT:	0.0	kt	
KI:	0.0	kt	
Tc:	0.0	kt	
Storm:	--	0-6km	
s-rH:	--	0-3km	
s-rH:	--	0-2km	
s-rH:	--	0-1km	
CAPE+ only:	181	J/kg	
CIN total:	-334	J/kg	
DCAPE 6.0km:	954	J/kg	
VGP 0-4km:	--		
EHI 0-2km:	--		
MVV:	19	m/s	
BRN:	--		

LFC Lift / LPL 987 mb

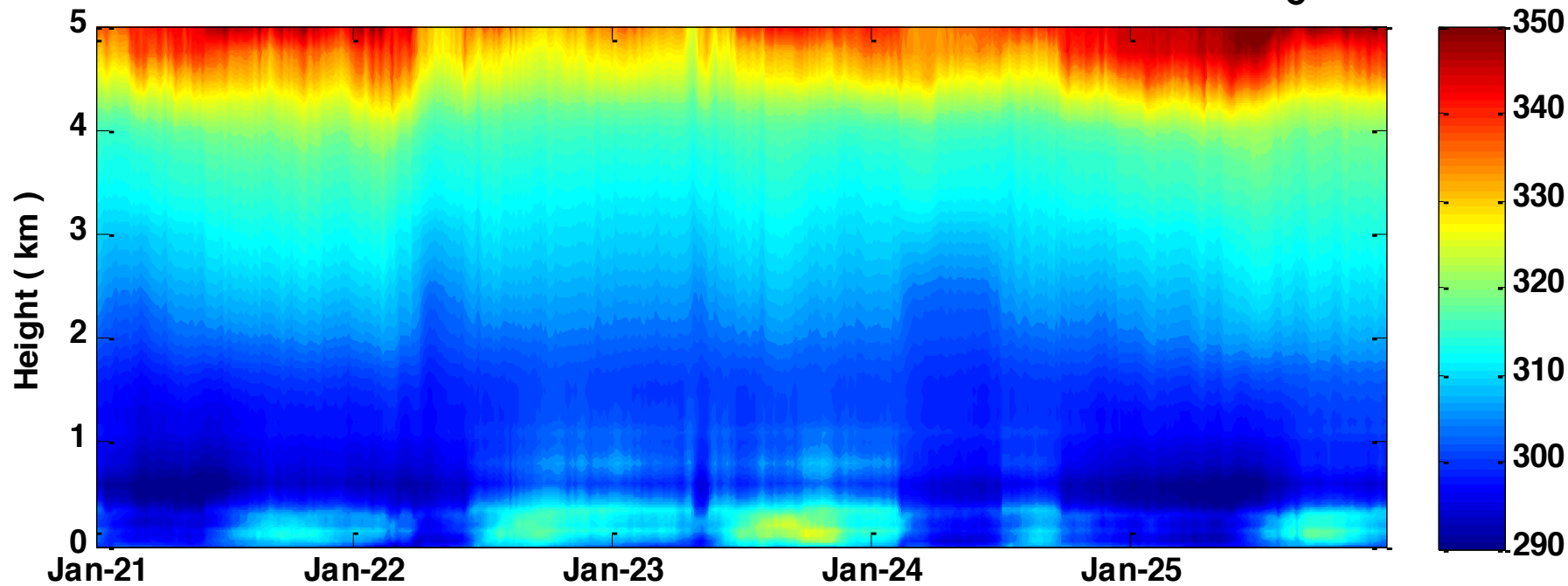
FOG FSI: --
Threat: 11.7
Point: 7.3 °C



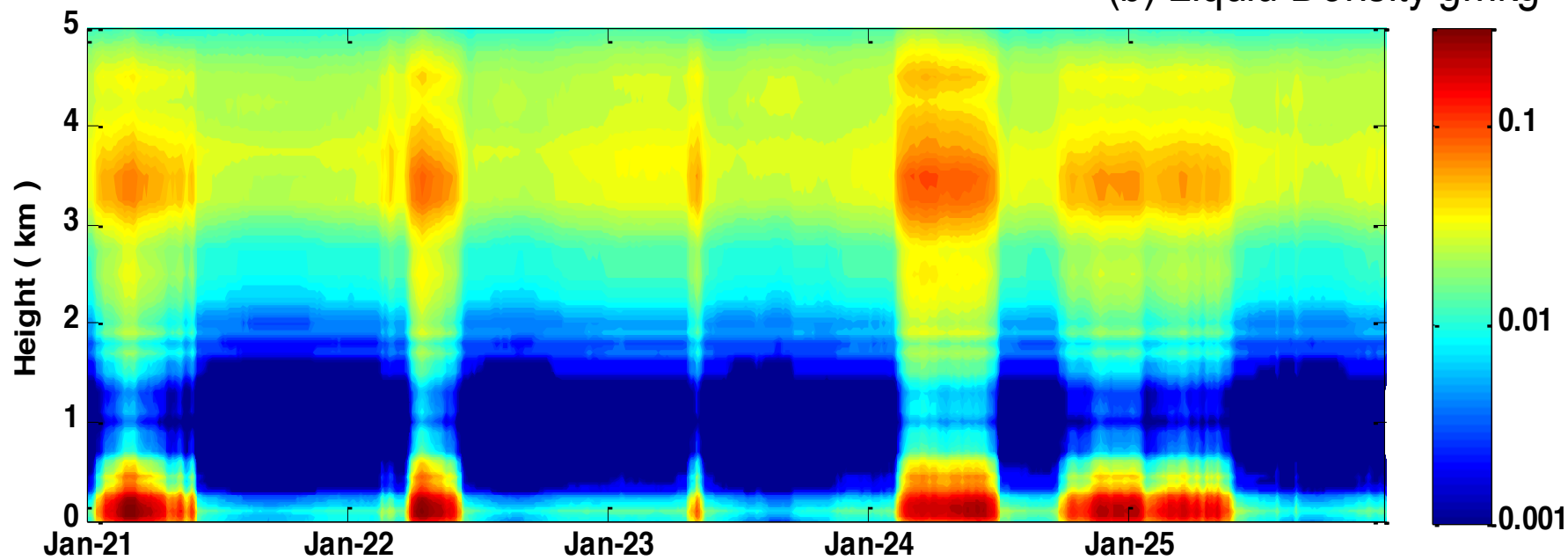
Stn Elev: 191 m
QNH = 1009.3 mb
DA: 846 m, ISA 1050



(a) θ_e K

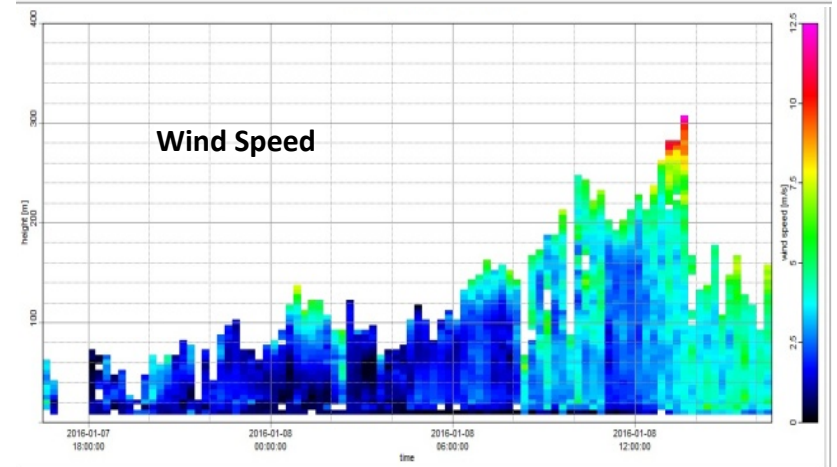
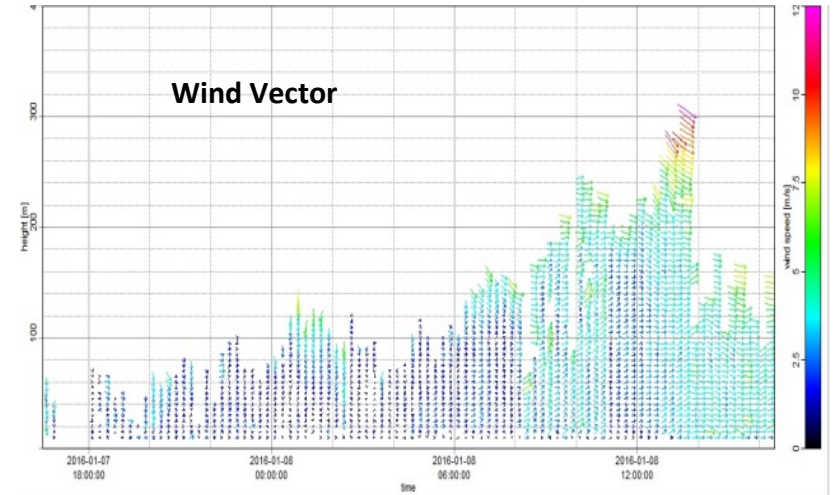


(b) Liquid Density gmkg^{-1}



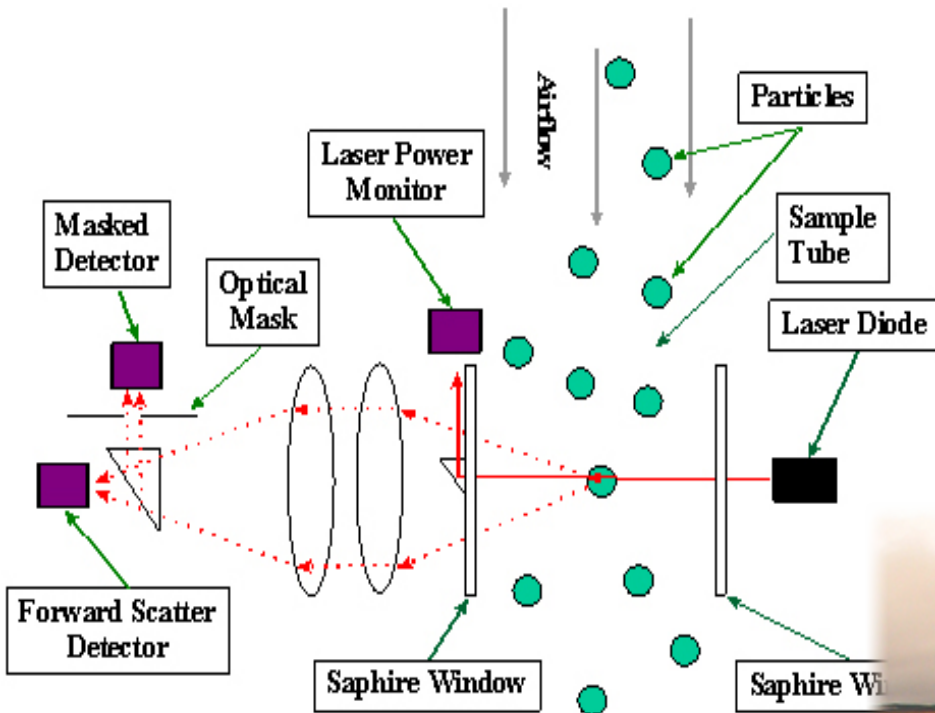
SODAR

Used as a wind profiler to measure the scattering of sound waves by atmospheric turbulence



Fog Monitor (FM-120)

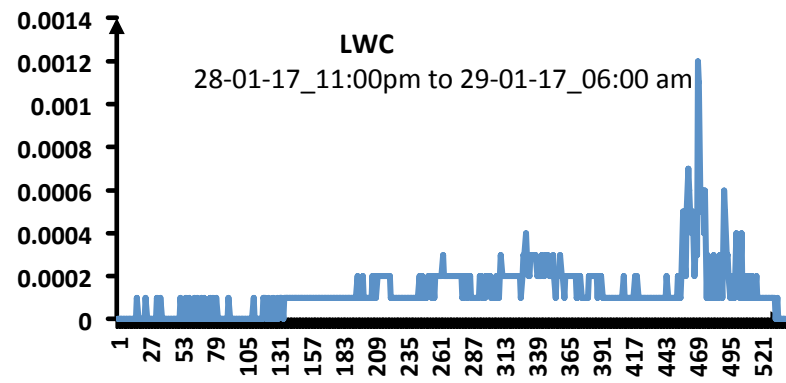
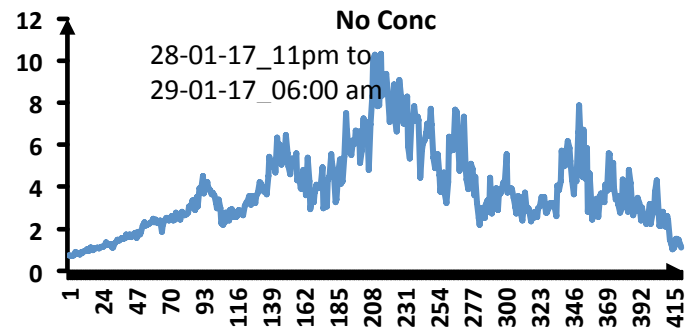
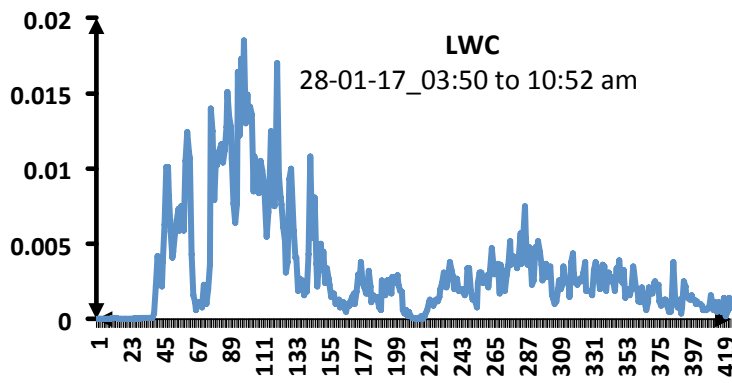
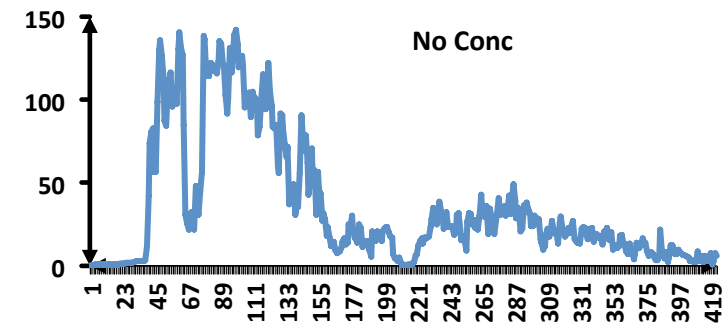
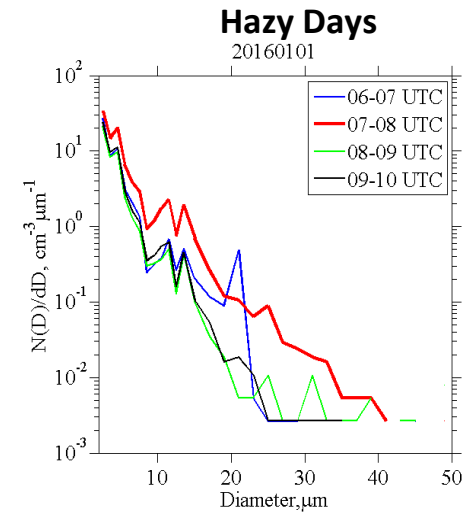
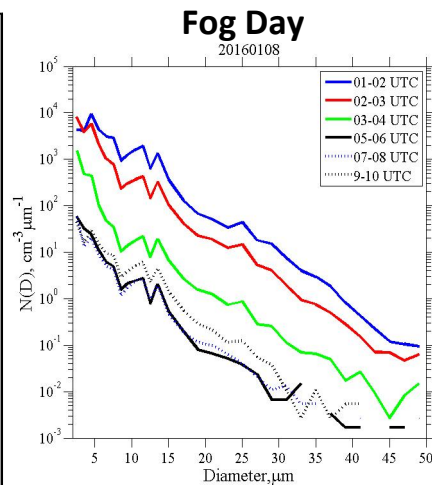
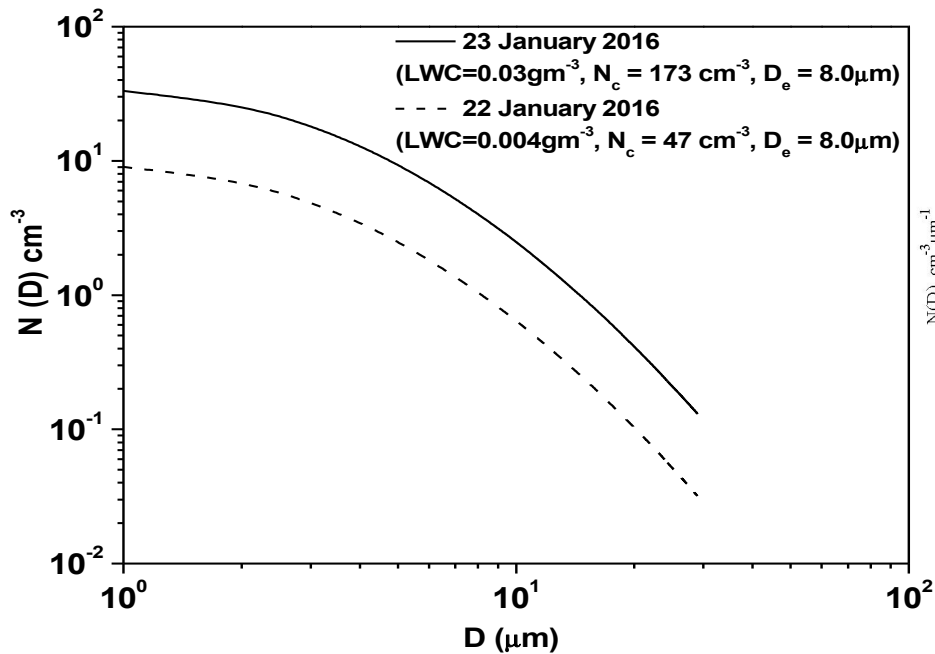
forward-scattering optical spectrometer



Droplets with 2-50 μm diameter

- Particle diameter
- Particle number concentration
- Liquid water content (LWC)
- Effective diameter (ED)
- Median volume diameter (MVD)





Emission

Outline

1. What is an Emissions Inventory and why do we need one?

1. Building and bottom up emission inventory

2. Satellite measurements and top down emission inventory

3. Data base for Global and regional Emission Inventory data sources

What is an Emission Inventory?

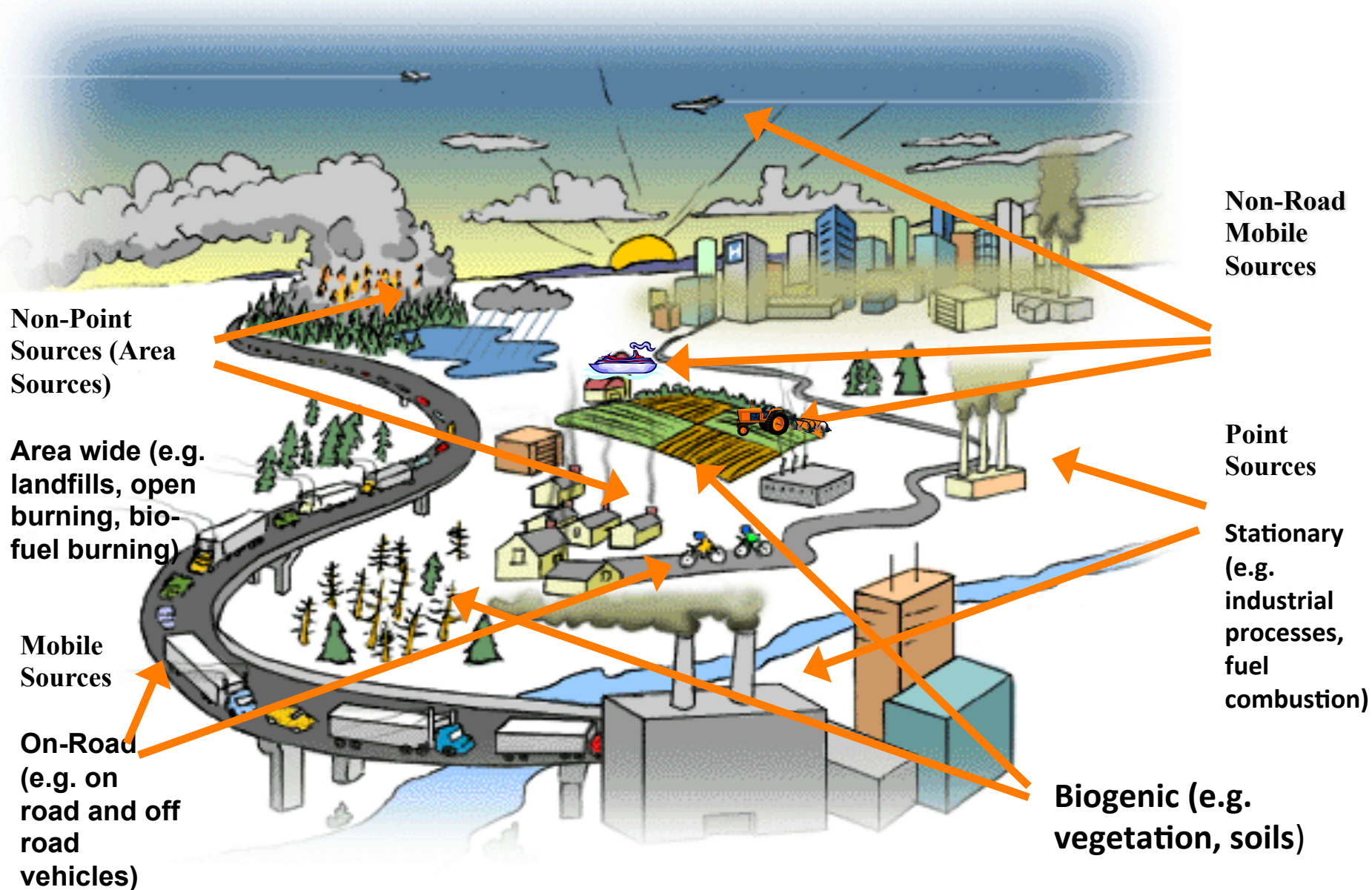
An emission inventory is an accounting of the amount of pollutants discharged into the atmosphere. An emission inventory usually contains the total emissions for one or more specific greenhouse gases or air pollutants, originating from all source categories in a

- Specific geographic area
- Certain period of time
- Organized by type of data (e.g., point, area, mobile, biogenic)



What are Emission Sources?

(Anthropogenic and Natural)



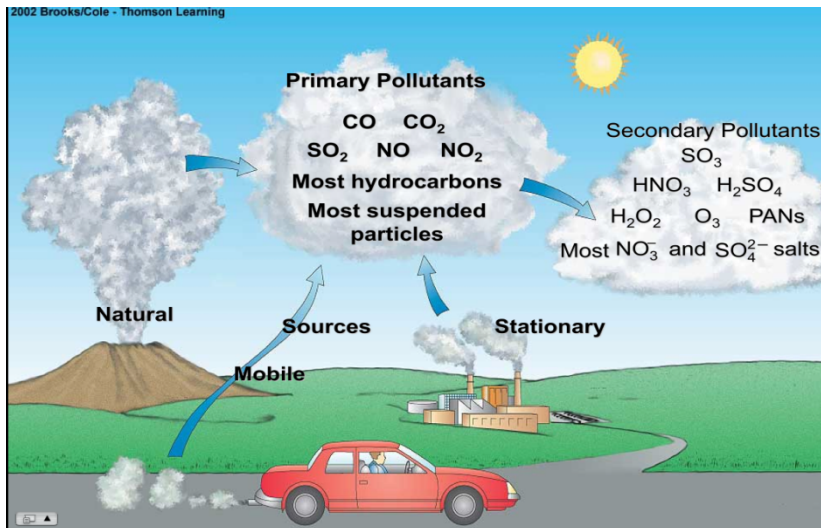
What information we get from EI?

- **Where air pollution is emitted**
(**at surface:** e.g. from mobile and st. sources, ocean, biomass,
non- surface: e.g. Air craft, lightening, etc, Indoor activity or outdoor activity)
- **How much is emitted from each source**
- **What sources would be most effective to control**



Information we could not get from EI?

- The distance that air pollutant emissions are transported
- The amount of air pollution to which people are exposed
- The health risk from the air pollution



Formation of secondary pollutants (e.g. ozone, PAN, HNO₃).

Windblown dust

Why is emission inventory needed?

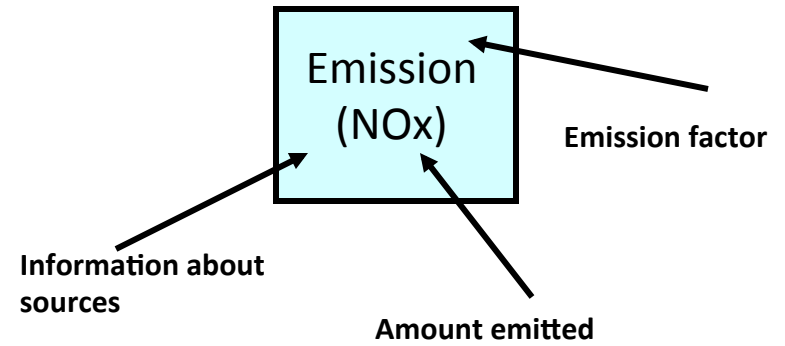
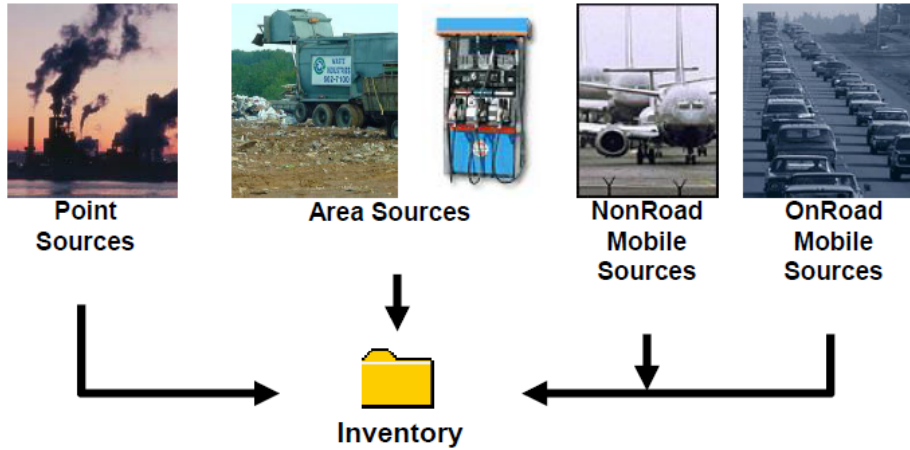
Air quality management (Identify sources of pollution and pollutants of concern)



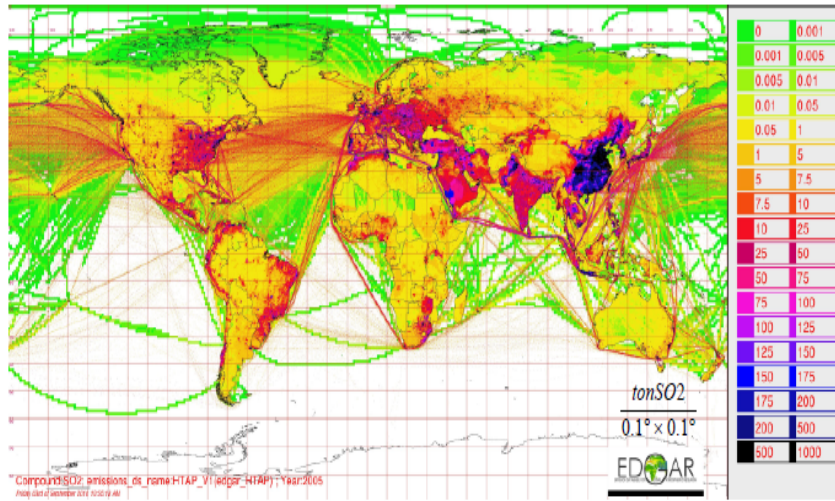
Types of Inventories

- **Bottom up EI** (emissions → concentrations)
- **Top Down** (concentrations → emissions)
 1. **Direct estimate**
 2. **Combination of satellite and bottom up**
- **Annual average**
- **Seasonal inventories**
- **Forecasted - future estimates**
- **Gridded / Modeling**

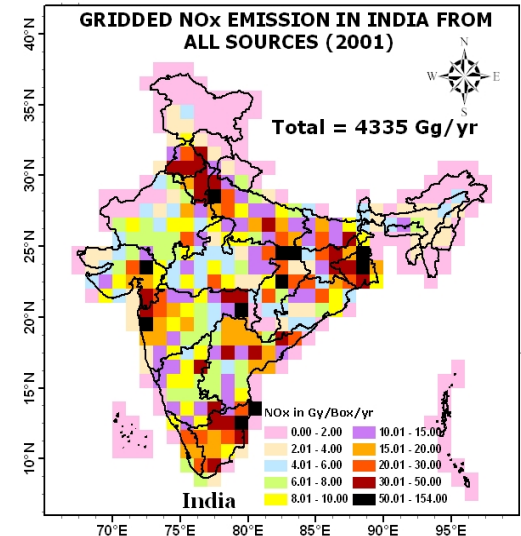
Building an bottom up Emission Inventory



Edgar (global)



India (regional)



Activity Data and Methodology

- **Methodology**

Emission Inventories are estimated as per the most widely used emission factor approach.

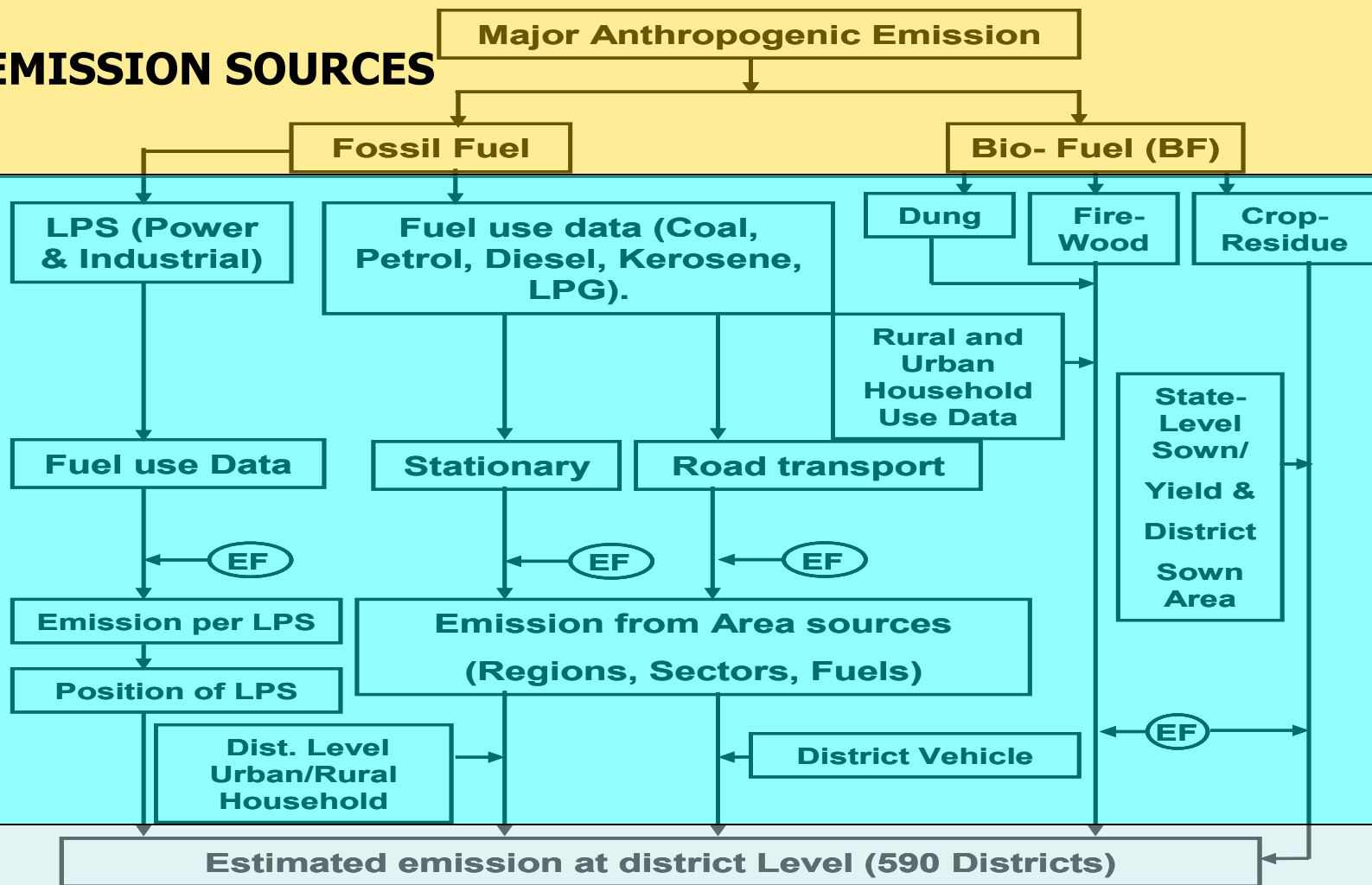
$$Em_{j,k} = \sum_l \sum_m FC_{k,l,m} \left[\sum_n EF_{j,k,m,n} X_{k,l,m,n} \right]$$

- j,k,l,m,n = Species, country, sector, fuel type, Technology
- Em = Total Emission
- FC = Fuel consumption , Kg/yr
- $EF_{j,k,m,n}$ = Emission factor specific to fuel/Technology
- X = Fraction of fuel for this sector consumed by a specific technology, where $\sum_n X_{k,l,m,n} = 1$ for each fuel and sector

Schematic Methodology for the Development of Indian emission estimation

MAJOR EMISSION SOURCES

EMISSION CALCULATION

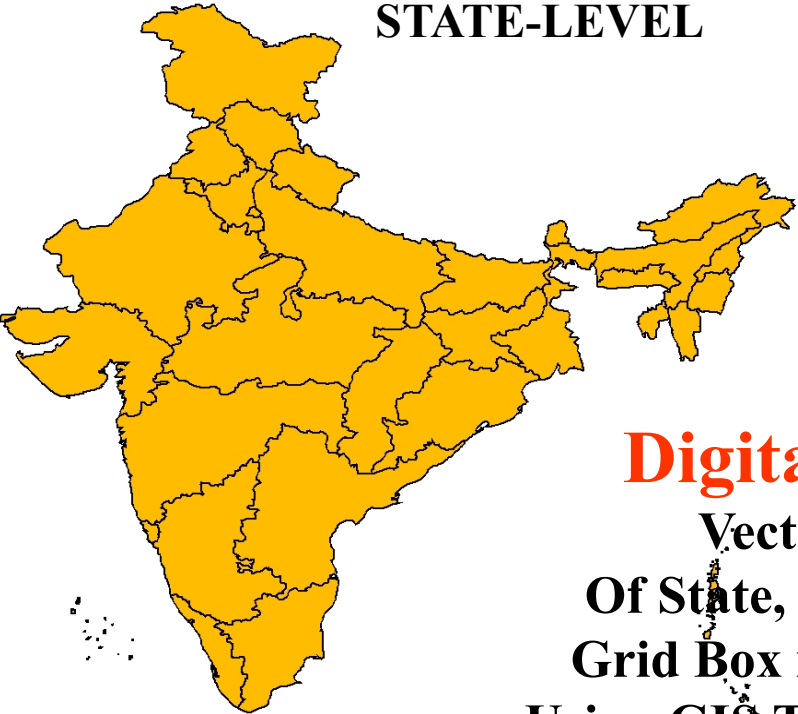


GIS APPLICATION

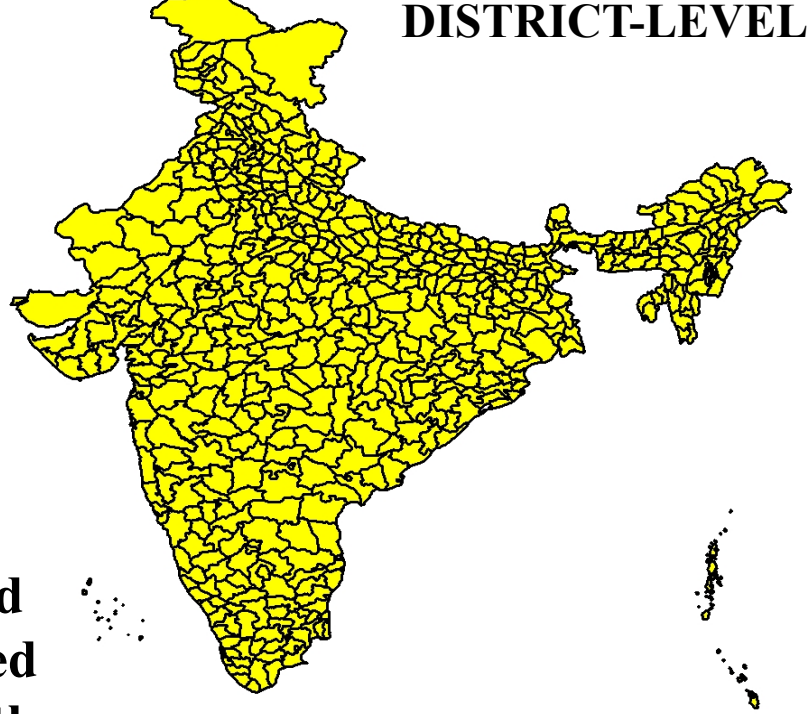
FINAL EMISSION

Gridded Emission values at 1° x 1° Resolution

STATE-LEVEL

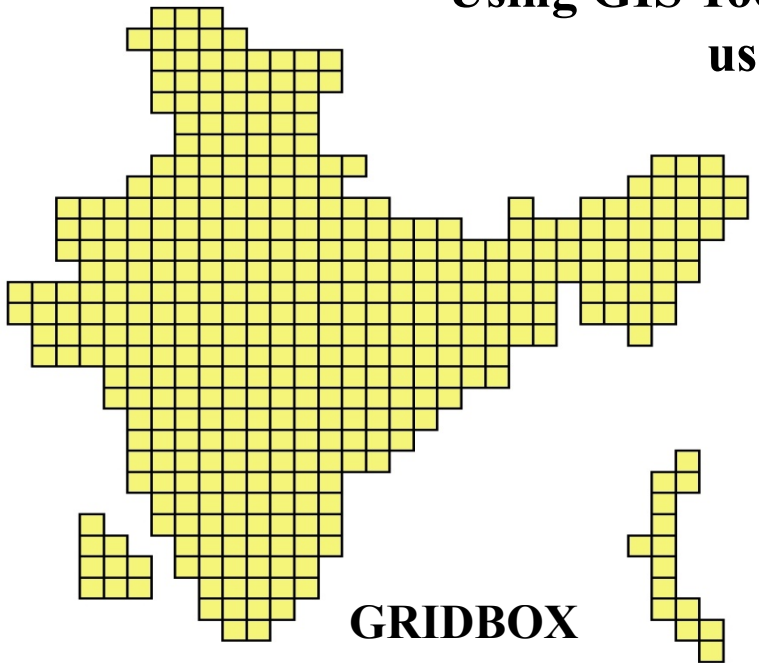


DISTRICT-LEVEL

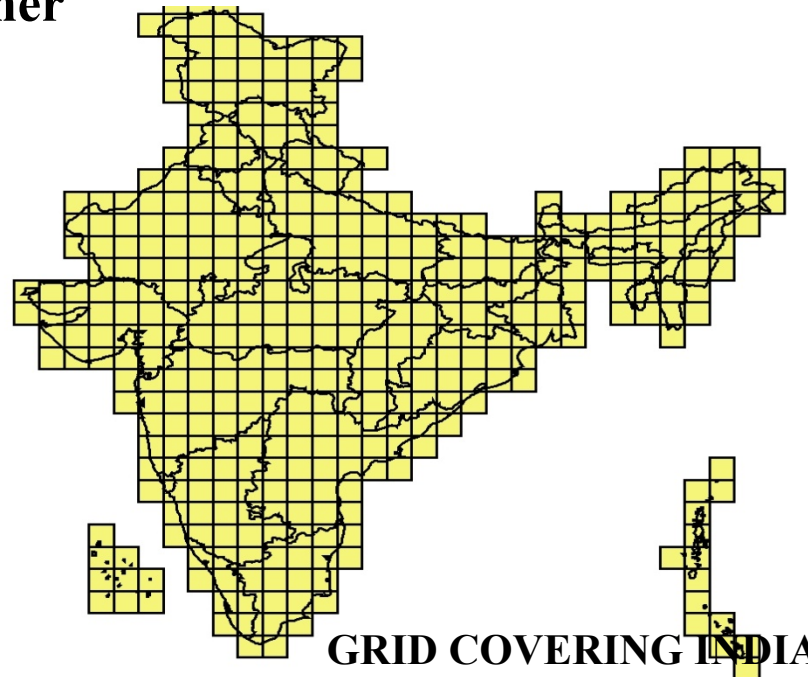


Digital Data:
Vector-Map
Of State, District and
Grid Box is Generated
Using GIS Tool for further
use.

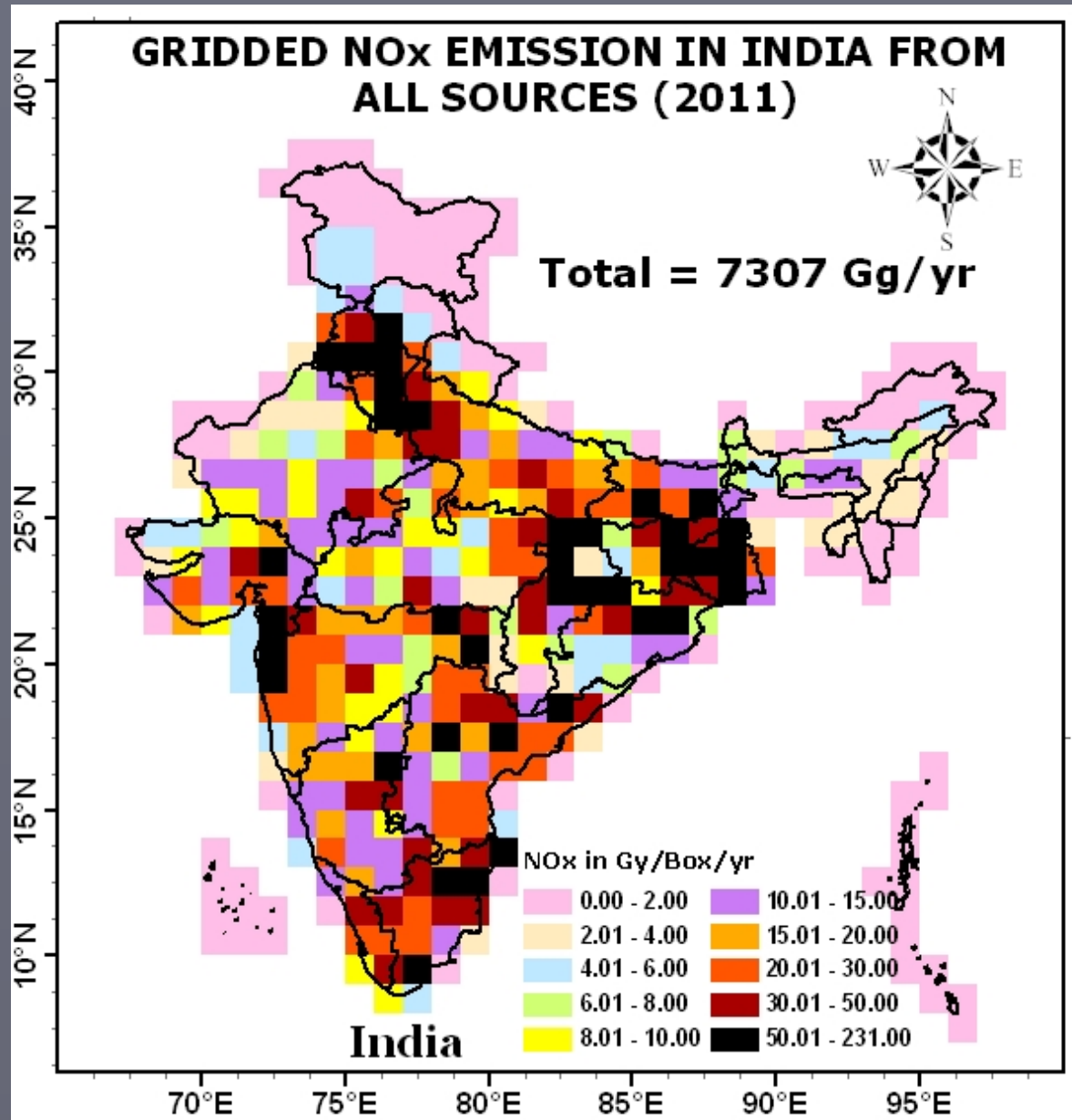
GRIDBOX



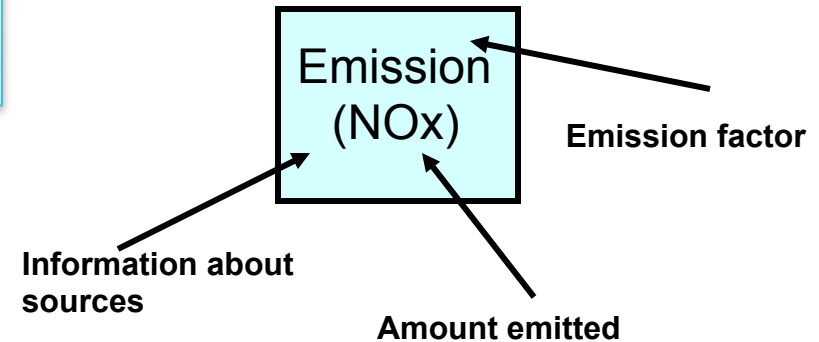
GRID COVERING INDIA



NO_x Emission in India (2011)



Drawbacks of bottom-up inventories



- Depend on the availability and reliability of the statistical information.
- Depend on historic information: easily out-dated.
- Uncertainties in spatial resolution if only area totals are available.

**If you don't like your result
change your approach!**

Constraining emissions with satellite observations

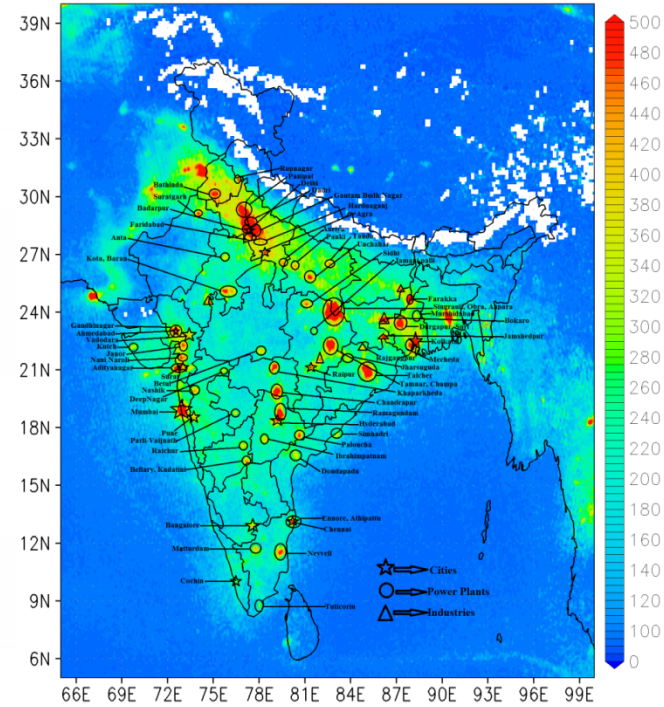
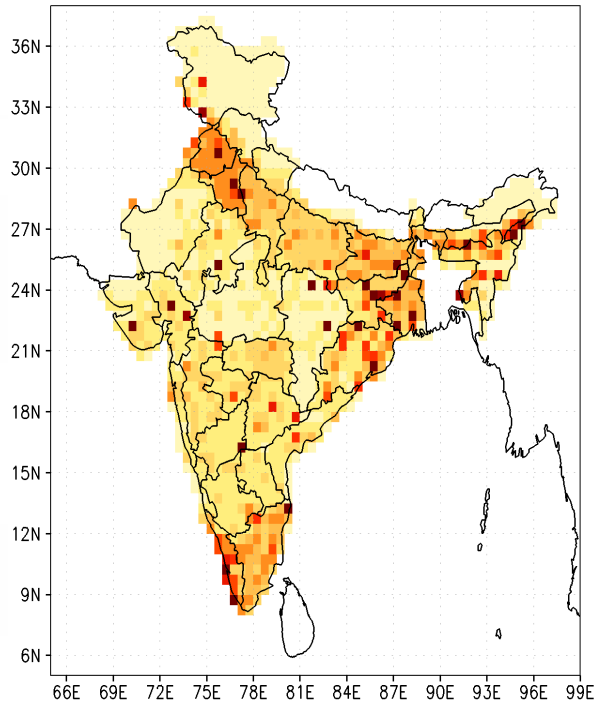
- Satellites have world-wide, homogeneous coverage.
- Correcting inventory for emission trends
- Detecting new (unknown) emission sources
- Emission trend analysis reveals effectiveness of air pollution policy
- Up-to-date emission inventories improve air quality forecasting

Top Down Approach

Emissions



Concentrations



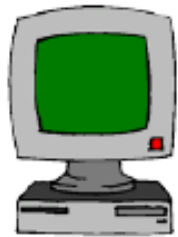
• **Top Down** (concentrations → emissions)

1. Direct estimate

2. Combination of satellite and bottom up

Top-down Approach

Martin et al. [2003, 2006]
Lamsal et al. [2011]



NO2_model



NO2_satellite

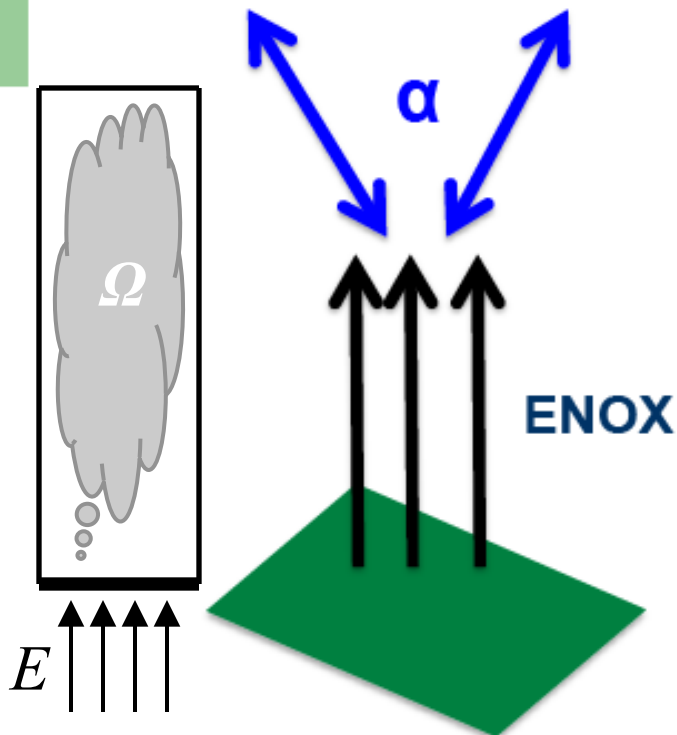
= Adjustment of the emissions with satellite observations to reduce the disagreement between model and observation.

Assuming that horizontal transport of NO_x is negligible, a posteriori emissions can be derived as following:

$$\alpha = \text{ENOX_apriori} / \text{NO2_model}$$

$$\Rightarrow \text{ENOX_aposteriori} = \alpha \times \text{NO2_satellite}$$

ENOX = anthropogenic NO_x emissions
NO2_model = Modeled NO₂ Tropospheric Column
NO2_satellite = Satellite NO₂ Tropospheric Column



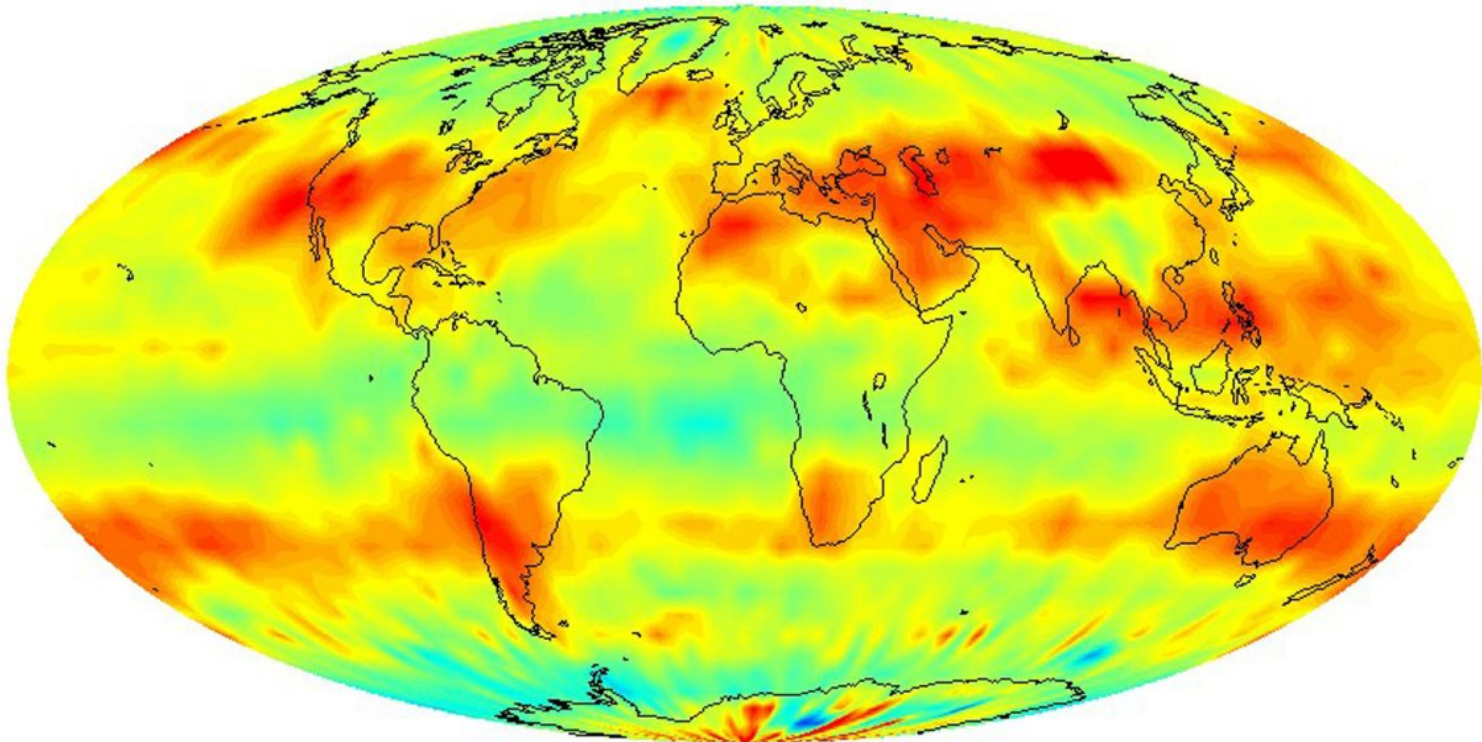
Satellite can only observe pollutant concentrations. These should be backtracked to their underlying emissions:

THIS IS AN INVERSE PROBLEM

Complicating factors:

- Transport of pollutant away from the source
- Lifetime of pollutant
- Variability in lifetime (temperature, chemical composition...)

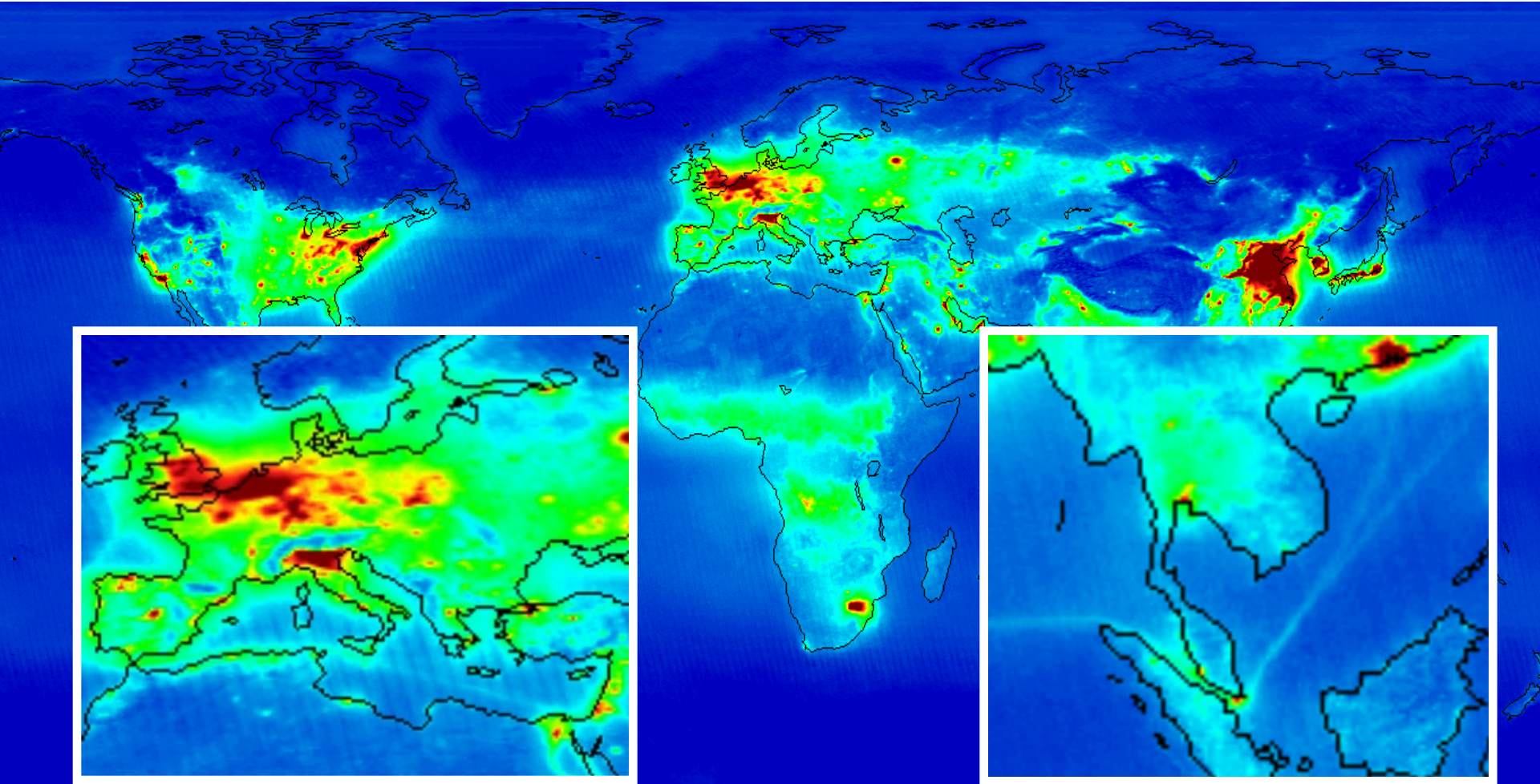
Lifetime example (1): CO₂



AIRS July 2008 CO₂ (ppmv)

Lifetime example (2): NO₂

tropospheric NO₂ in summer: ~4h, in winter: ~10h



OMI 2005-2008

Local, linear relation concentration and emission

Martin et al. (2006) Space-based constraints on NO_x emission, J. Geophys. Res.

Jaeglé et al. (2005) Global partitioning of NO_x sources using (...), Faraday Discuss.

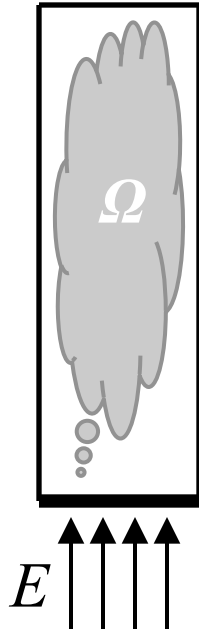
Assume linear relation between NO_x emission and NO₂ concentration:

Direct Approach

$$E_t = \alpha \Omega_{obs} \quad , \quad \alpha = (\Omega_{NO_x} / \Omega_{NO_2}) / \tau_{NO_x}$$

Combine Approach

$$E = \alpha \Omega_{obs} \quad , \quad \alpha = E_{a\text{mod}} / \Omega_{No2m}$$

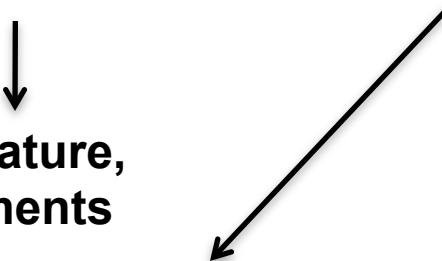


Local, linear relation concentration and emission

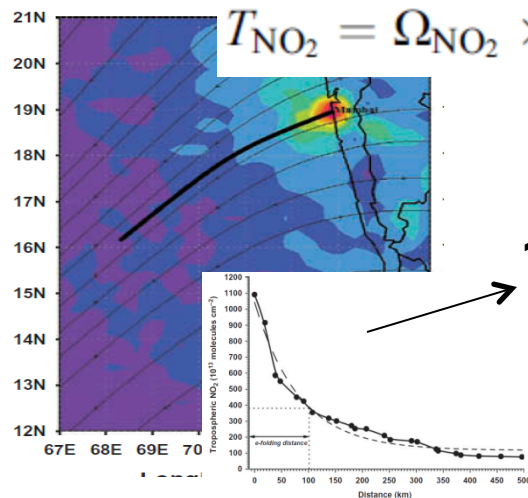
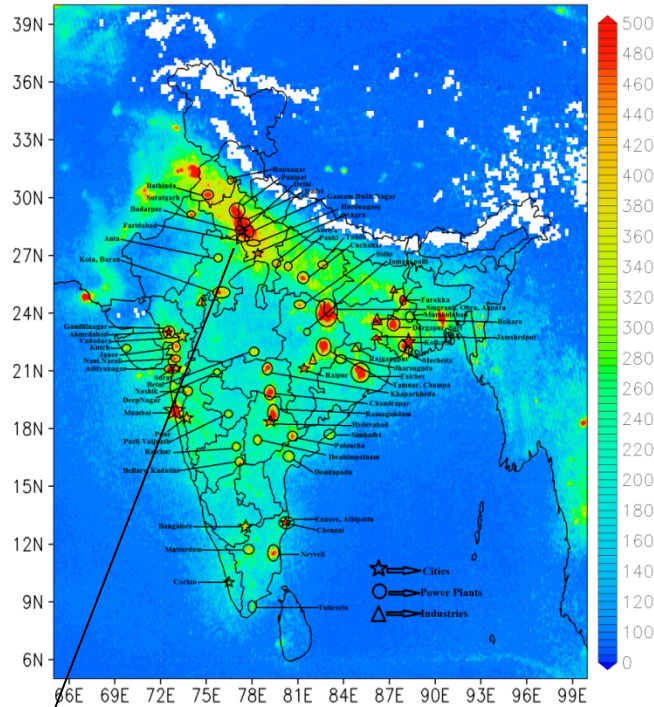
Direct Approach

$$E_t = \alpha \Omega_{obs} \quad , \quad \alpha = (\Omega_{NO_X} / \Omega_{NO_2}) / \tau_{NO_X}$$

From literature, measurements



$$T_{NO_2} = \Omega_{NO_2} \times \exp(-t/\tau_{NO_x}),$$



11.2 hours = 11.2*3600 sec

1.7×10^{15} mole/c m²
(25km x 25 km)

$$E_t = (0.31/11.2) * 1.7 \times 10^{15} \text{ mole/cm}^2$$

Combine Approach (Lab + Satellite)

Satellite based estimates of Fire emission

$$E_i = A(x, t) \times B(x) \times FB \times ef_i$$

Use of fire hot spots

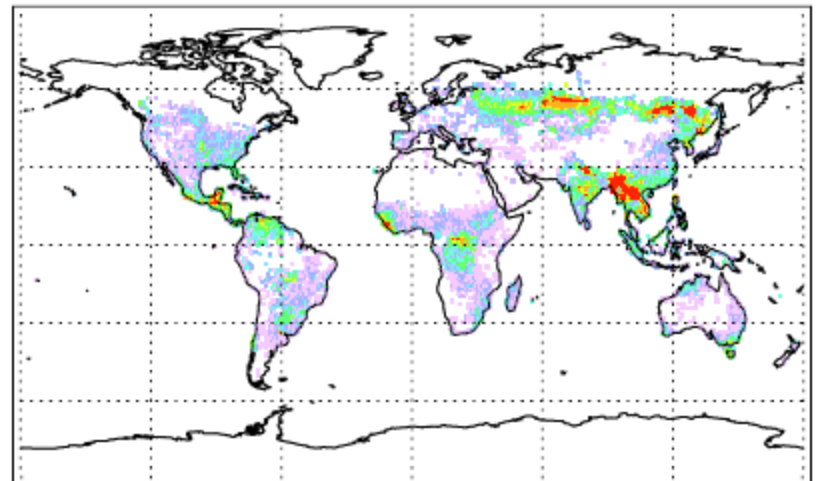
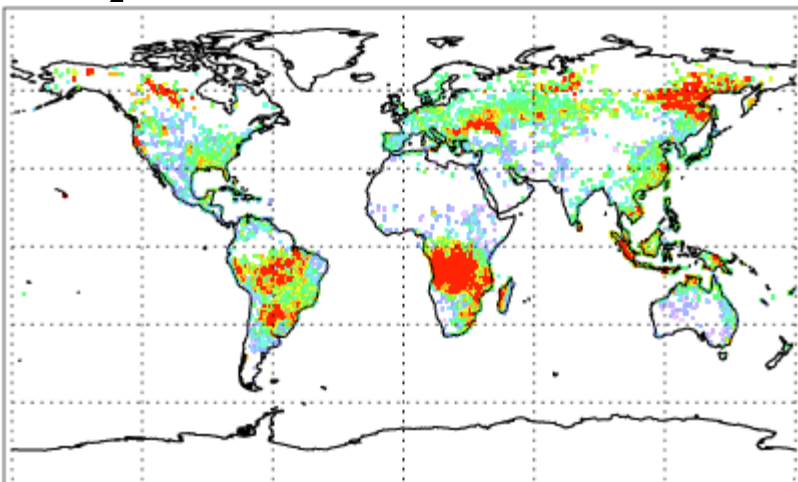
Burnt area at location x and time t : $A(x,t)$

land cover maps (identify vegetation type)

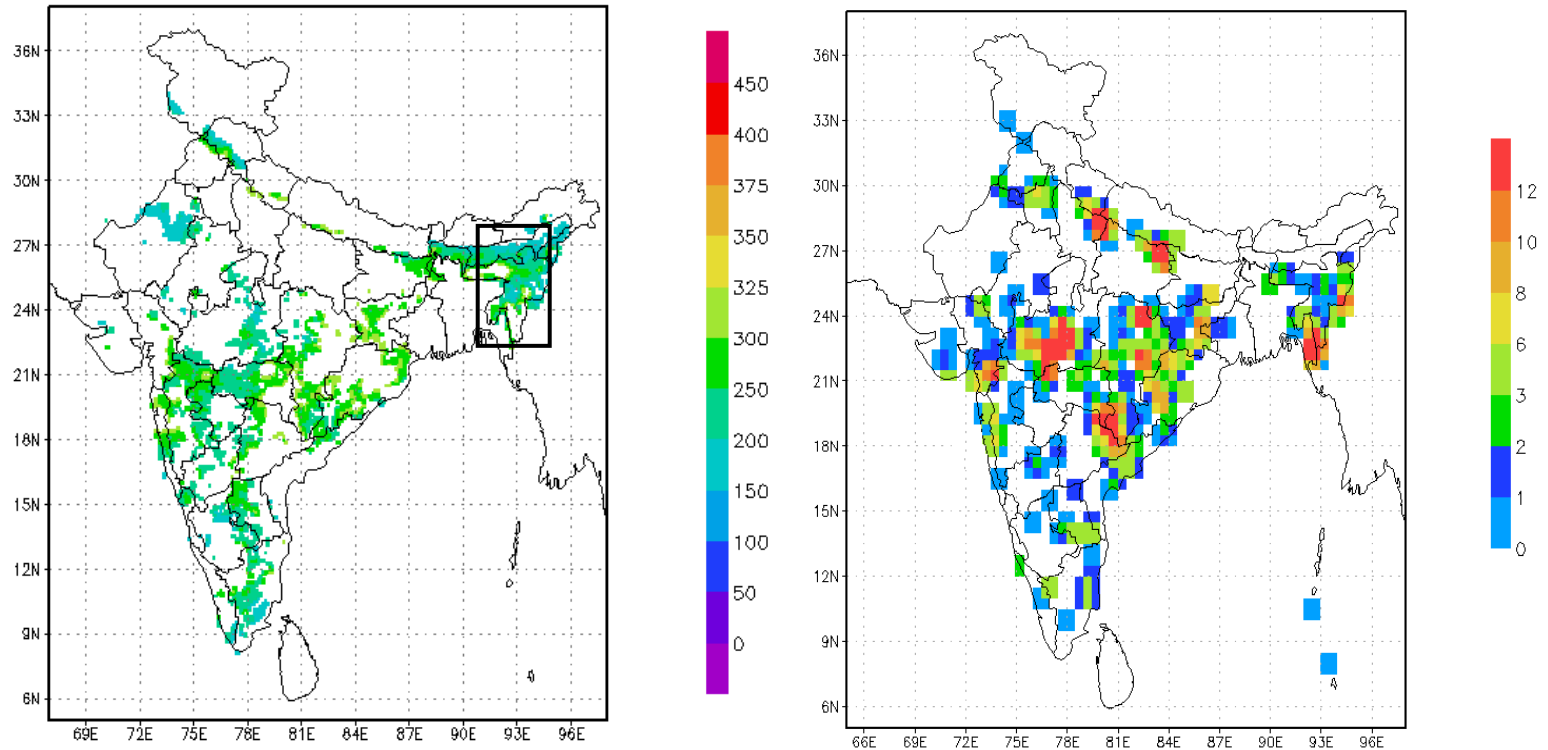
Biomass loading : $B(x)$

consumption estimates at location x : FB

Emission factors (emission factors from filed lab experiment):



Complete Satellite based Approach



(a) Spatial distribution of OMI NO₂ (1×10^{13} molecules cm⁻²) for 2005 for regions with a maximum in the seasonal cycle in tropospheric NO₂ during March-April. In these regions the dominant source type is estimated as biomass burning. (b) ATSR fire counts over the India region during March-April 2005 averaged over $0.5^\circ \times 0.5^\circ$.

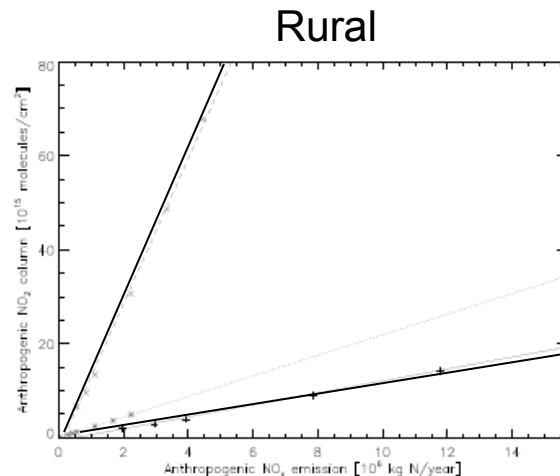
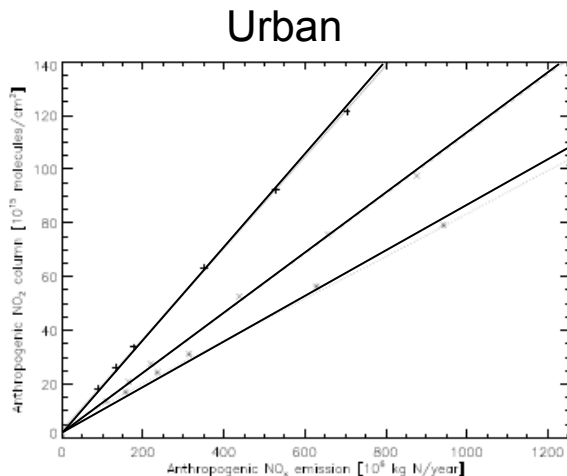
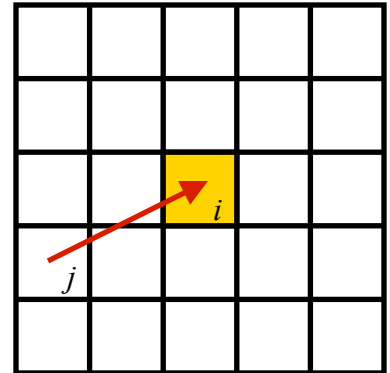
Local, linear relation concentration and emission

Advantages

- Fast, no inverse modeling needed

Disadvantages

- Transport to neighbouring grid cells neglected
- Only one emission update possible
- No new sources detected if *a priori* emission is 0



Local, linear relation applied iteratively

(Ghude et al., (2013), GRL

Assume linear relation between NOx emission and NO2 concentration:

$$E_{i+1} = \alpha \Omega_{obs,i} \quad , \quad \alpha = E_i / \Omega_i$$

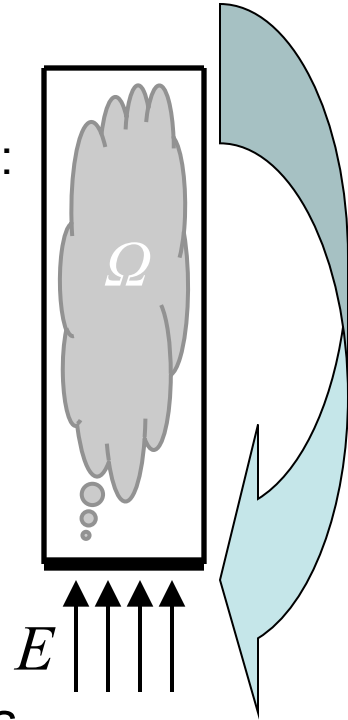
Iterate until convergence criteria are met.

Advantages

- Iteration compensates for transport to neighbouring grid cells
- Accuracy of emissions improves

Disadvantages

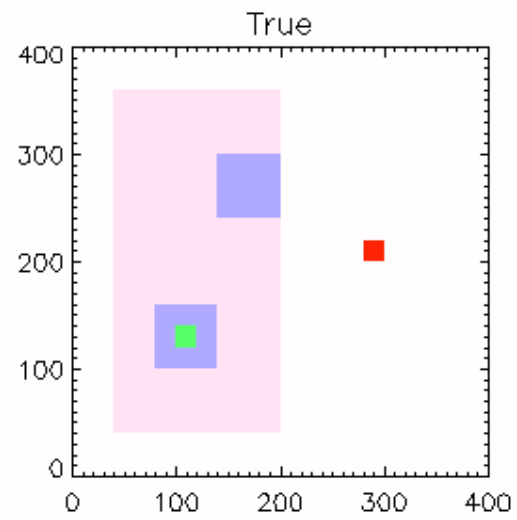
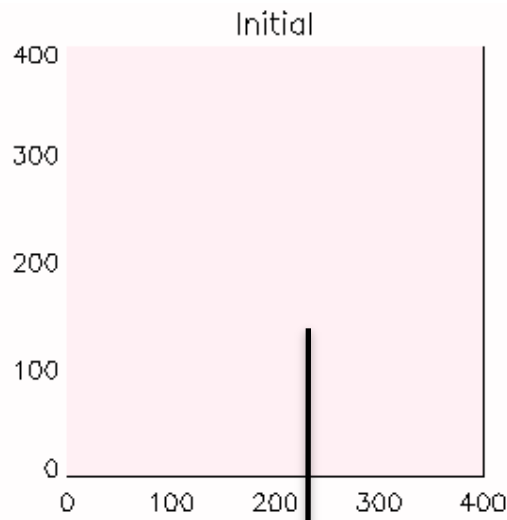
- No new sources detected if *a priori* emission is 0



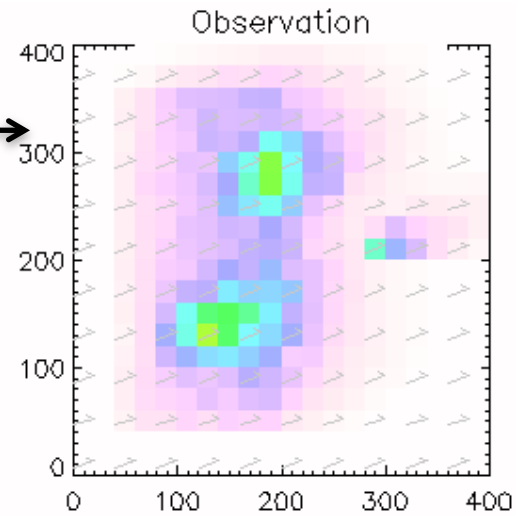
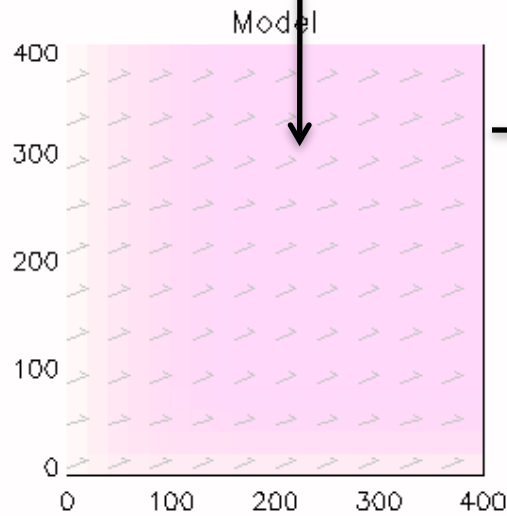
Flatland simulation

- “Toy” transport model in two dimension
- Simplified advection model allows analytic calculation of sensitivities

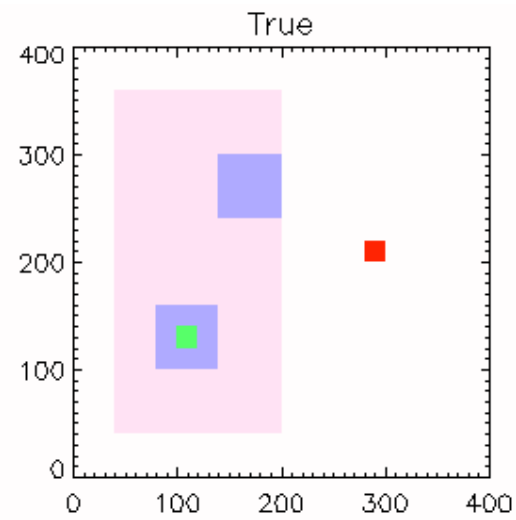
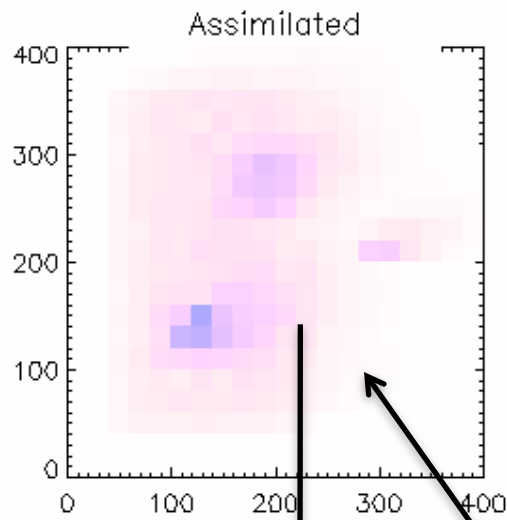
emissions



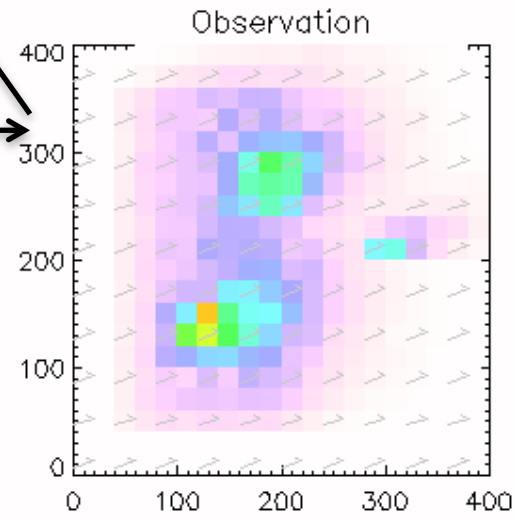
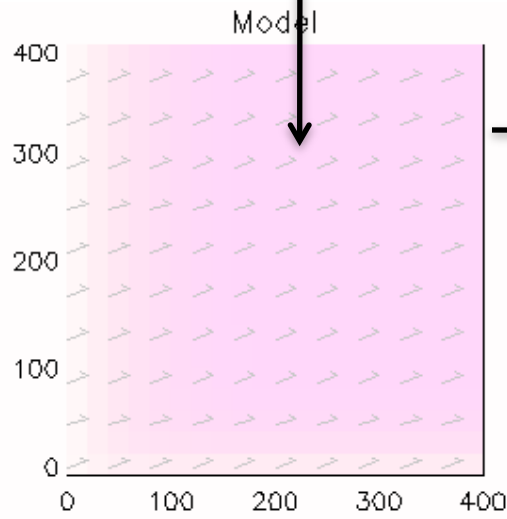
concentrations



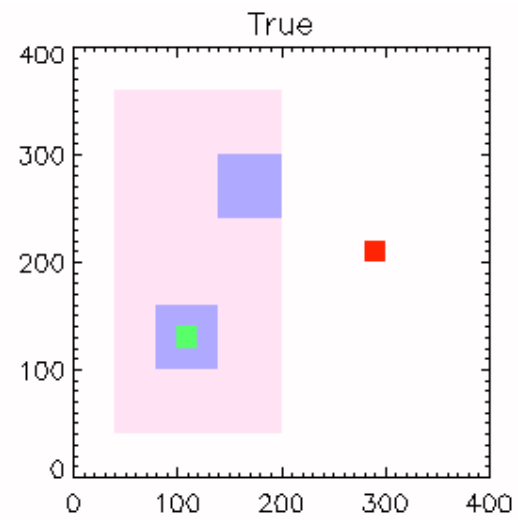
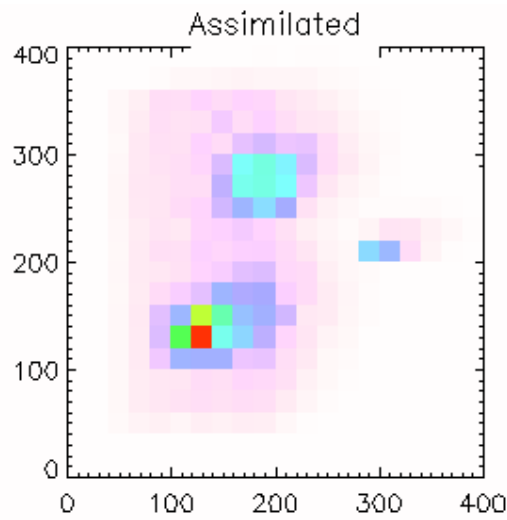
emissions



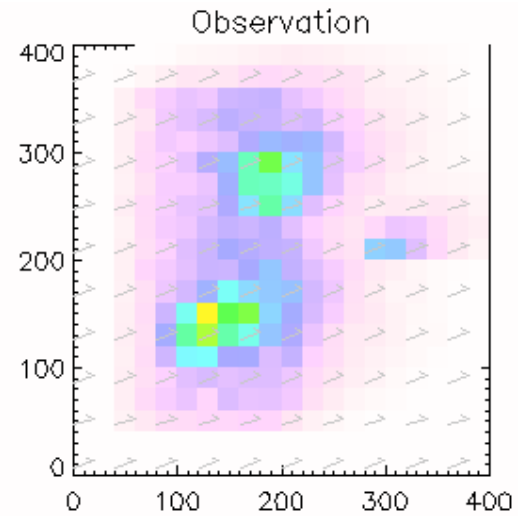
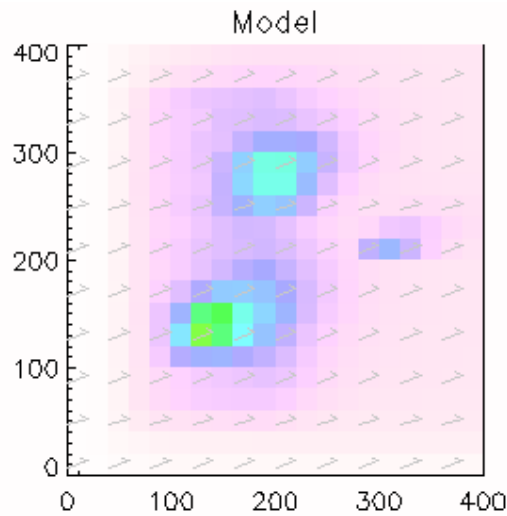
concentrations



emissions



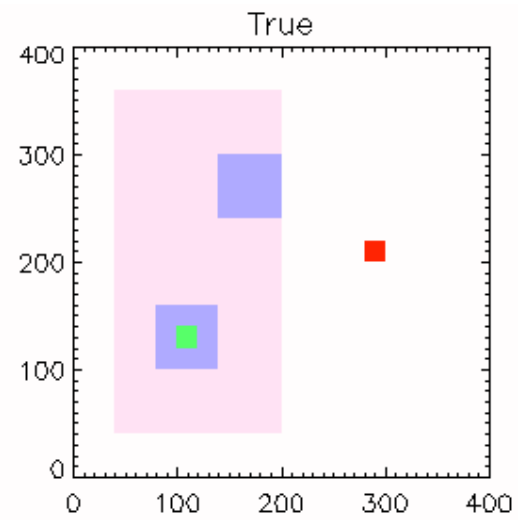
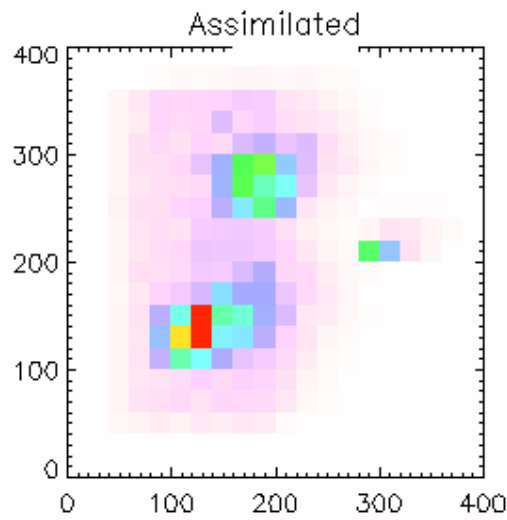
concentrations



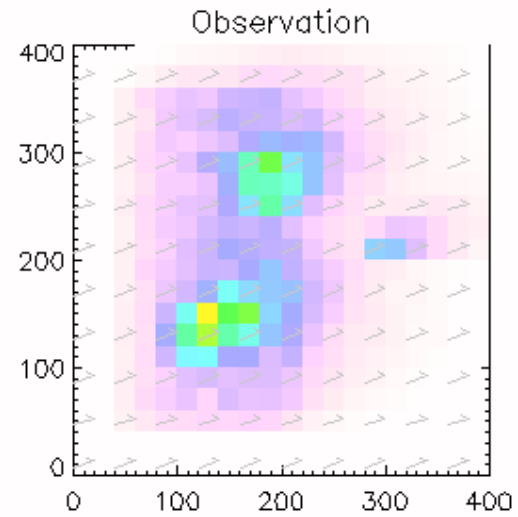
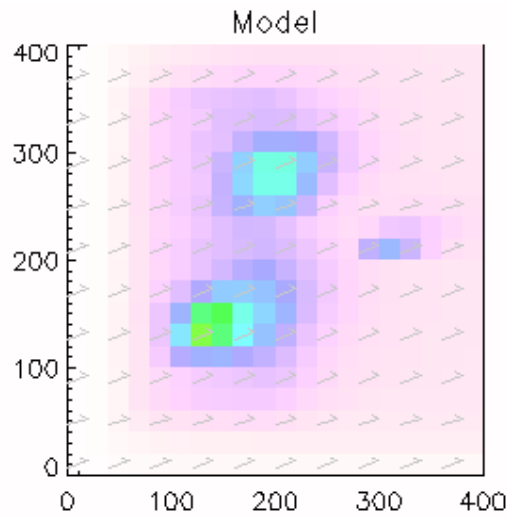
Local, linear

(3/20)

emissions

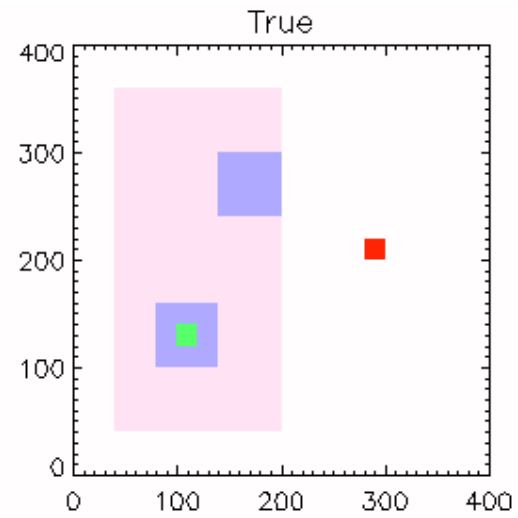
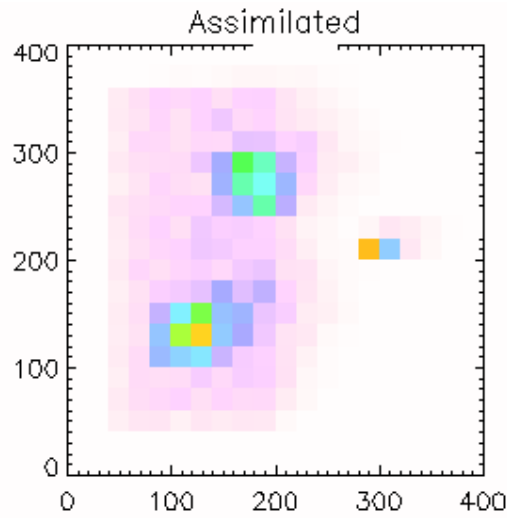


concentrations

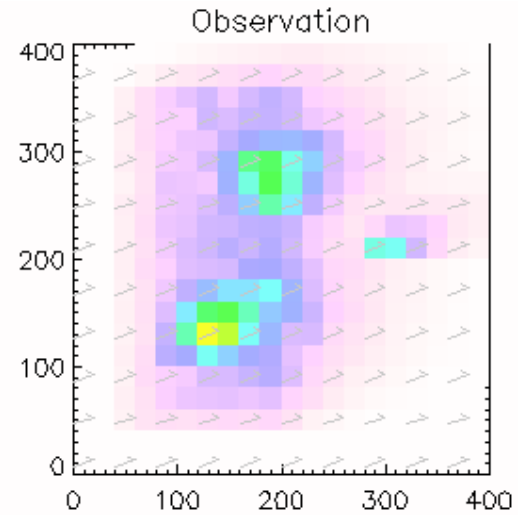
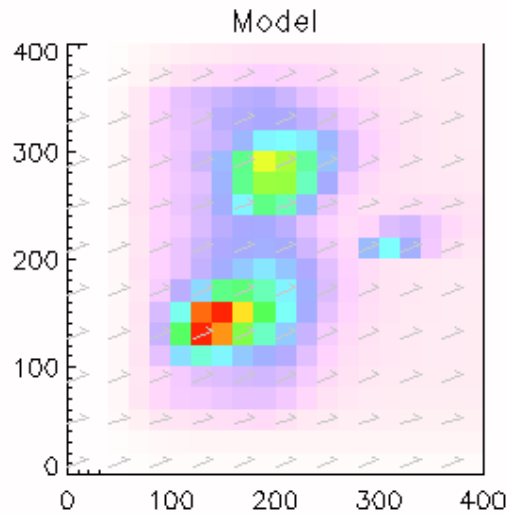


Local, linear

emissions

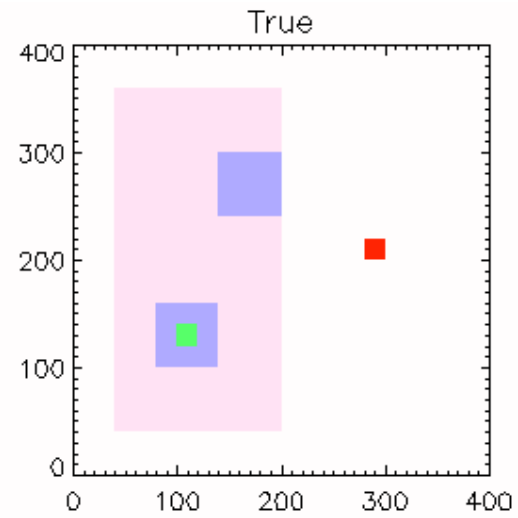
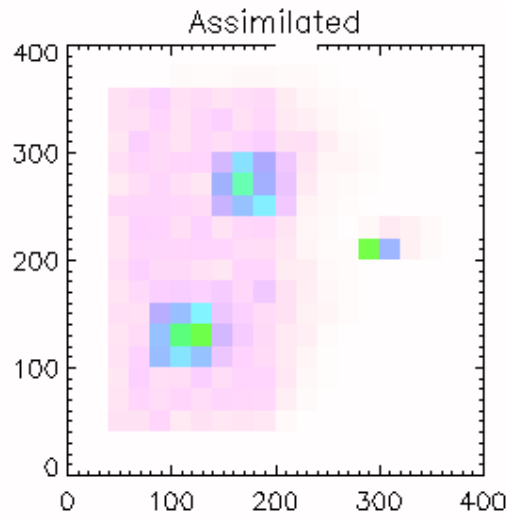


concentrations

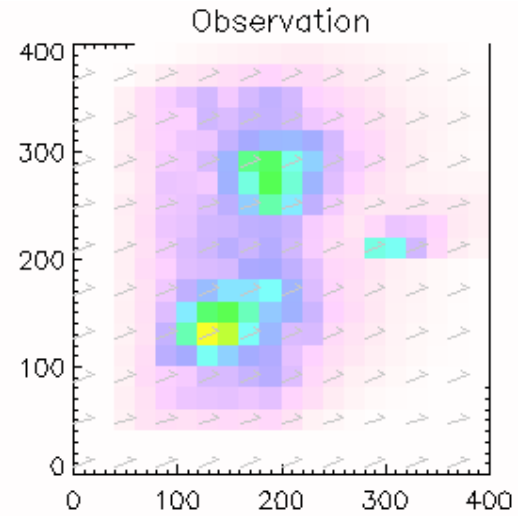
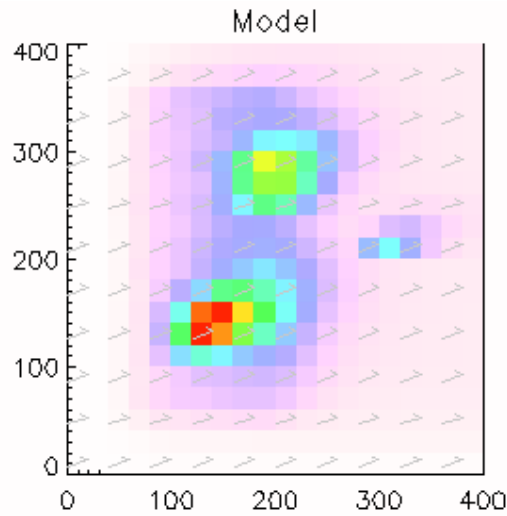


Local, linear

emissions

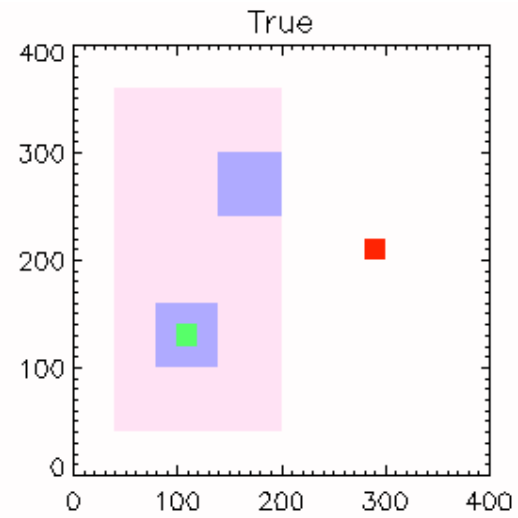
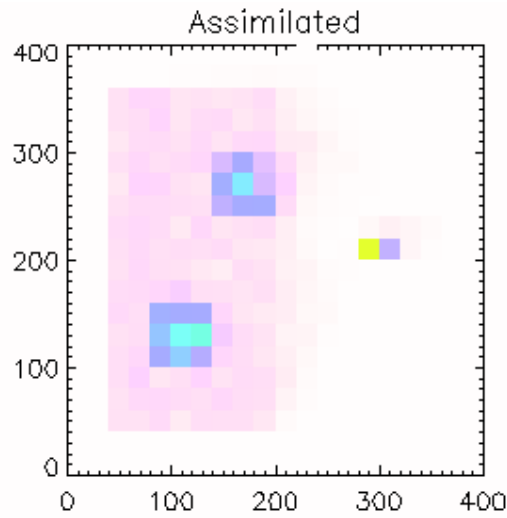


concentrations

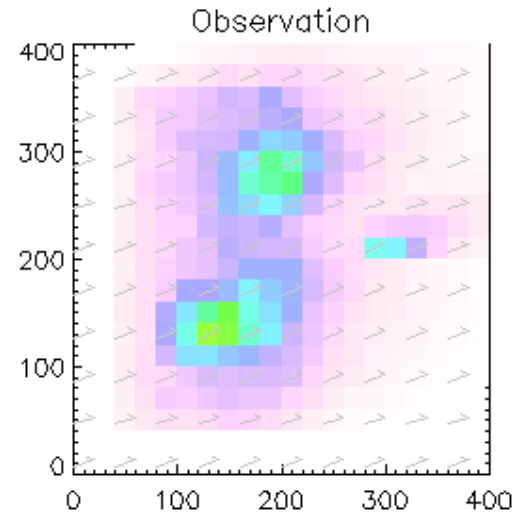
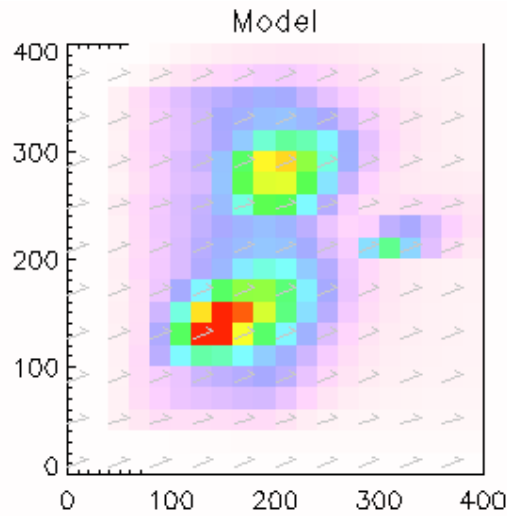


Local, linear

emissions



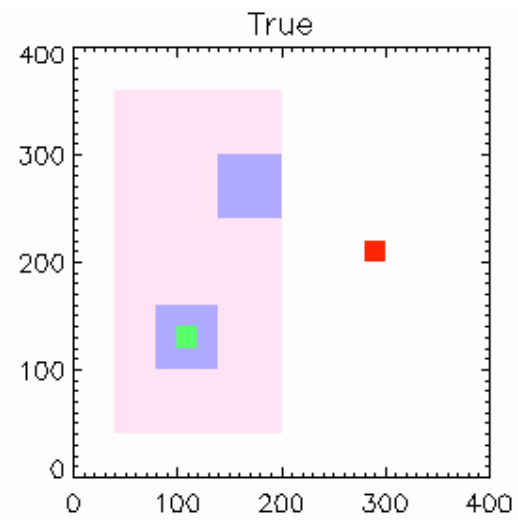
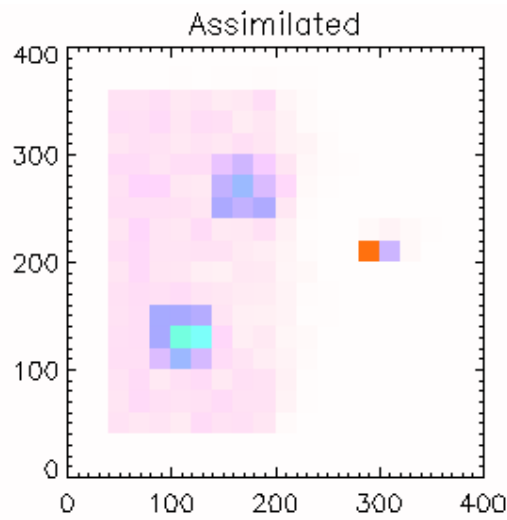
concentrations



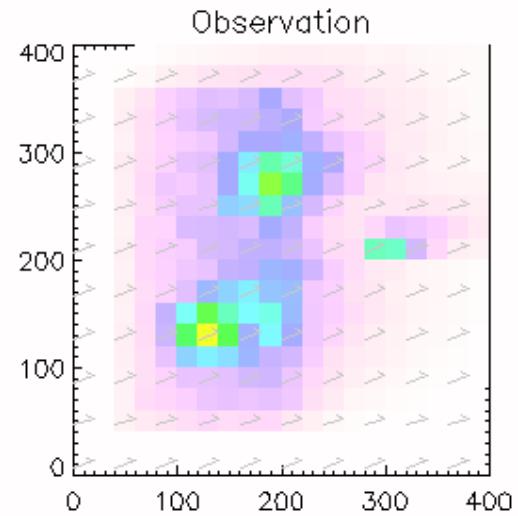
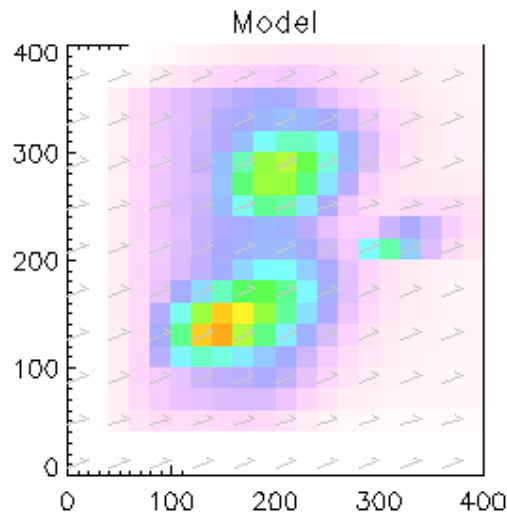
Local, linear

(7/20)

emissions

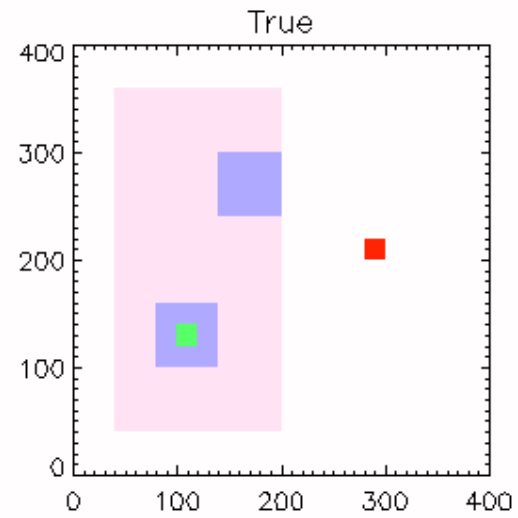
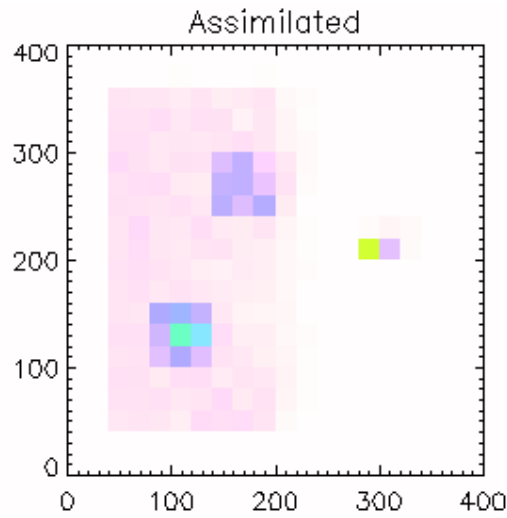


concentrations

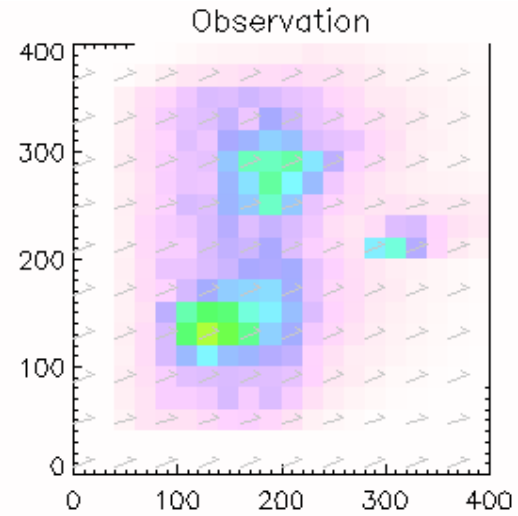
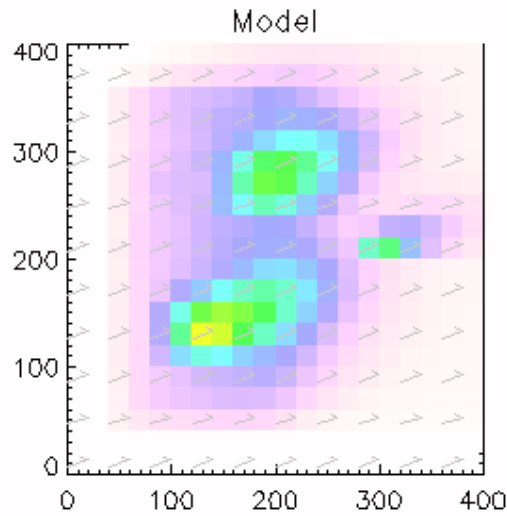


Local, linear

emissions

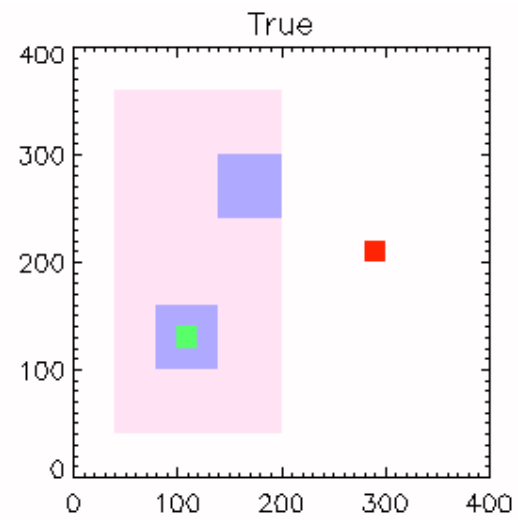
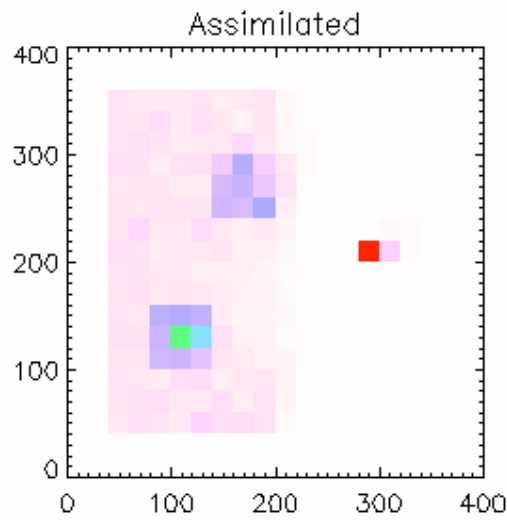


concentrations

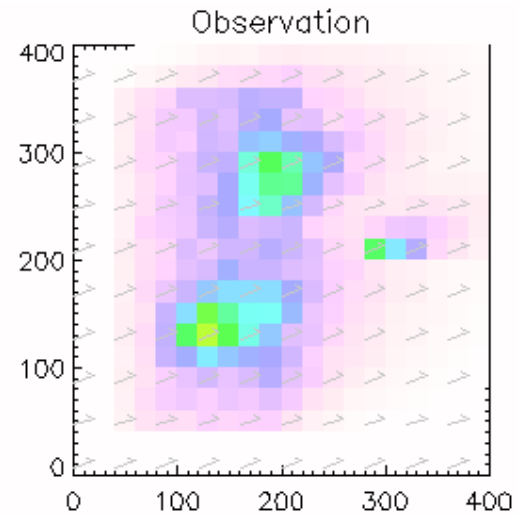
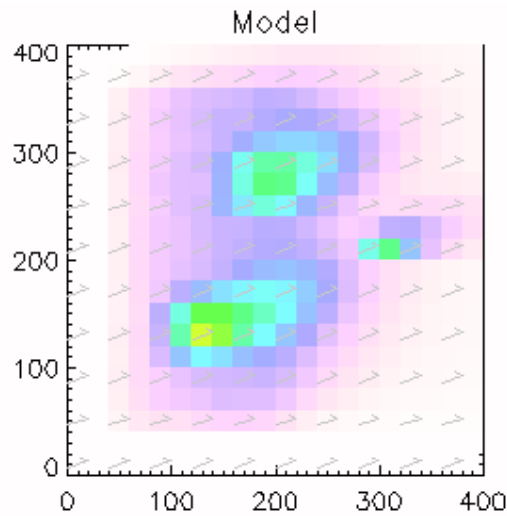


Local, linear

emissions



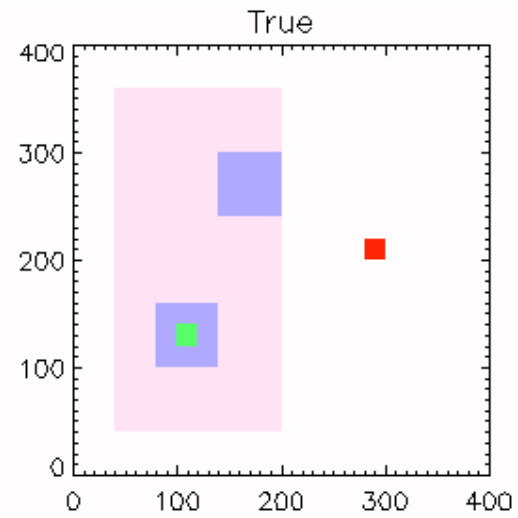
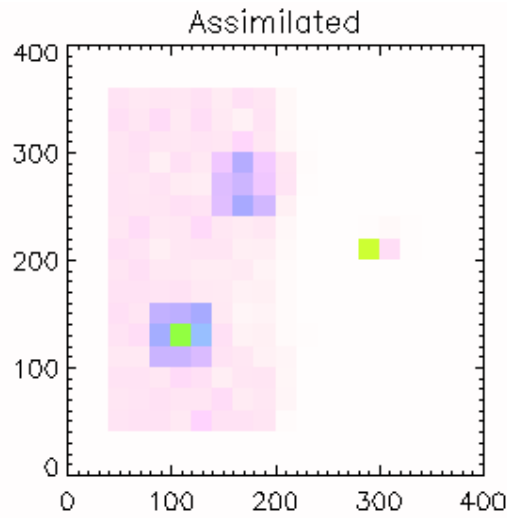
concentrations



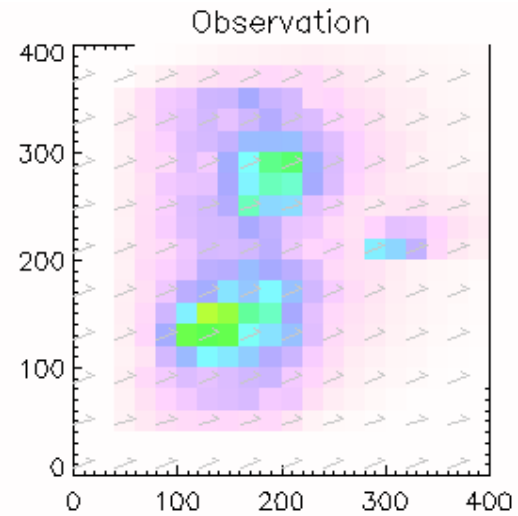
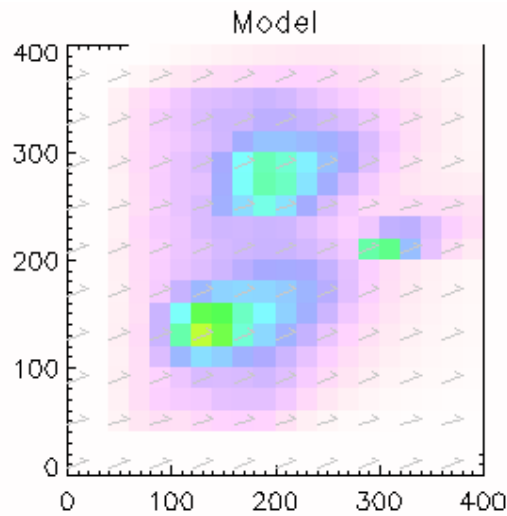
Local, linear

(10/20)

emissions

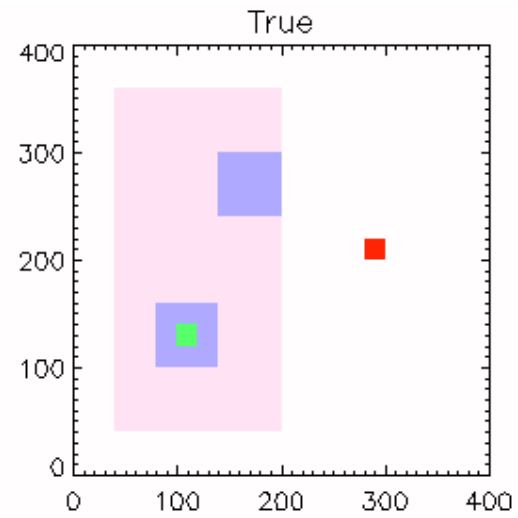
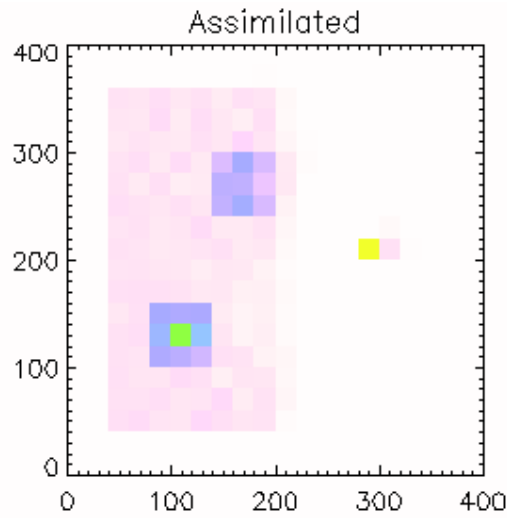


concentrations

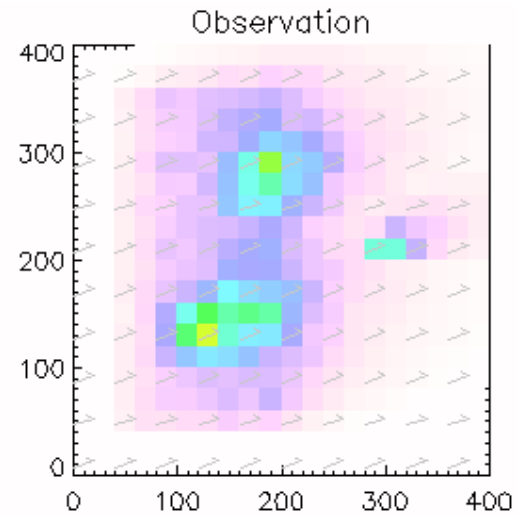
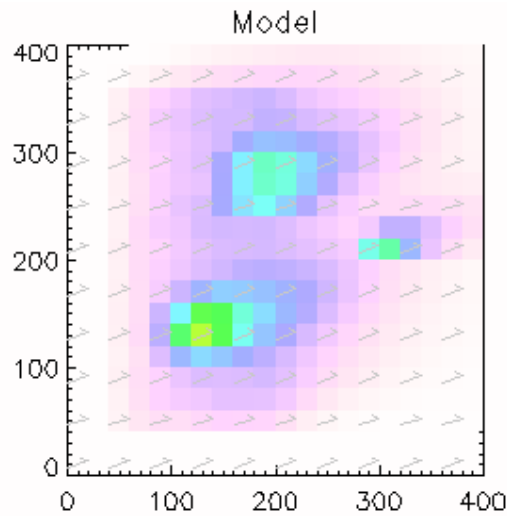


Local, linear

emissions



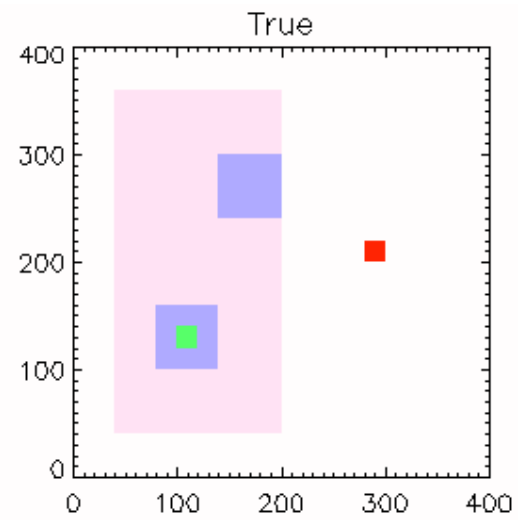
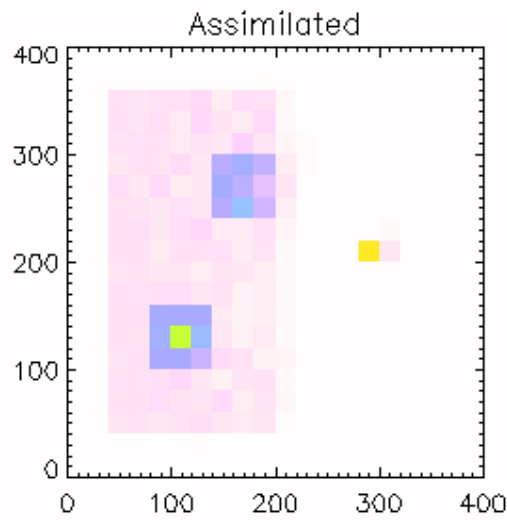
concentrations



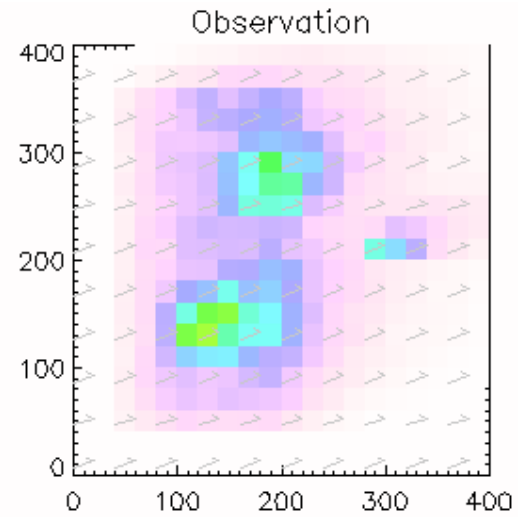
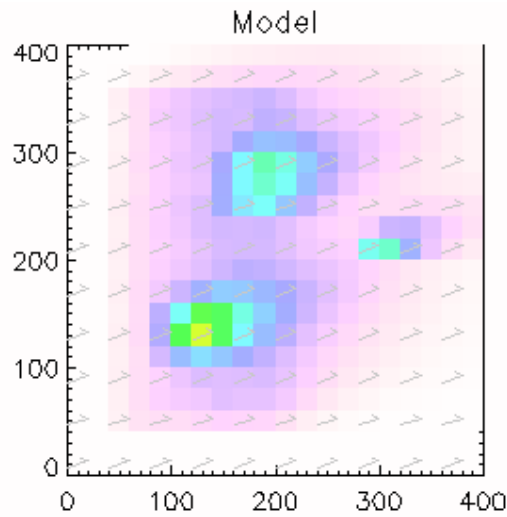
Local, linear

(12/20)

emissions



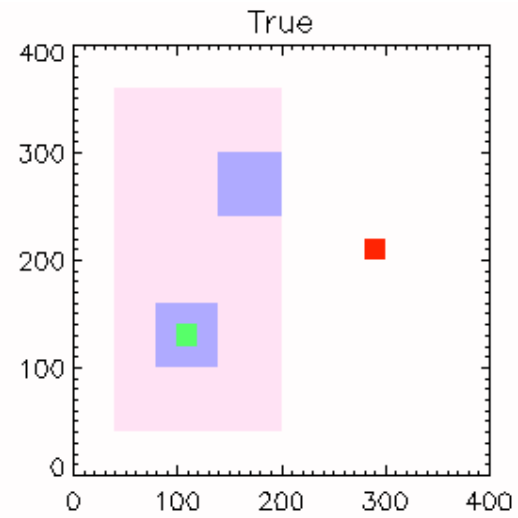
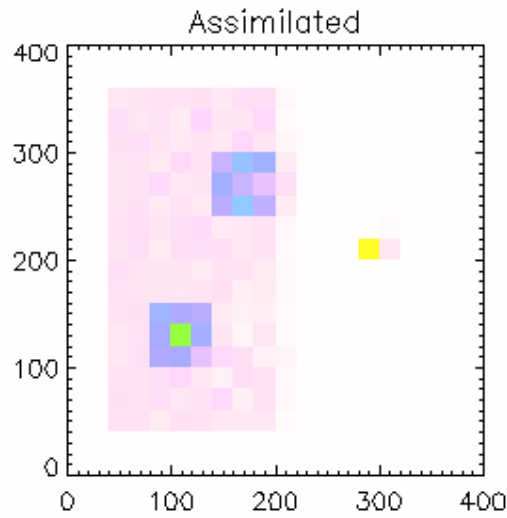
concentrations



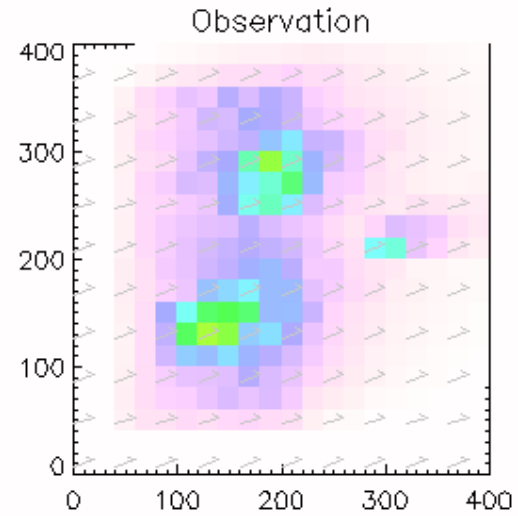
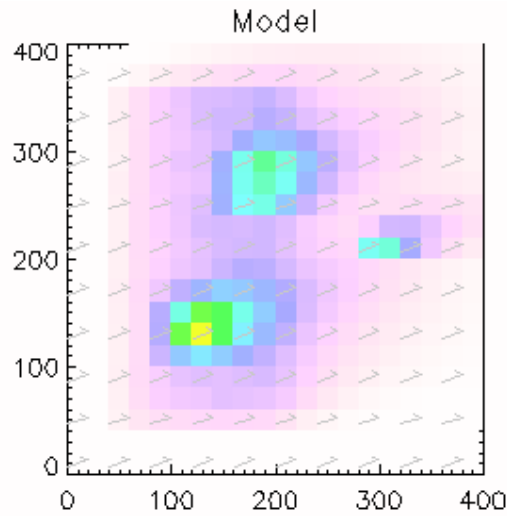
Local, linear

(13/20)

emissions



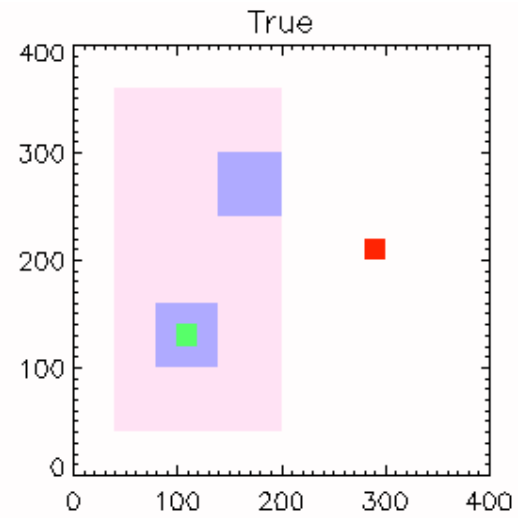
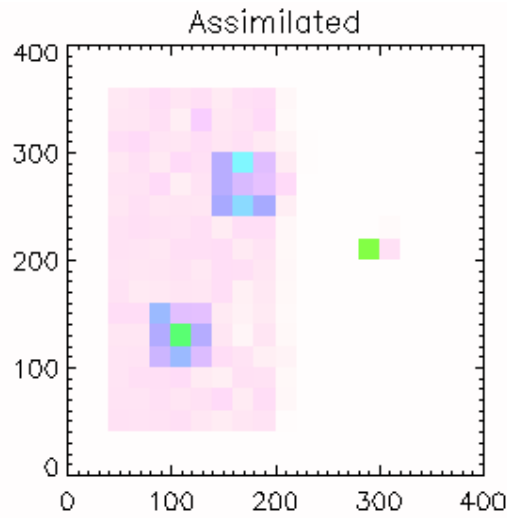
concentrations



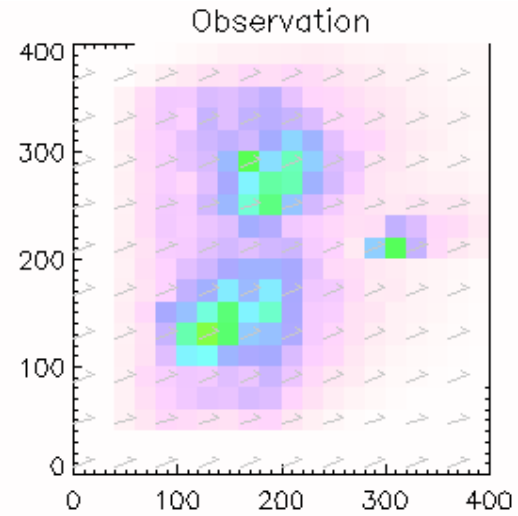
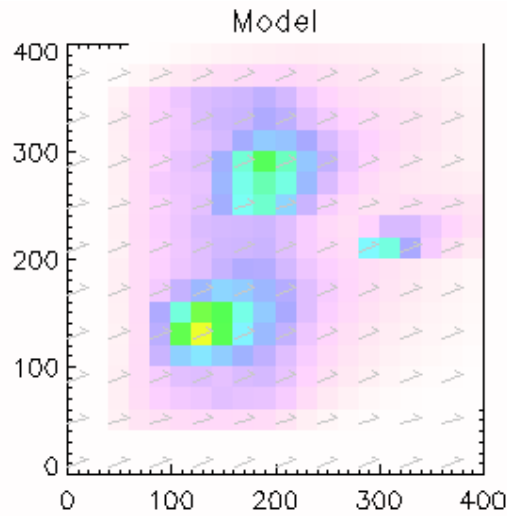
Local, linear

(14/20)

emissions



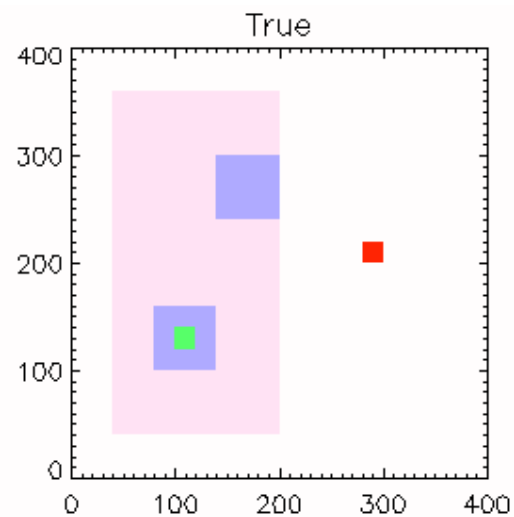
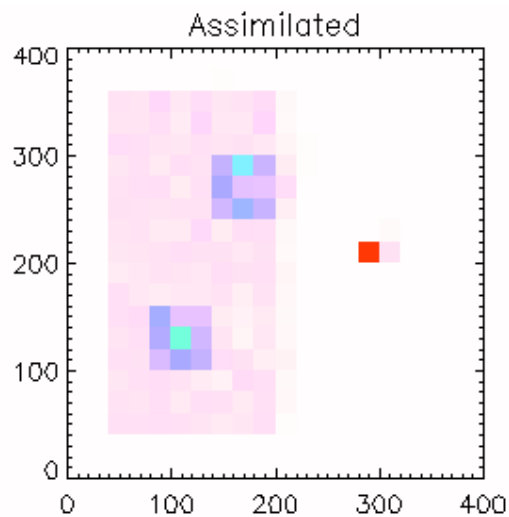
concentrations



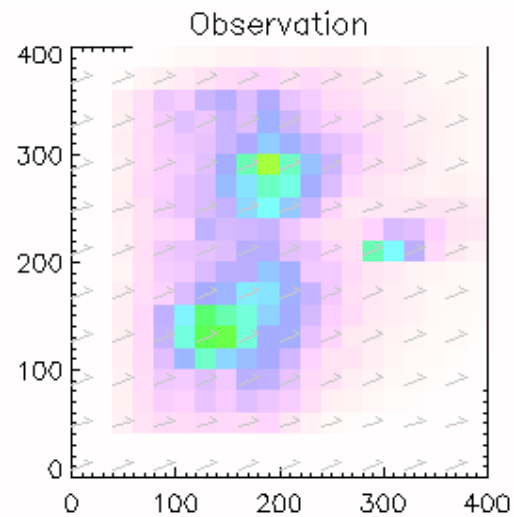
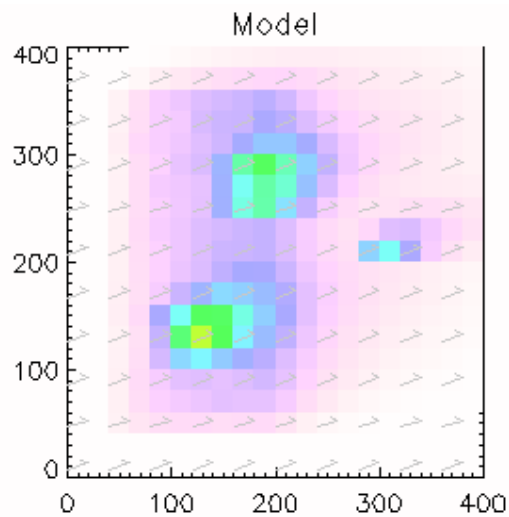
Local, linear

(15/20)

emissions

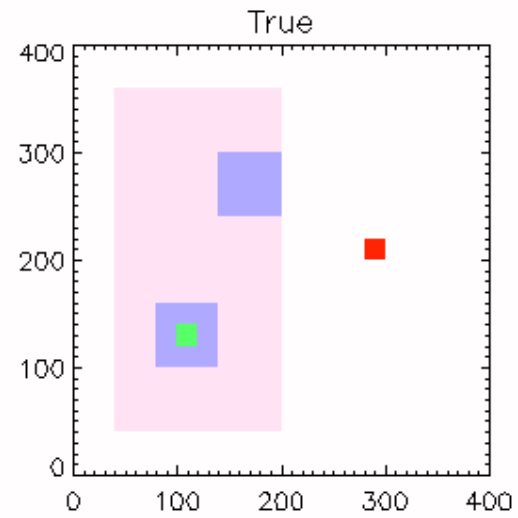
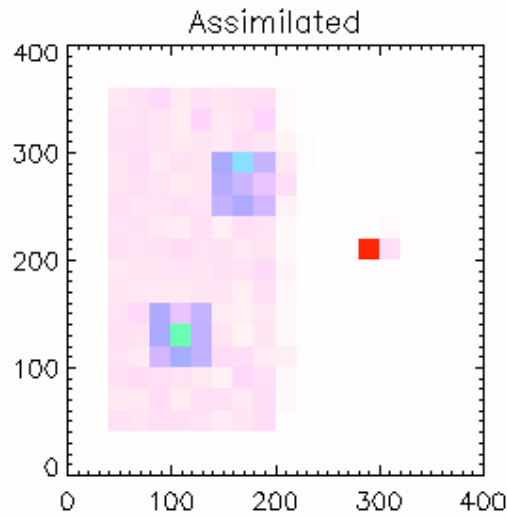


concentrations

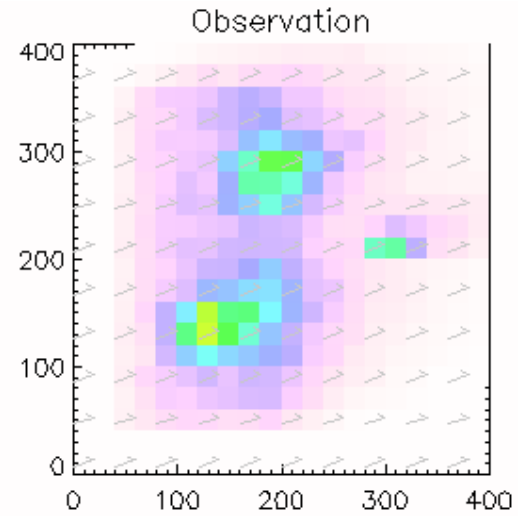
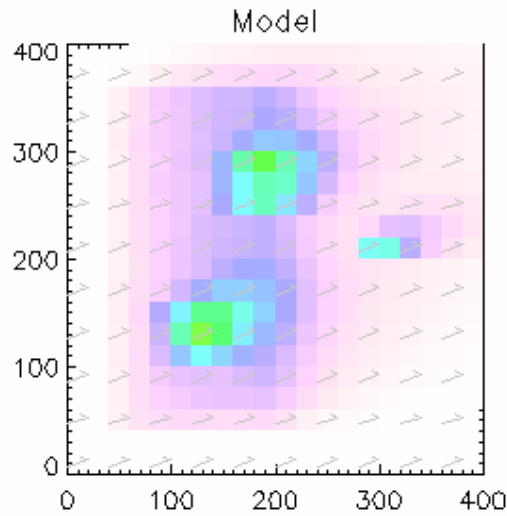


Local, linear

emissions



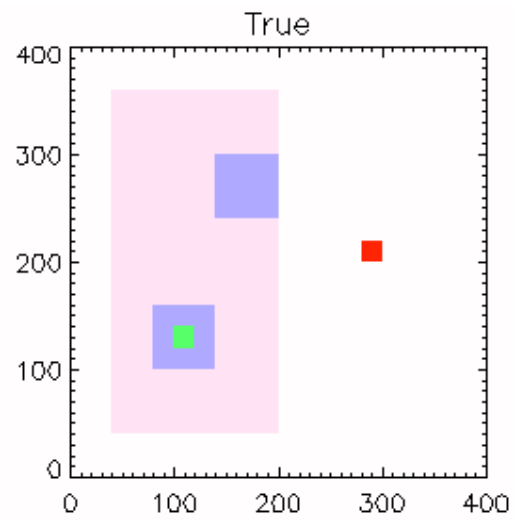
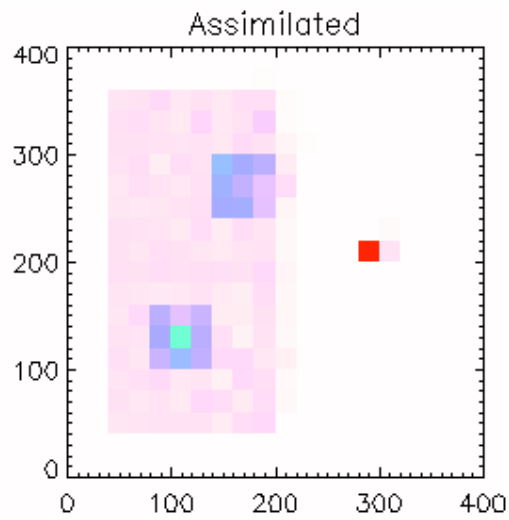
concentrations



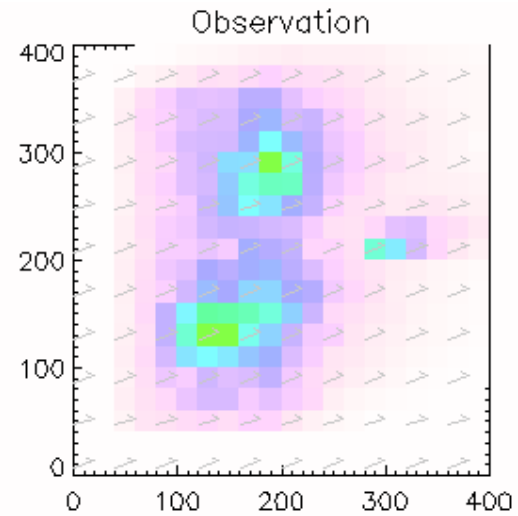
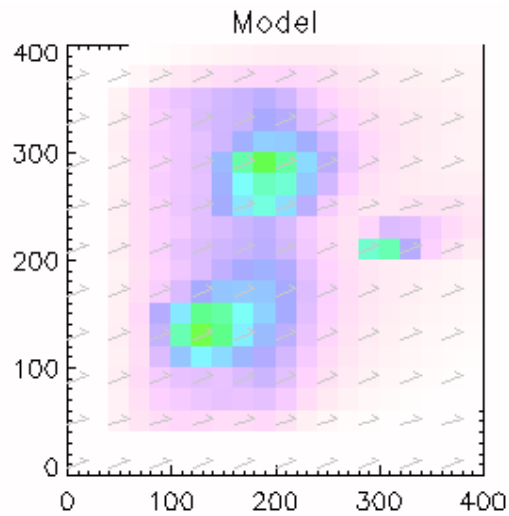
Local, linear

(17/20)

emissions



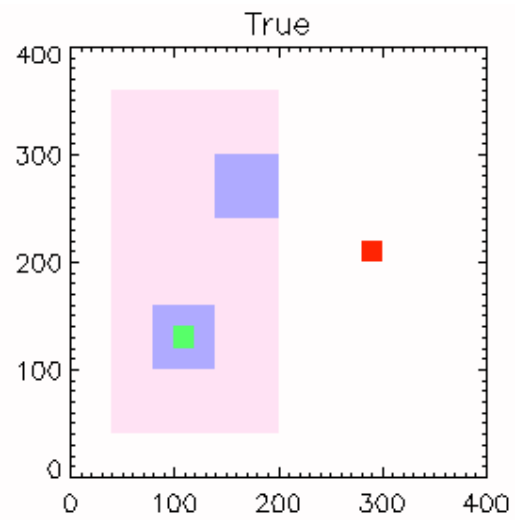
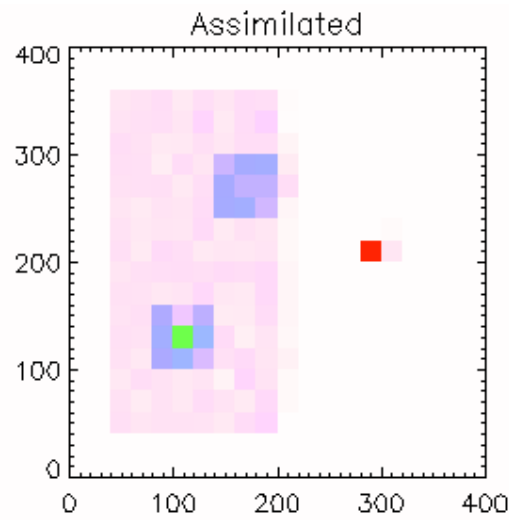
concentrations



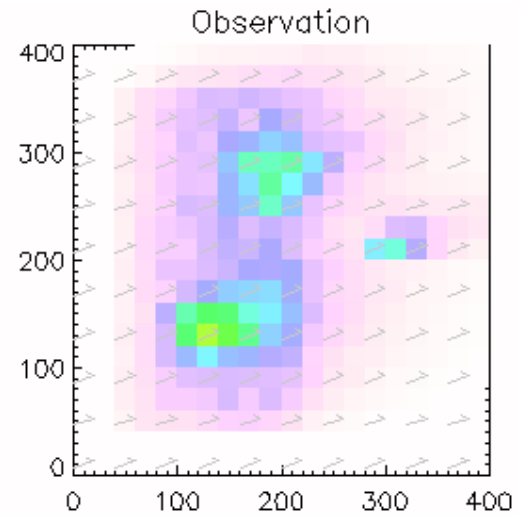
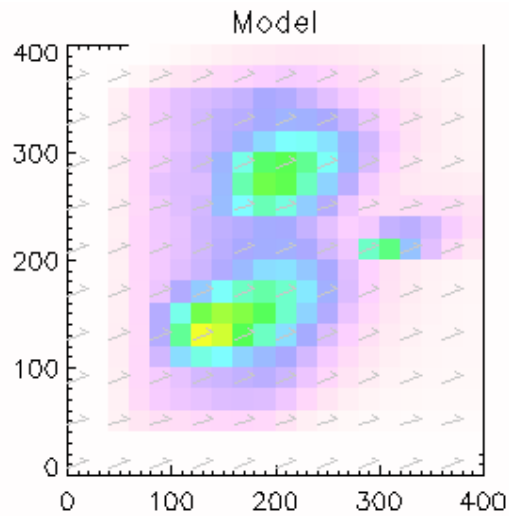
Local, linear

(18/20)

emissions



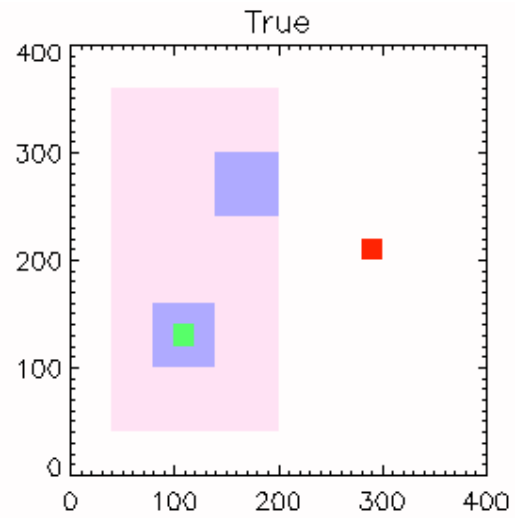
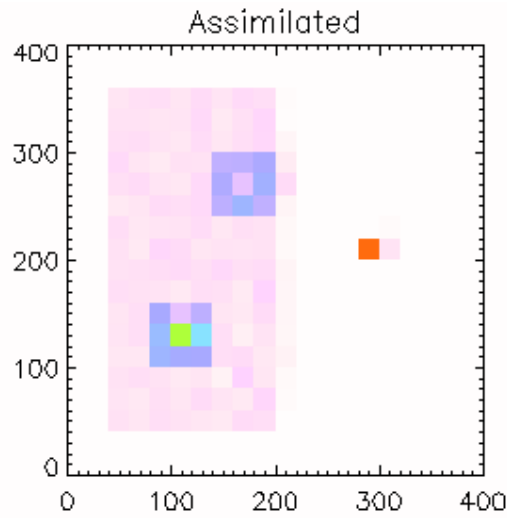
concentrations



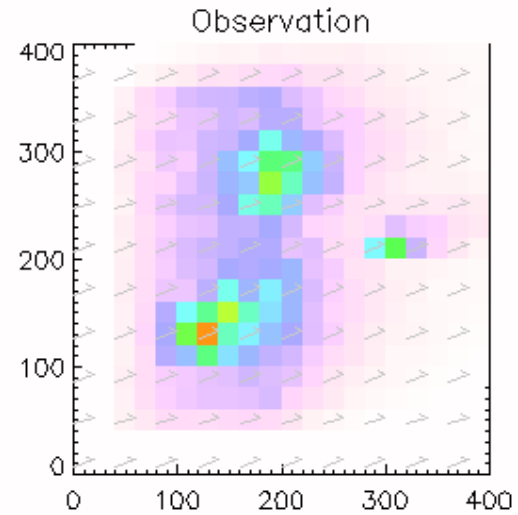
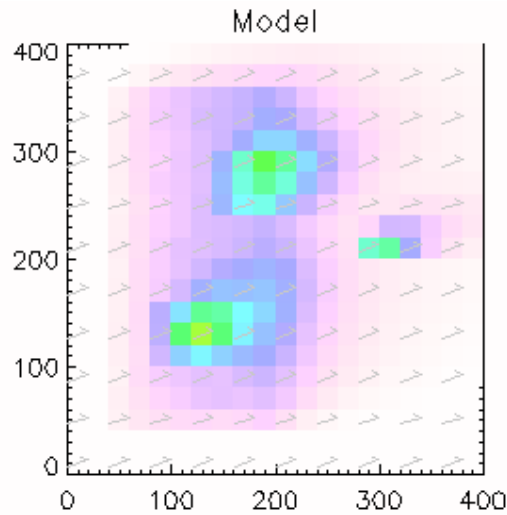
Local, linear

(19/20)

emissions



concentrations

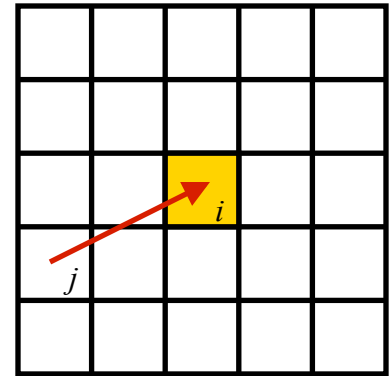


Local, linear

(20/20)

More realistic inversion: Sensitivities

When transport is taken into account, emissions in all grid cells can contribute to the observed concentration:



$$\Delta\Omega_i = \sum_j \alpha_{j \rightarrow i} \Delta E_j, \quad \alpha_{j \rightarrow i} = \frac{\partial \Omega_i}{\partial E_j}$$

Monte Carlo method

Konovalov et al. (2006), Inverse modeling of NOx emission on a continental scale (2006), ACP

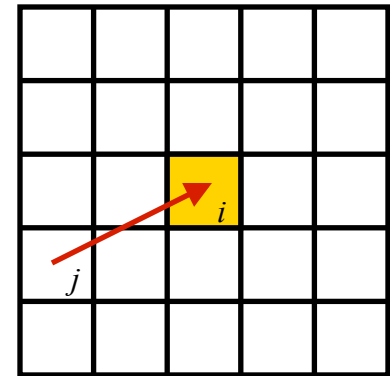
Perform model runs with random perturbations on the *a priori* emissions to get a set of linear equations from which the sensitivities can be solved:

$$\Delta\Omega_1^i = \sum_j \alpha_{j \rightarrow i} \Delta E_1^j$$

$$\Delta\Omega_2^i = \sum_j \alpha_{j \rightarrow i} \Delta E_2^j$$

...

$$\Delta\Omega_N^i = \sum_j \alpha_{j \rightarrow i} \Delta E_N^j$$



The optimal number of random model runs depends on the desired accuracy.
For two next neighbours: $N = 100$

Monte Carlo method

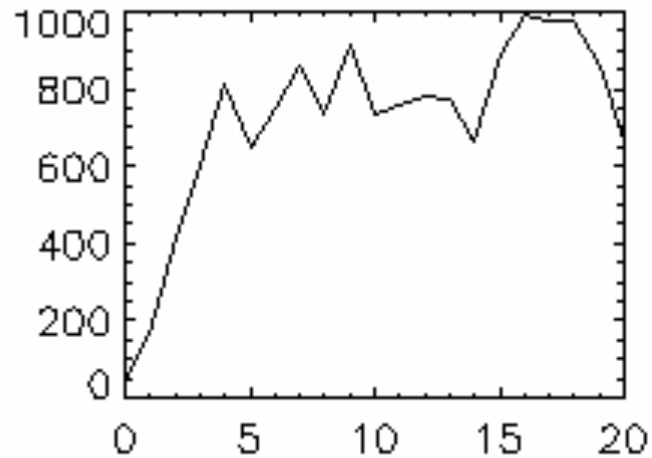
Advantages

- Takes transport of nearest neighbours into account

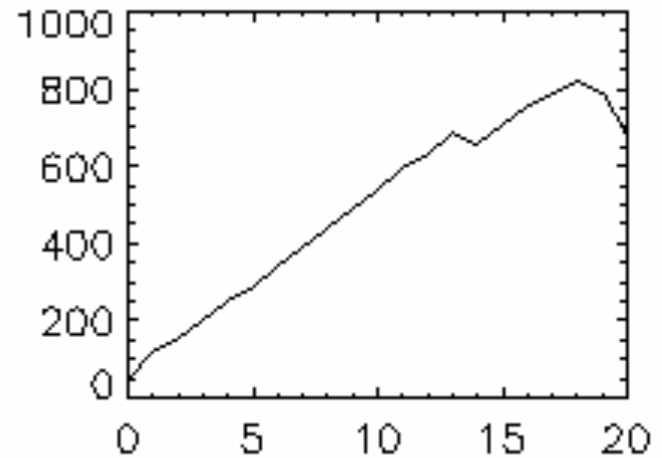
Disadvantages

- Time consuming calculations:
~100 model runs needed to solve transport from 2 nearest neighbours.

Convergence behaviour



Local, linear



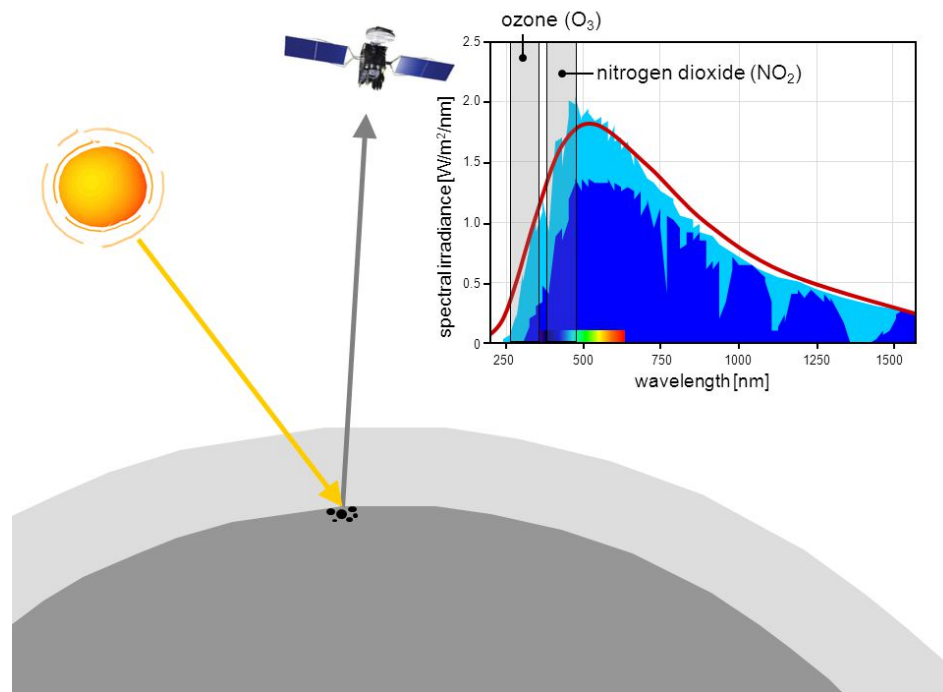
Kalman

Top-Down Approach for NO_x- emission over India

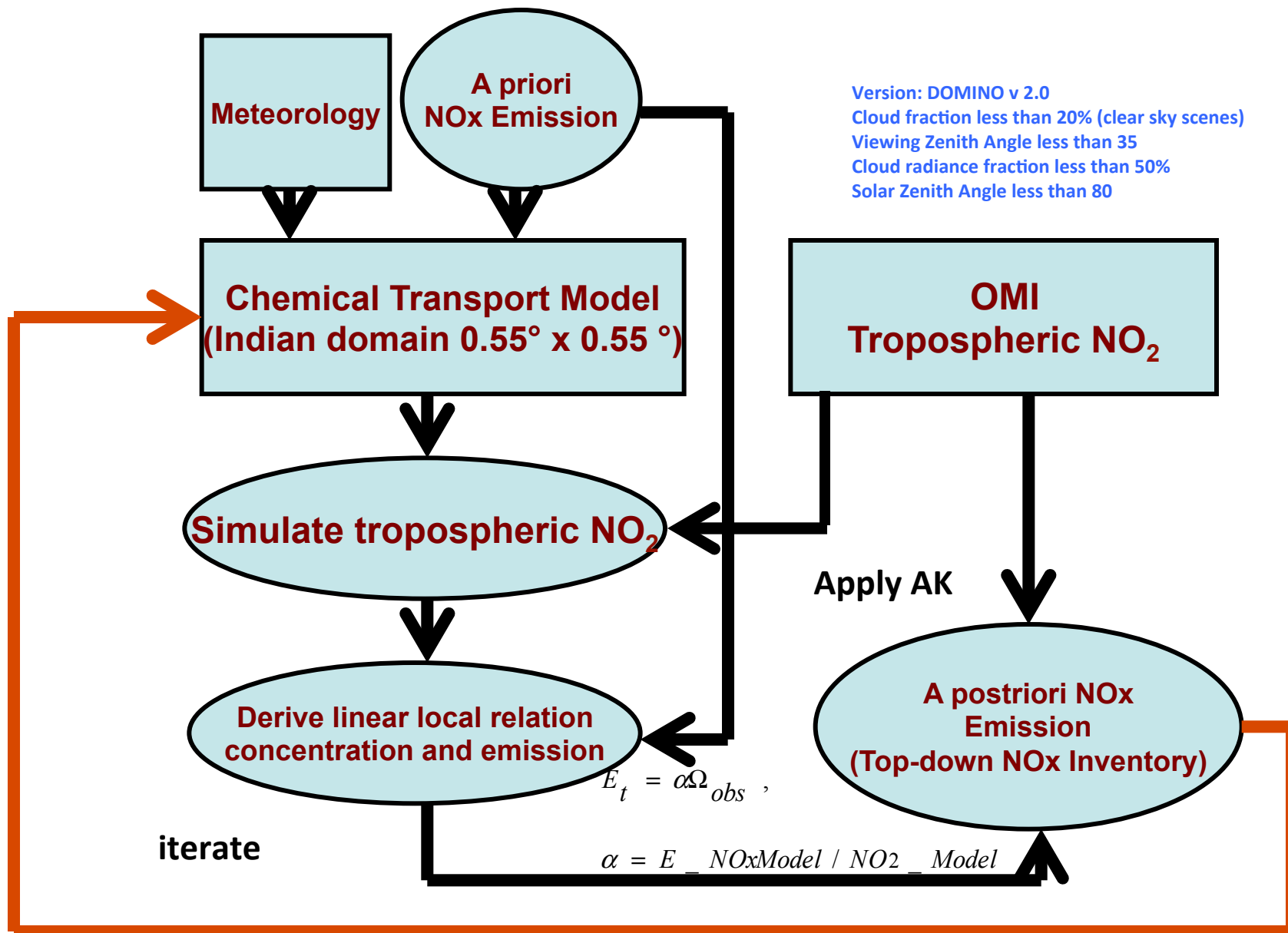
- **What we Need?**
- Satellite observations of NO₂
- Chemical Transport Model

Tropospheric NO₂ column is derived in three main steps involving the calculation of

- (1) slant column (using Differential Optical Absorption Spectroscopy (DOAS) approach in the 405–465 nm spectral window),
- (2) tropospheric slant column (using modeling/ assimilation approach),
- (3) tropospheric vertical column (using air mass factor—AMF).



Top-Down Approach for NOx- emission over India



OMI Tropospheric NO₂ columns for the year 2005 Over India

Version: DOMINO v 2.0 (Level 2 data set)

Cloud fraction less than 20% (clear sky scenes)

Viewing Zenith Angle less than 35 (pixel size smaller than 34x14km²)

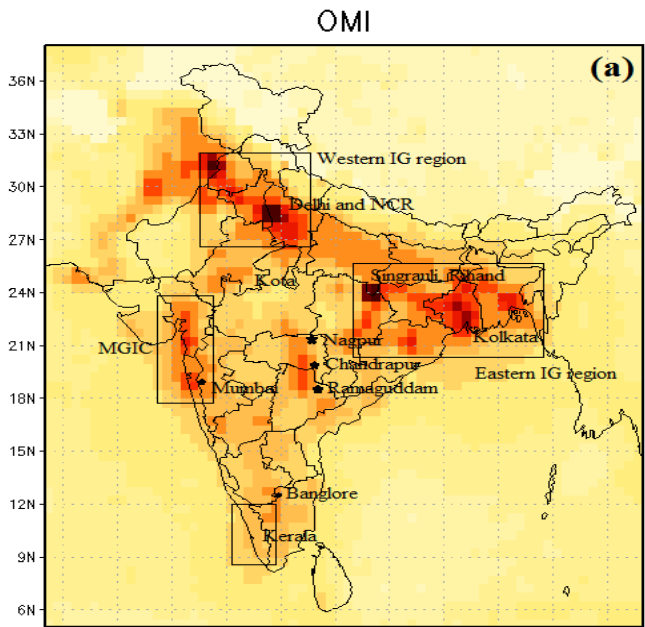
Cloud radiance fraction less than 50%

Solar Zenith Angle less than 80

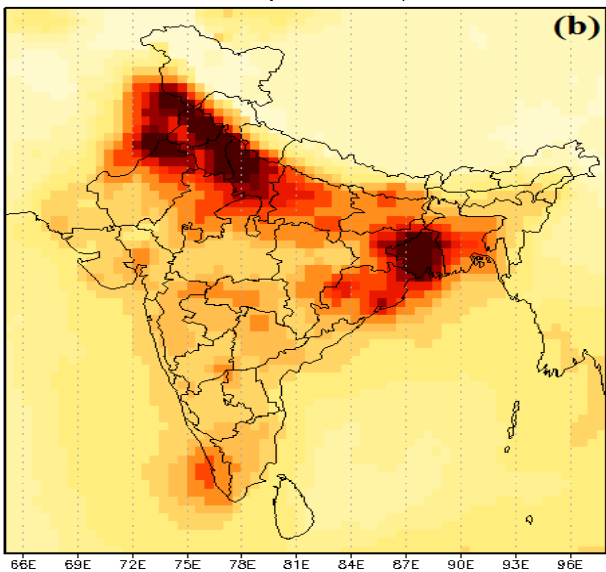
WRF-Chem Simulation for Jan 2005

Domain	: South Asia (0 - 45° N, 55 -110 ° E)
Period	: 2005
Resolution	: 55 km x 55 km
Emissions	: INTEX-B (A Priori)
Fire Emission	: NCAR Fire Inventory (FINN)
Biogenic	: MEGAN
Gas Ph. Chem	: MOZART
Aerosol Ph. Chem	: GOCART

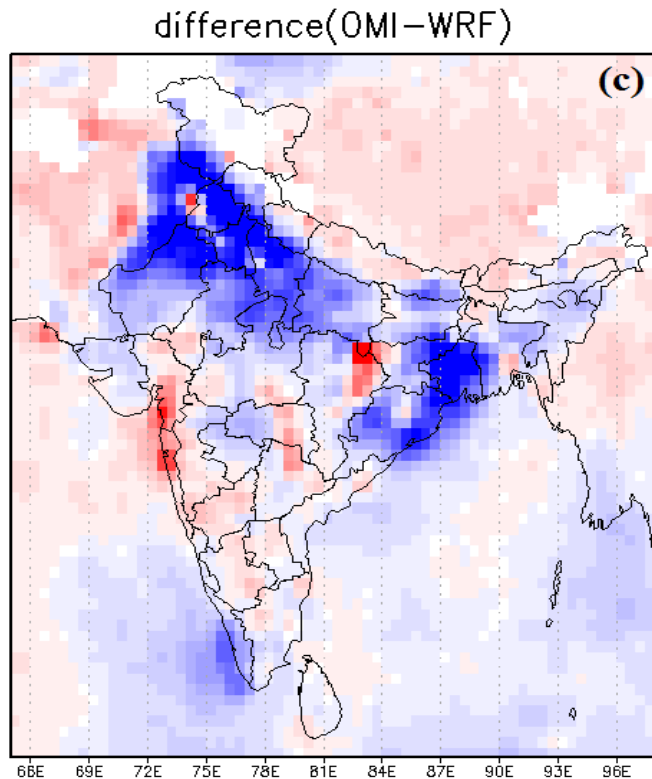
WRF simulation with *a priori* emission (Intex-B)



WRF(Intex-B)



(b)



(c)

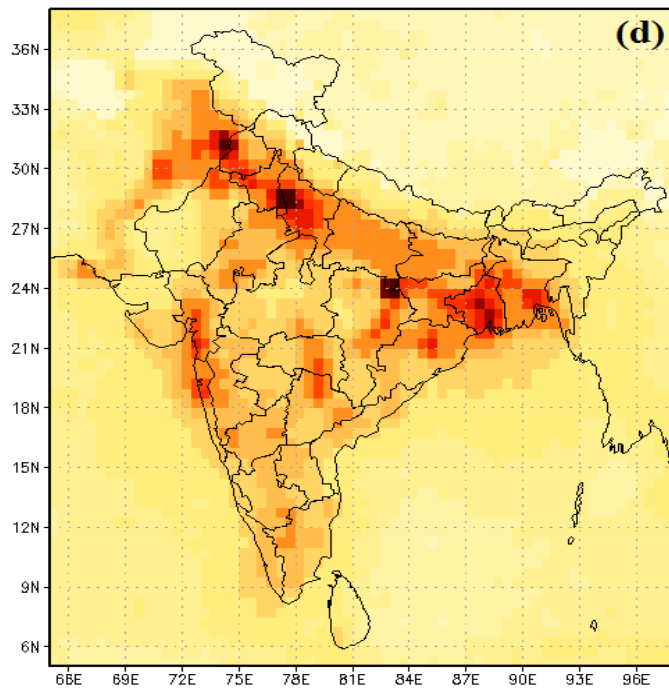
With OMI Averaging Kernel



Correlation $R^2 = 0.68$

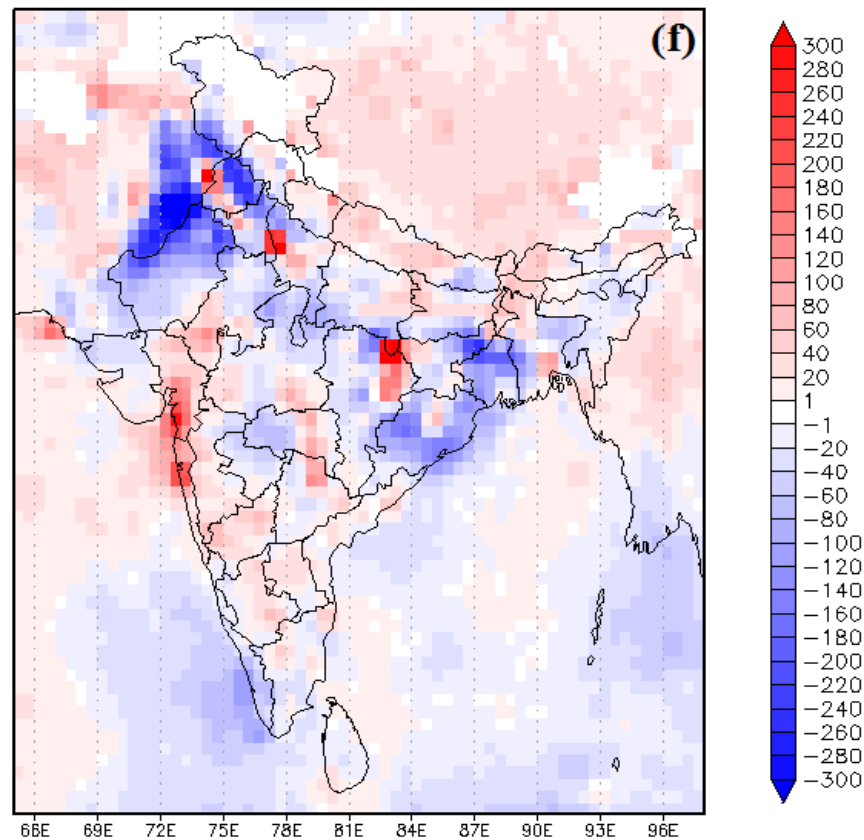
Spatial distribution of OMI and WRF-Chem tropospheric NO₂ column for 2005 and its difference. Tropospheric column NO₂ unit is 10¹³ molecules/cm²/s.

OMI

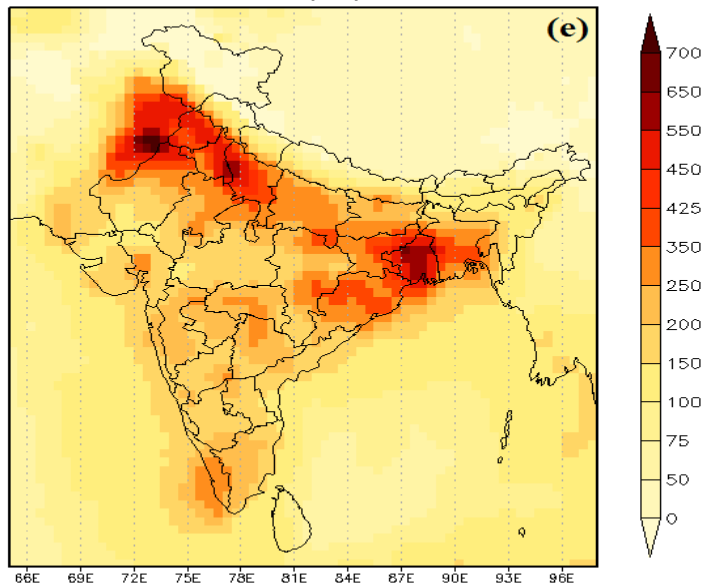


WRF simulation with *a postrioro* emission (T1)

difference(OMI-WRF)



WRF(T1)



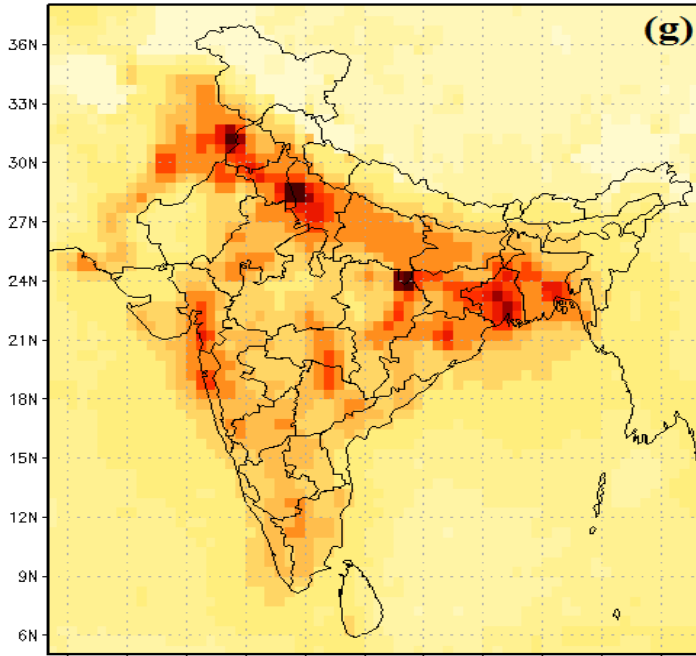
Correlation $R^2 = 0.76$

With OMI Averaging Kernel

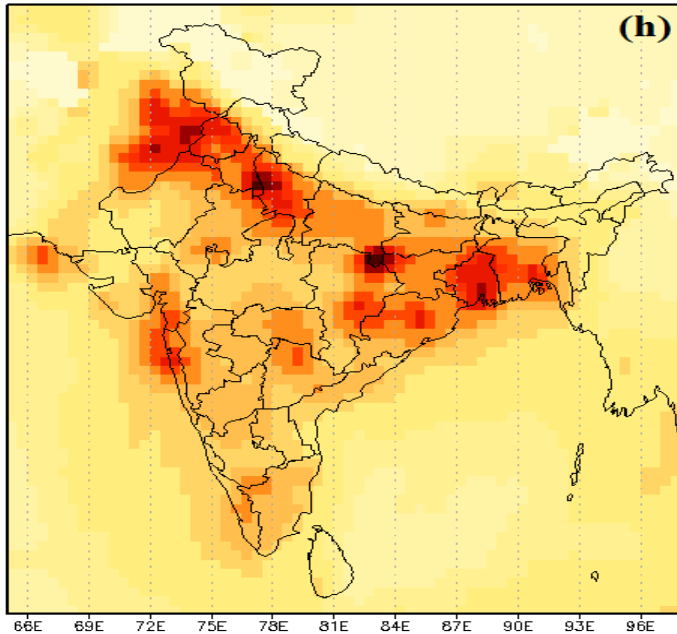


WRF simulation with *a postrioro* emission (T7)

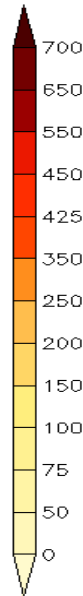
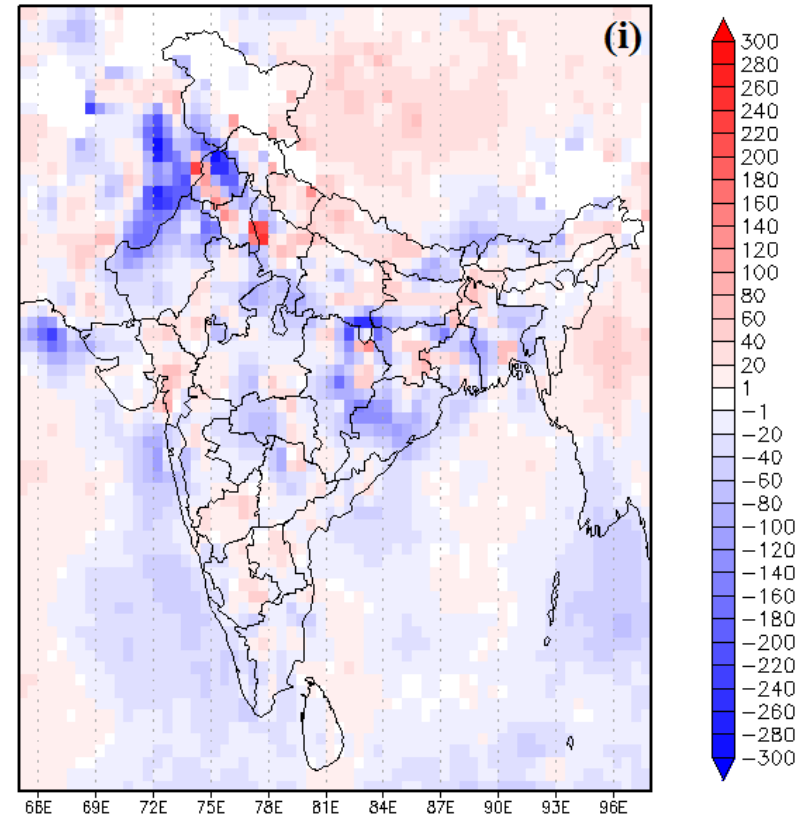
OMI



WRF(T7)



difference(OMI-WRF)

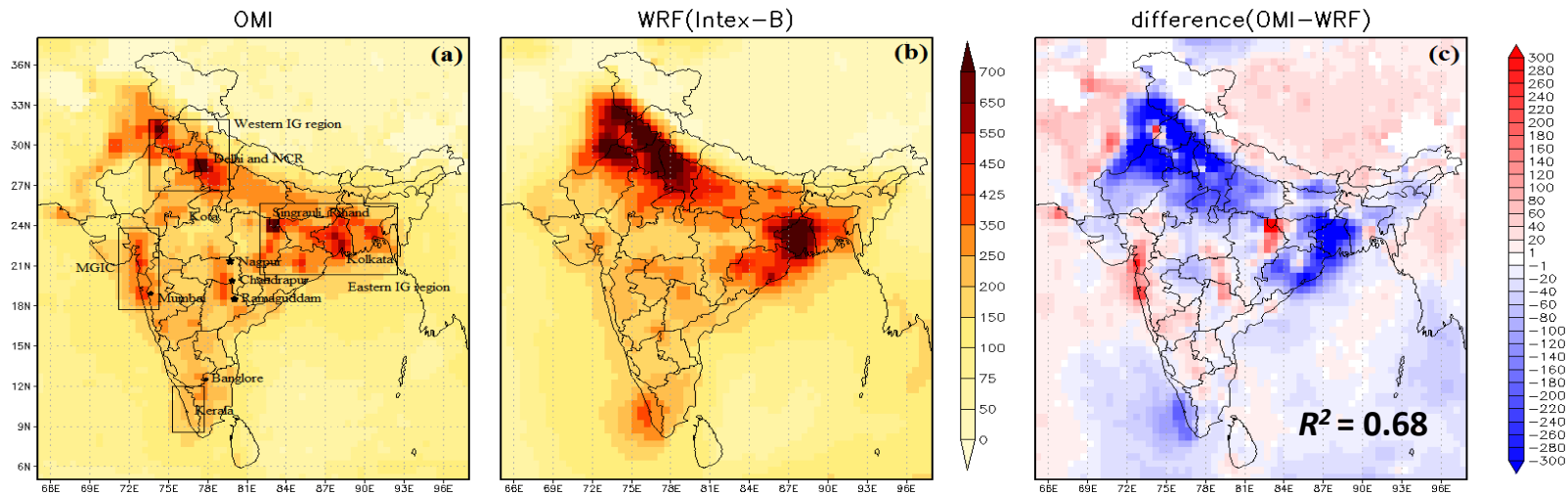


With OMI Averaging Kernel

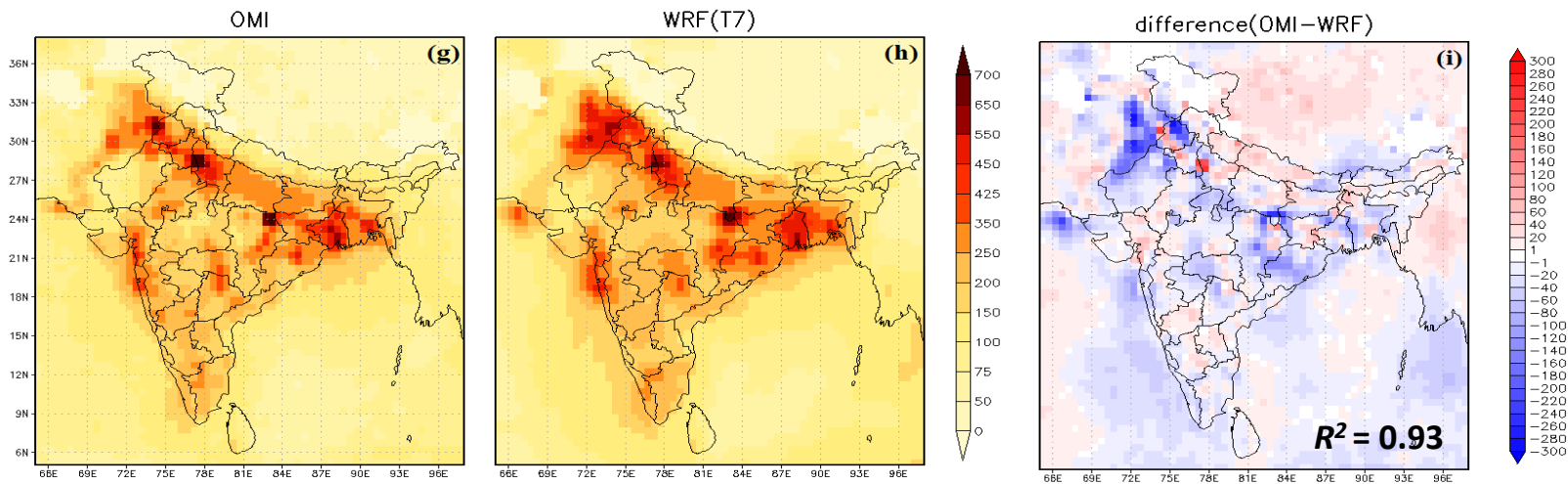


Correlation $R^2 = 0.93$

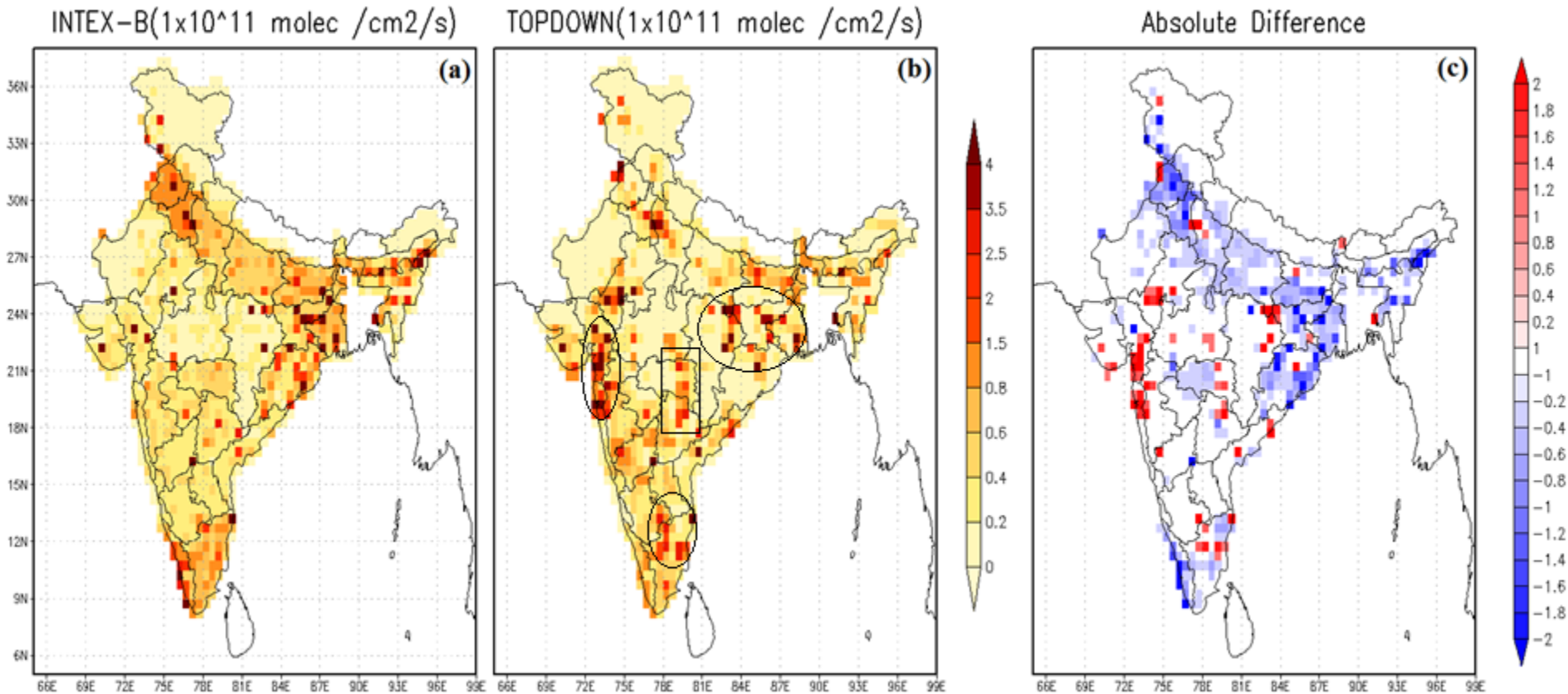
WRF simulation with *a priori* emission (Intex-B)



WRF simulation with *a posteriori* emission (T7)



A priori versus A posteriori Emissions



Spatial distribution of NO_x emission from (a) INTEX-B inventory, (b) optimized top-down Inventory and (c) their difference (emission unit is 10¹¹ NO molecules/cm²/s)

A priori emission

1.56Tg N /year

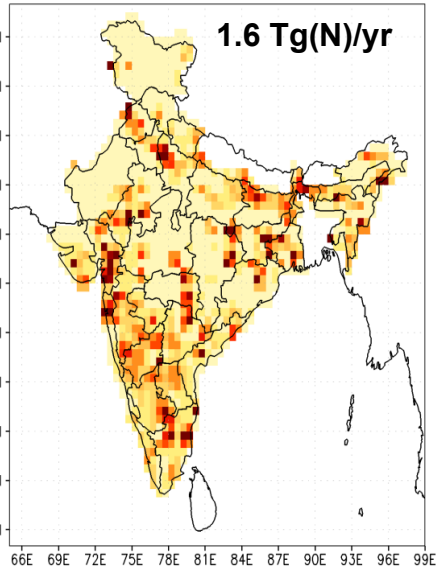
A posteriori emission

1.42Tg N /year

Seasonal Top-down Emission

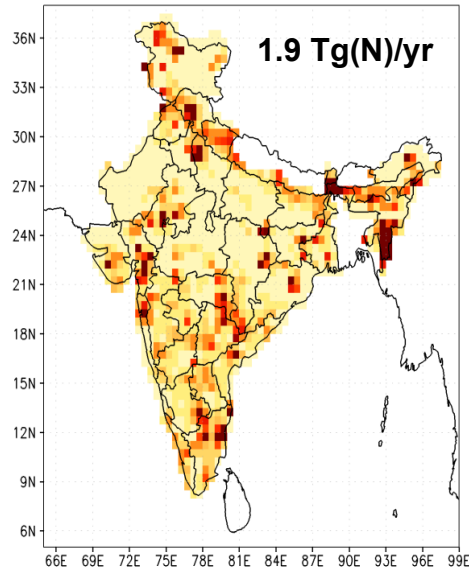
Winter

1.6 Tg(N)/yr



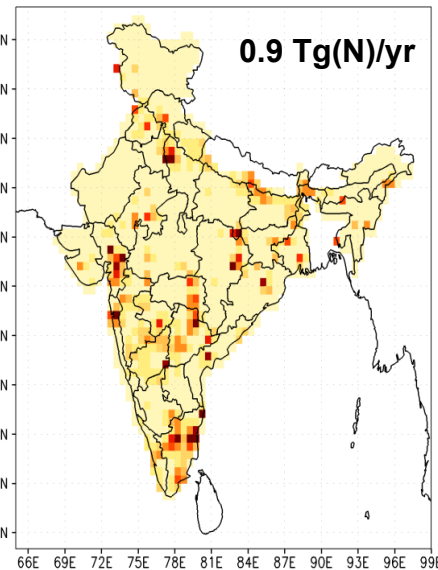
Pre Monsoon

1.9 Tg(N)/yr



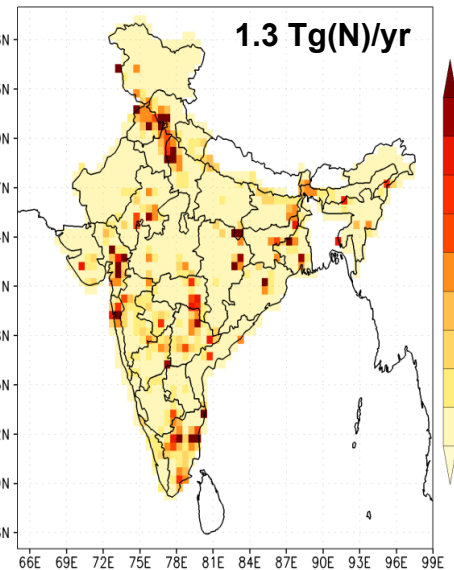
Monsoon

0.9 Tg(N)/yr



Post Monsoon

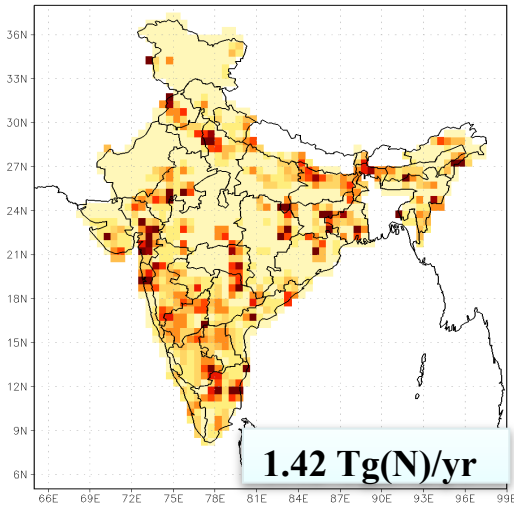
1.3 Tg(N)/yr



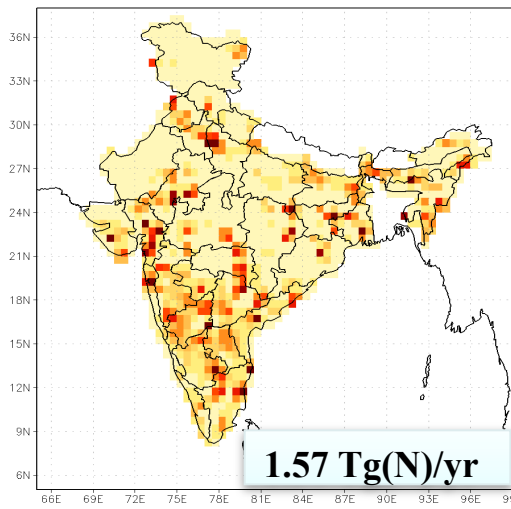
Spatial distribution NO_x emissions from optimized Top-Down inventory (emission unit is 10^{11} NO molecules/cm²/s) for winter, pre-monsoon, monsoon and post-monsoon seasons for 2005.

Annual Top-down Emission

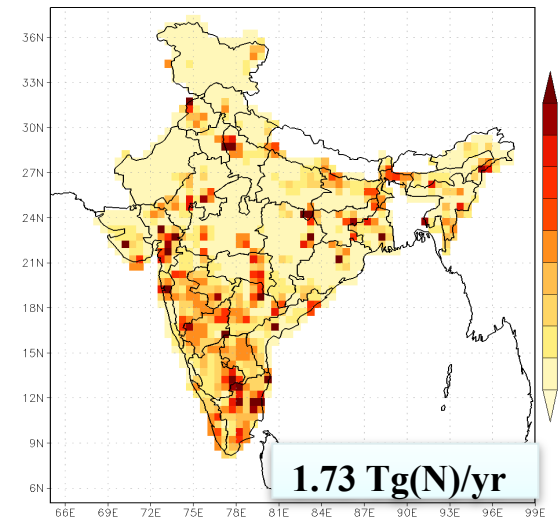
2005



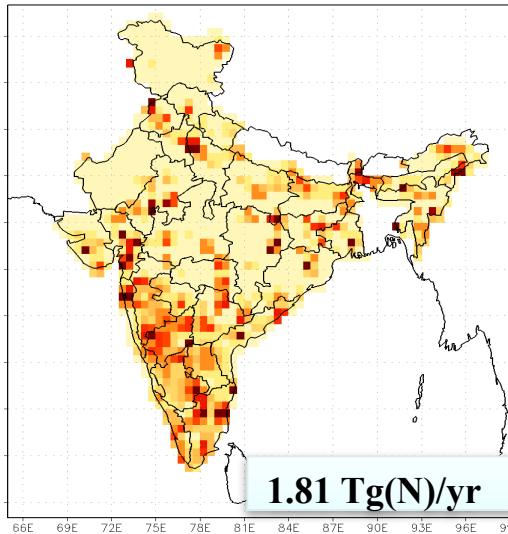
2006



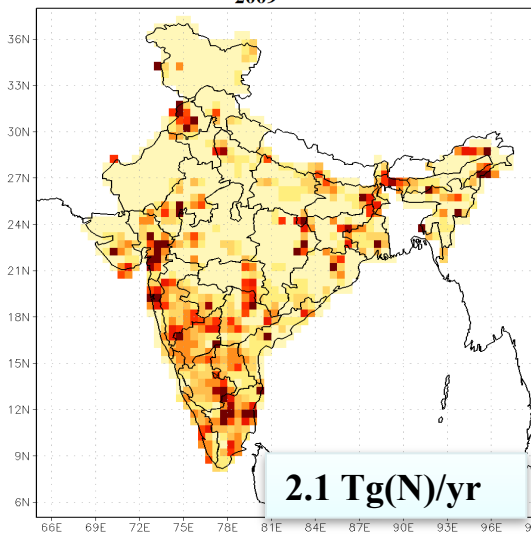
2007



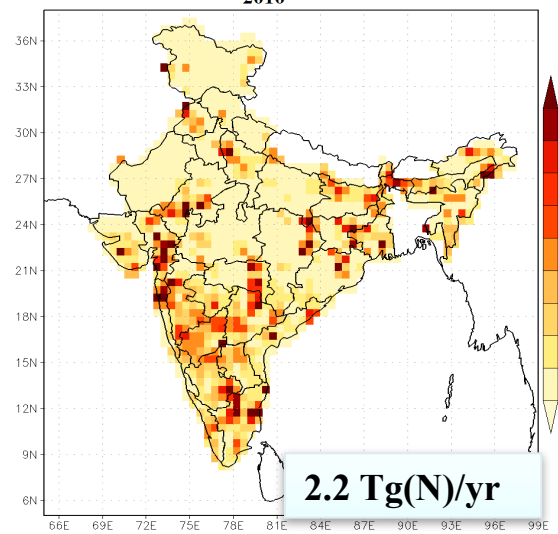
2008



2009



2010



Spatial distribution of annual NO_x emissions from optimized Top-Down inventory (emission unit is 10^{11} NO molecules/cm²/s) for 2005, 2006, 2007, 2008, 2009, 2010 years.

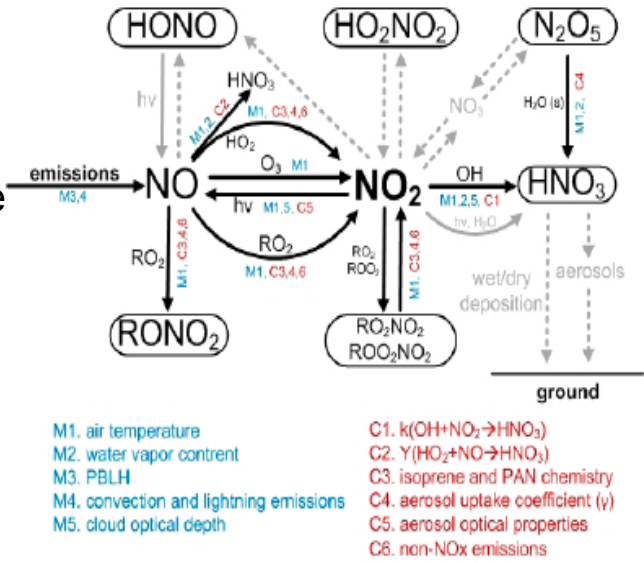
Sources of Uncertainties in Top-Down Inventory:

1. Uncertainty in satellite retrievals

OMI filtering error,
 Air Mass Fraction (clod fraction, cloud pressure, NO2 Profile shape, aerosols calculation,
 Stratospheric corrections, etc)

2. Model Uncertainties

- Error in model metrology
- Model profile shape
- Simulation of fate of Nitrogen emitted into the atmosphere (chemical mechanism and heterogeneous aerosol NOx reactions)
- dry and wet deposition



Conclusions

Every advantage has its disadvantage

Fig. 1. Tropospheric chemistry involving NO_x and impacts of meteorological and chemical parameters evaluated in the present study. Processes shown in solid grey arrows are discussed without sensitivity simulations. Processes shown in dashed grey arrows are not discussed explicitly. Note that PBL mixing and convection affect vertical distributions of NO_x and related species. Heterogeneous uptake on aerosols depends on the amount of aerosol surfaces as well. Evaluation on the RONO₂ pathway is focused mainly on isoprene nitrates. Clouds and water vapor have indirect influences on radicals through effects on solar radiation.



**Access to information on emissions:
The ECCAD Emissions Database**

**ECCAD = Emissions of Chemical Compounds and
Compilation of Ancillary Data**

Goal of ECCAD

Provide access to surface emissions and associated ancillary data to support many projects:

- International projects such as:

 - **GEIA** (Global Emission Inventories Activity)

 - **IGAC** (International Global Atmospheric Chemistry Project)

 - **CCMI, AEROCOM, , CMIP5/CMIP6, etc.**

- European research projects such as:

 - **MACC-II** (Monitoring of the Atmospheric Composition and Climate)

 - **PANDA** (Forecasting Air Pollution in China)

 - **ChArMEx** (The Chemistry-Aerosol Mediterranean Experiment)

 - **ECLIPSE, PEGASOS, PANDA, ACCESS, etc.**

Website: pole-ether.fr/eccad

From the ECCAD website

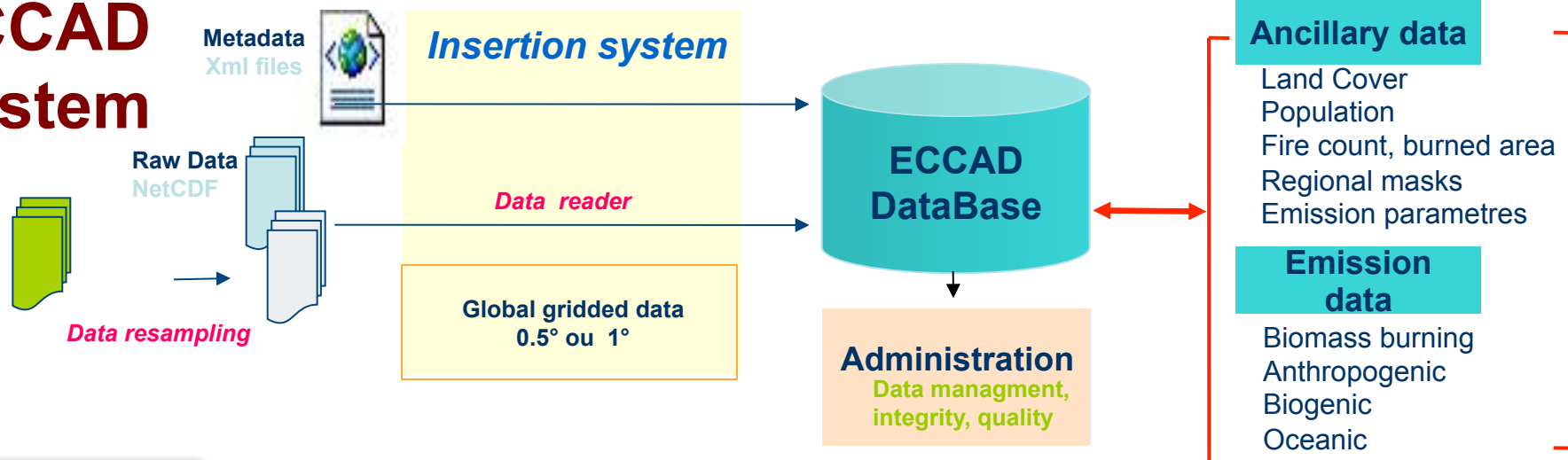
The ECCAD database includes currently a large diversity of datasets, which provide global and regional surface emissions for a large set of chemical compounds. All the data are at a 0.5x0.5 or 1x1 degree resolution. ECCAD provides detailed metadata for each dataset, including information on complete references and methodology, and links to the original inventories.

Several tools are available for the visualization of the data, for computing global and regional totals and for an interactive spatial and temporal analysis. The data can be downloaded as NetCDF CF-compliant files.

NetCDF-CF = Climate and Forecast (CF) Metadata Convention

See <http://cfconventions.org/>

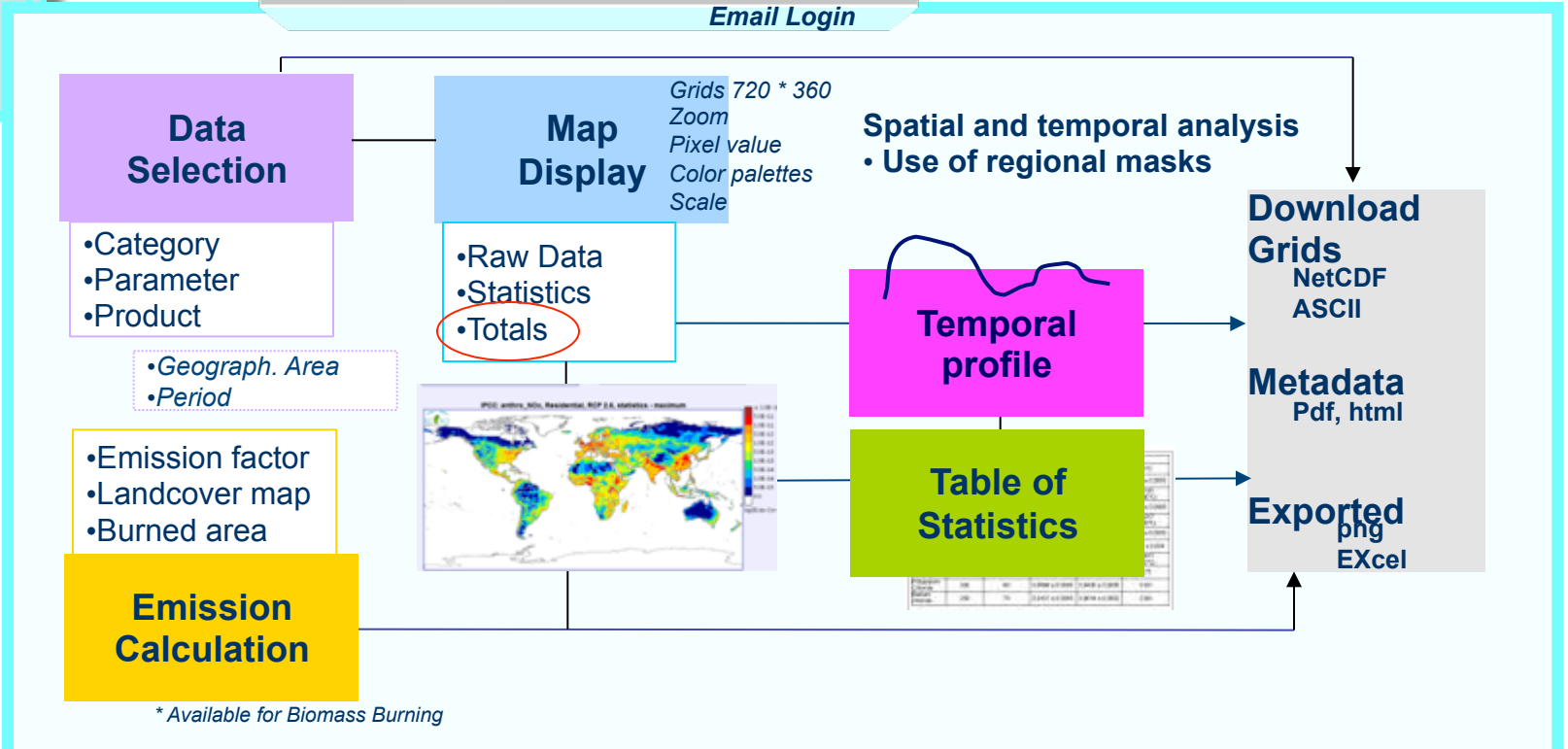
ECCAD System



Web Site
Documentation

Web Application

Email Login



ECCAD = Emissions of Atmospheric Compounds and Compilation of Ancillary Data

THE ECCAD - THE GEIA DATABASE Claire Granier

Emissions of atmospheric Compounds & Compilation of Ancillary Data

[Data Catalogue](#)
[Data Visualization](#)
[Emission Calculation](#)

Emissions Inventories

■ Anthropogenic
 ■ Biomass burning
 ■ Natural

GLOBAL INVENTORIES

- MACCity ACCMIP RCPs PEGASOS_PBL EDGARv4.2 EDGARv3.2FT2000 RETRO
- ECLIPSE_GAINS_4a Junker-Liousse HYDE1.3 Andres_CO2_v2013 AMAP_Mercury
- GFASv1.0 GFED3 GFED2 GICC AMMABB
- MEGANv2 MEGAN-MACC MEGANv2-CH3OH
- ■ GEIAv1 POET

Developed for ongoing projects

- IS4FIRES
- GUESS-ES
- ■ CCMi

REGIONAL INVENTORIES

- TNO-MACC-II (Europe) TNO-MACC (Europe)
- EMEP (Europe) Assamoi-Liousse (Africa)
- India_NOx (India) SAFAR-India (India)
- REAS (Asia)

Developed for ongoing projects

- ■ ChArMEx (Mediterranean)

Ancillary Datasets

LAND COVER

- UMD CLM3 GLC2000

FIRES

- WFA GBA2000 Geoland2_BAv1_Africa

POPULATION

- GPW3_Population















GEOGRAPHICAL INFORMATION

- GPW3 Region_IMAGE2.4 Pixel_Area


ECCAD v6.6.3 ©2006-2013 CNRS/SEDOO

ECCAD is the GEIA emissions database

REGIONAL INVENTORIES (7)

TNO-MACC-II Europe 2013	2003 - 2009	yearly	Anthropogenic	0.5°		
TNO-MACC Europe 2009	2003 - 2007	Yearly	Anthropogenic	0.5°		
EMEP Europe 2007	1980 - 2020	Yearly	Anthropogenic	0.5°		
Assamoi-Liousse Africa 2012	2005 - 2030	Decadal	Total	0.5°		
India_NOx India 2012	2005	Yearly	Anthropogenic	0.5°		
SAFAR-India India 2012	1991 - 2011	Decadal	Anthropogenic	1°		
REAS Asia 2007	1980 - 2020	Yearly	Anthropogenic	0.5°		

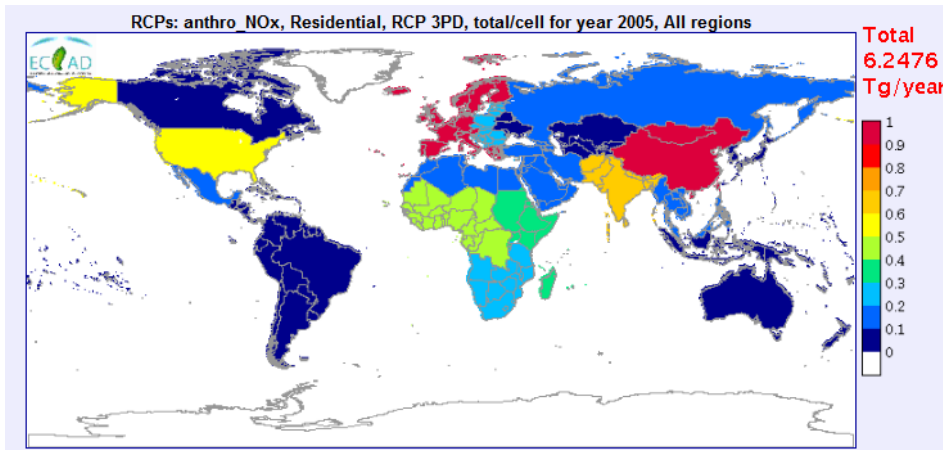
REGIONAL INVENTORIES DEVELOPED FOR ONGOING PROJECTS (1)

ChArMEx Mediterranean 2012	2000	Varied	Anthropogenic Biomass burning Biogenic Oceanic Volcanic	0.25/0.5/1°		
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Many different inventories, including global and regional datasets
 Only gridded inventories can be inserted currently in the database
 New version under development (Thanks to EPA)
 → data at any resolution can be included

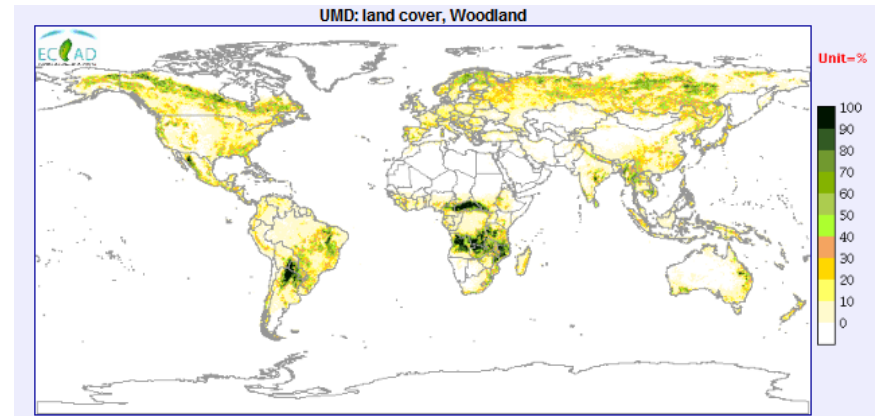
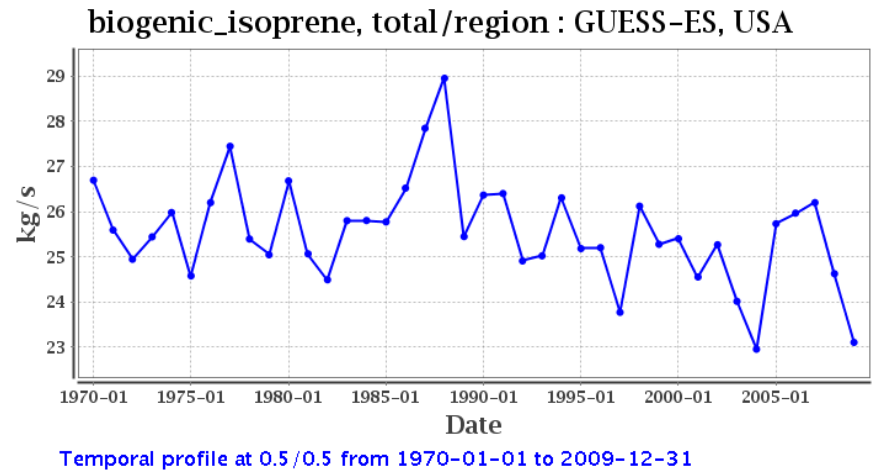
ECCAD tools

Total emitted for different regions



Ancillary data

Temporal variation for a country over a period of time



Currently under development: comparisons of maps with algebraic calculations, scattered plots

<http://www.temis.nl/airpollution/no2.html>

https://disc.gsfc.nasa.gov/Aura/data-holdings/OMI/omno2_v003.shtml

http://projects.knmi.nl/omi/research/product/product_generator.php?&info=intro&product=

<http://www.tropmet.res.in/emission/>

http://eccad.sedoo.fr/eccad_extract_interface/JSF/page_map_instituts.jsf

Estimate Surface $PM_{2.5}$ From Satellite AOD

AOD is related to $PM_{2.5}$

...but how to separate surface aerosol from column concentrations?

$PM_{2.5}$ / AOD ratio is a function of:

- vertical structure
- aerosol type/hygroscopicity
- meteorology

Multiple approaches:

- surface monitors calibration
- empirical relations
- model output

Approach

satellite-based measurements of *aerosol optical depth* to $PM_{2.5}$ using a global/regional chemical transport model

Following *Liu et al.*, 2004:

$$\text{Estimated } PM_{2.5} = \eta \cdot \tau$$

Combined MODIS/MISR
Aerosol Optical Depth

Global/Regional CTM

- AOD/ $PM_{2.5}$ relationship (η)
- *A priori* values

Satellites provide:

- High resolution ($0.1^\circ \times 0.1^\circ$)
- Observational constraint
- Column-integrated value

η

- vertical structure
- aerosol type
- meteorological effects
- meteorology
- diurnal effects

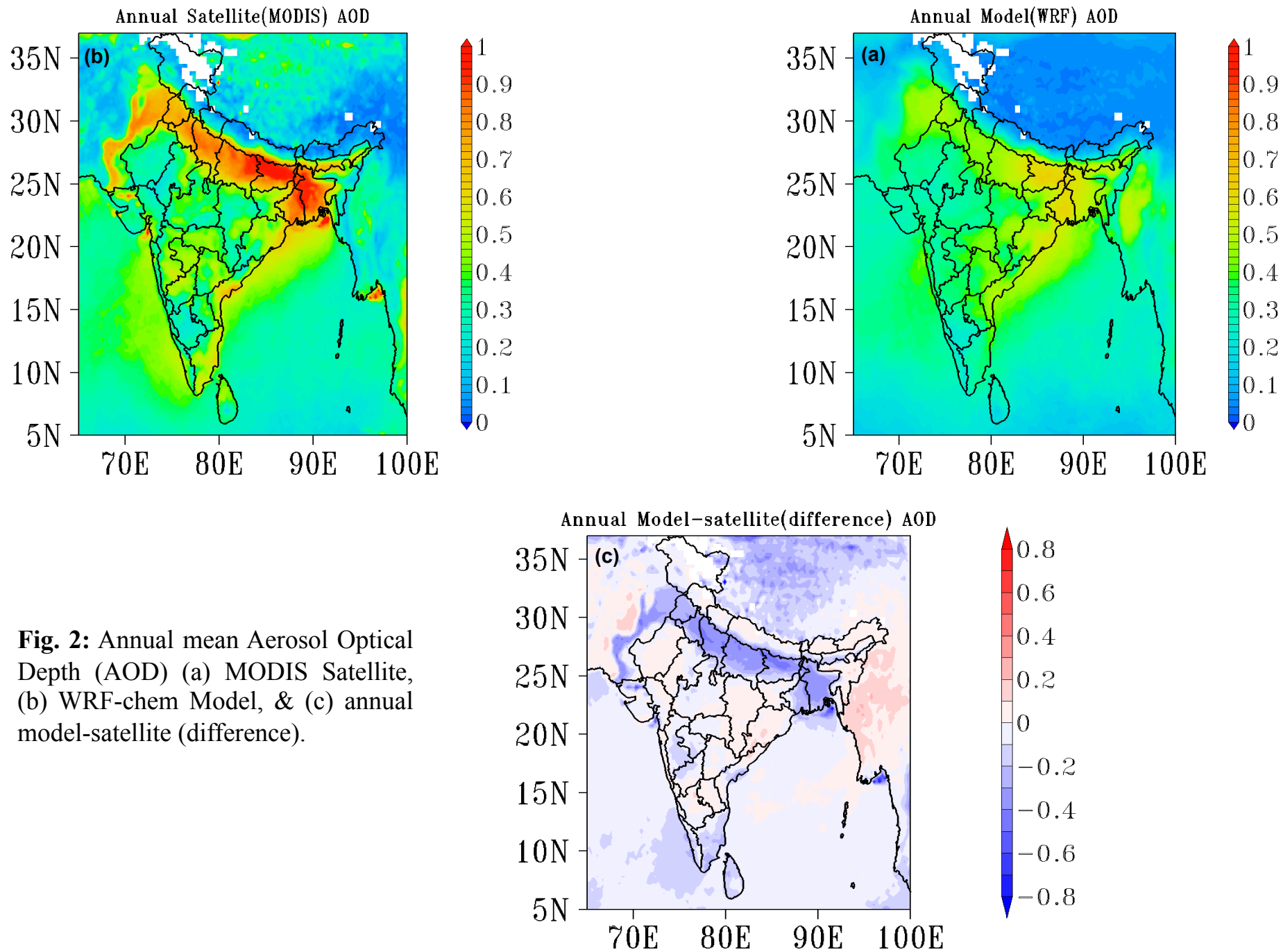


Fig. 2: Annual mean Aerosol Optical Depth (AOD) (a) MODIS Satellite, (b) WRF-chem Model, & (c) annual model-satellite (difference).

$$\text{Estimated PM}_{2.5} = \eta \cdot \tau$$

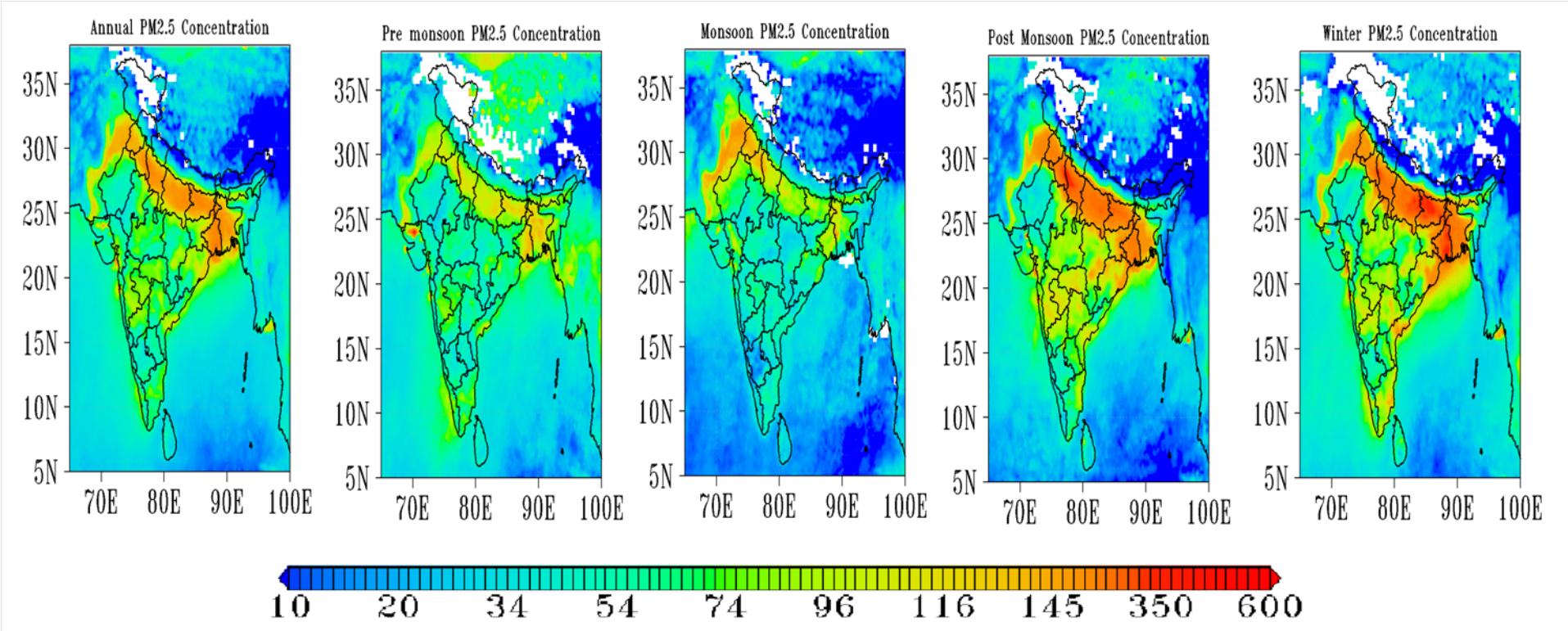


Fig. 7: Estimated PM_{2.5} values for annual and seasonal during the study period.

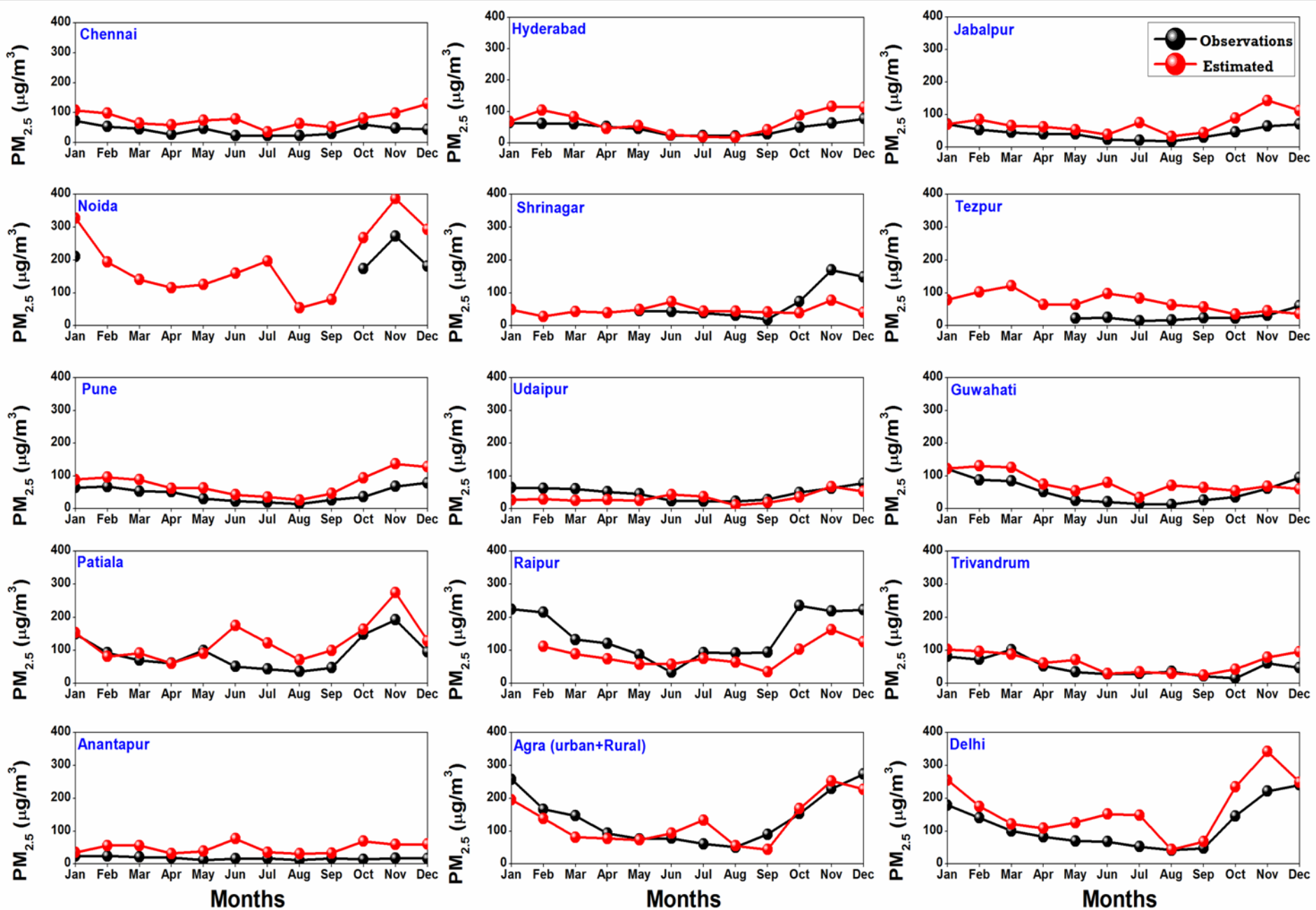


Fig. 3: Variability of monthly mean satellite derived (red), and observed (black) surface $PM_{2.5}$ (in $\mu g/m^3$) over 15 monitoring locations.

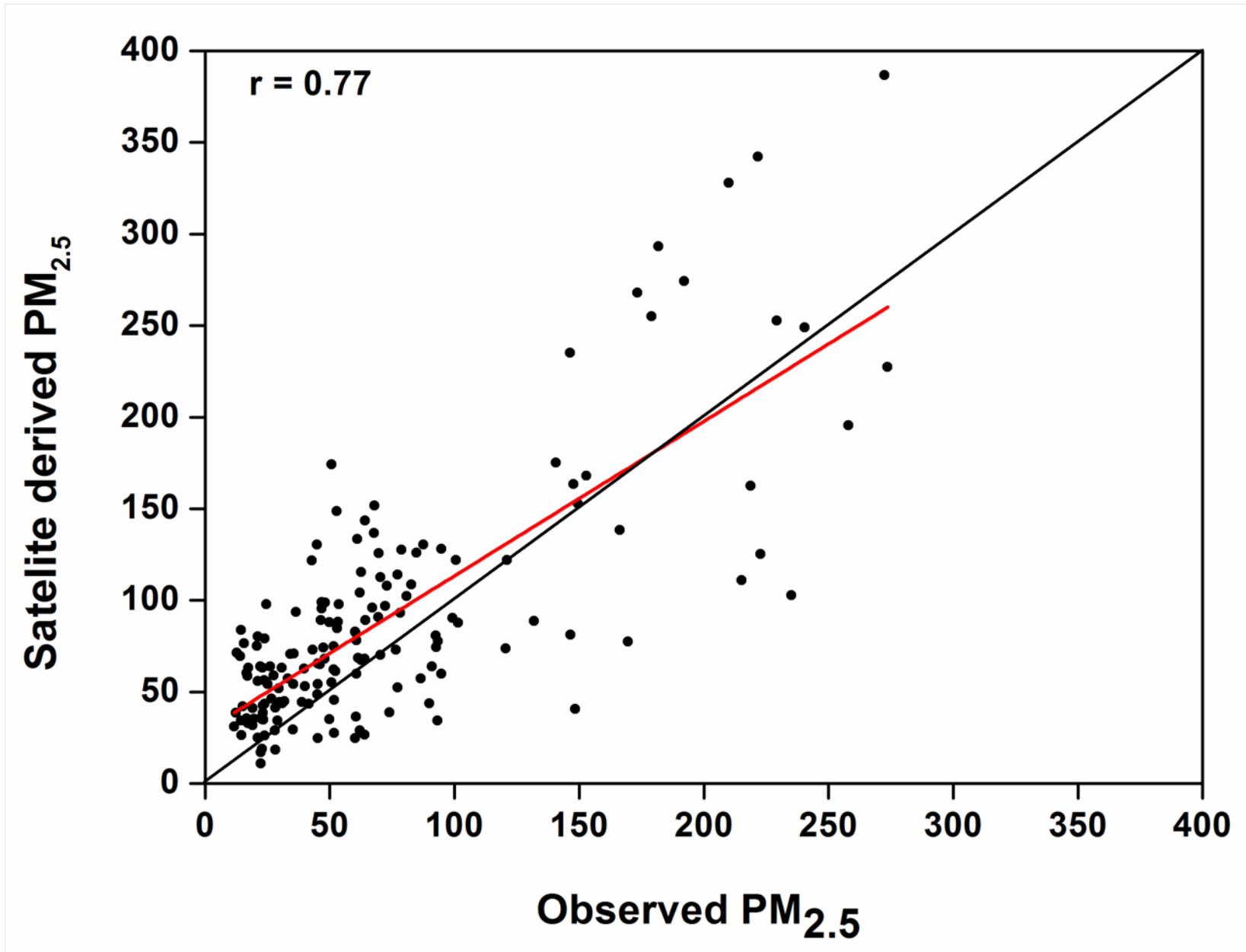


Fig. 4b: Correlation coefficients between observed (monthly mean of 15 stations) and estimated PM_{2.5} concentrations

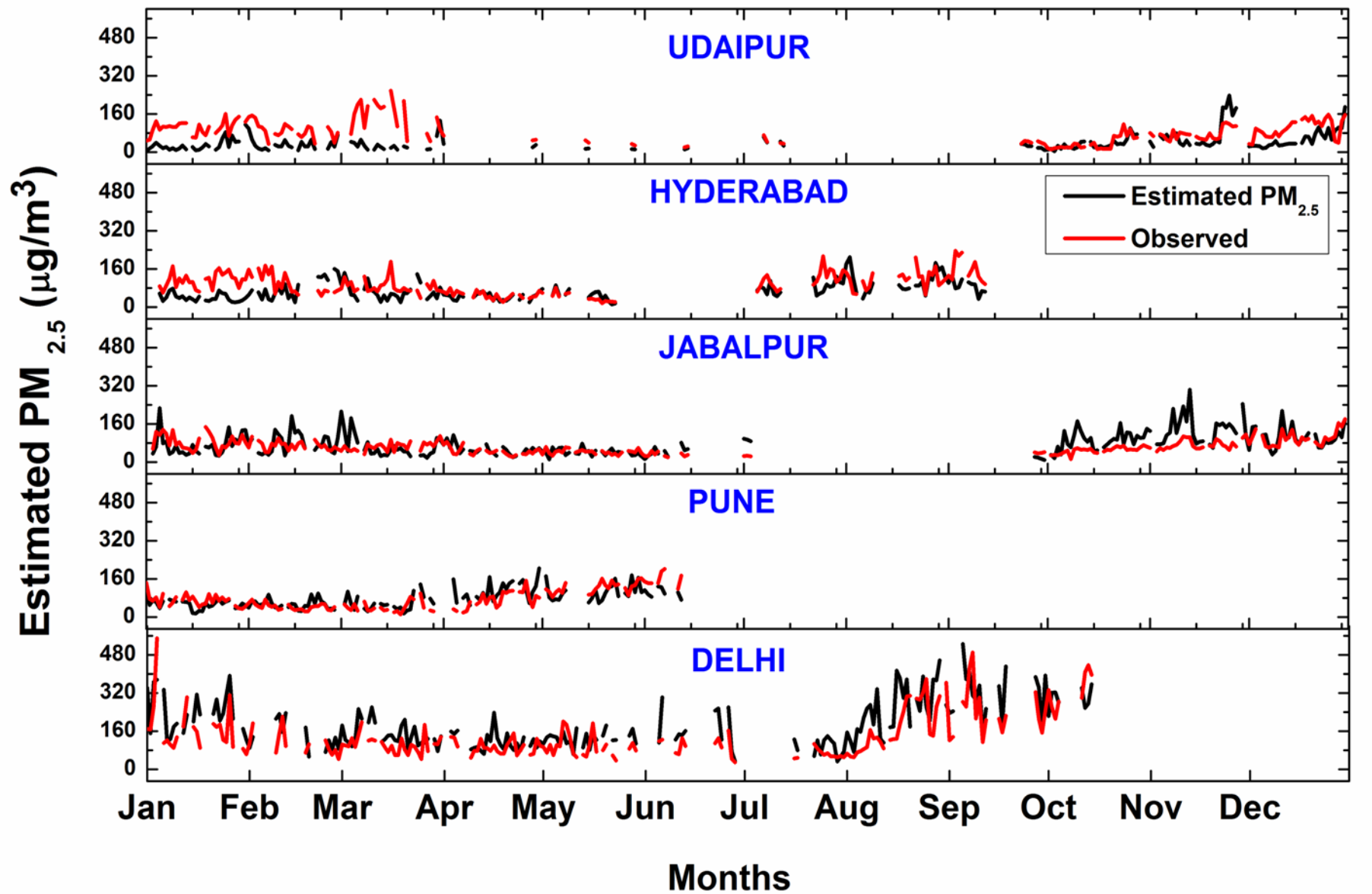


Fig. 5: Variability in observed and estimated PM_{2.5} concentrations over Delhi, Pune, Jabalpur, Hyderabad and Udaipur

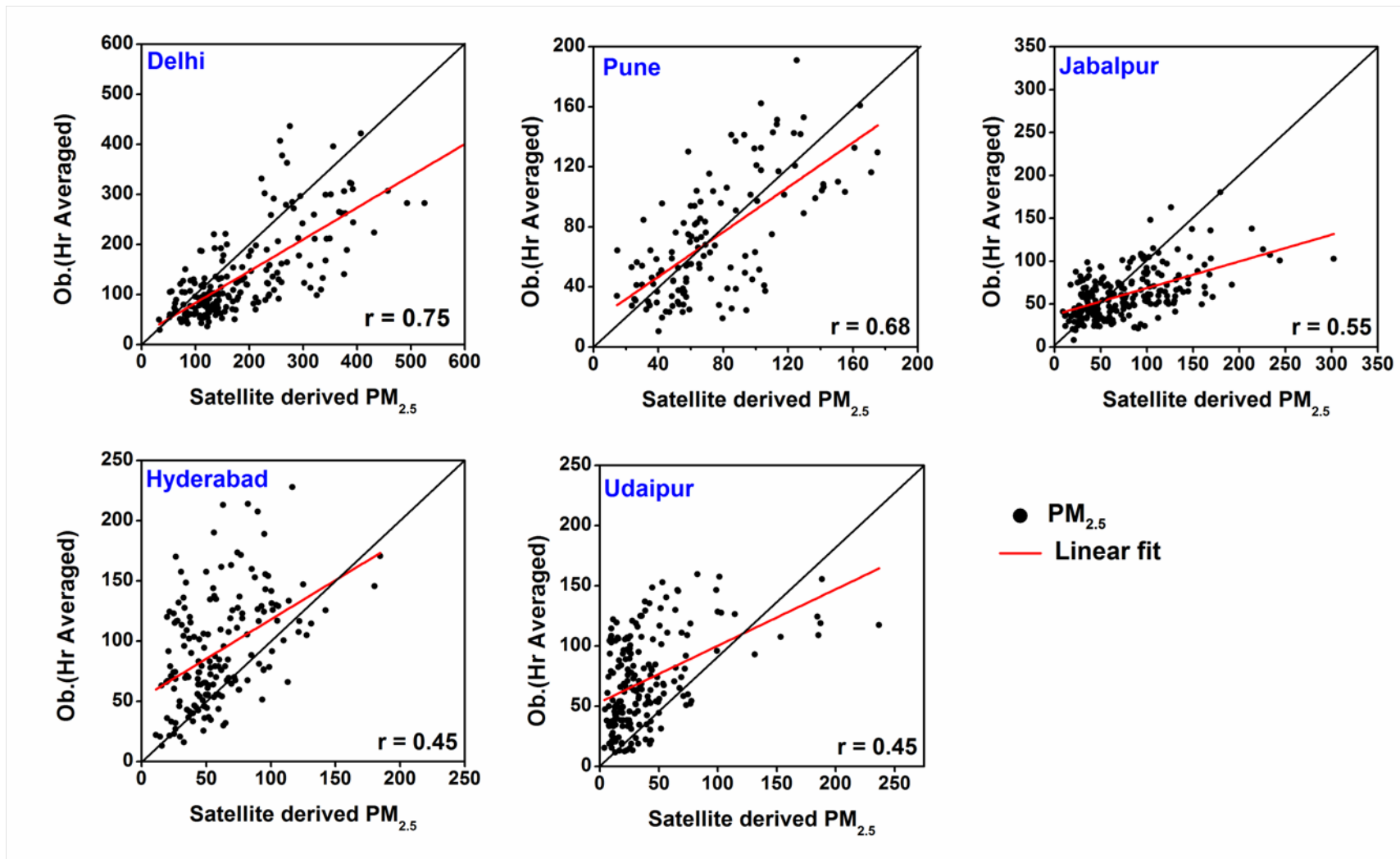


Fig. 6: Correlation coefficients values between observed and Satellite derived PM_{2.5} concentrations over Delhi, Pune, Jabalpur, Hyderabad and Udaipur

Emission Exercise ---- Transportation Sector (Delhi)

Categories of vehicles

1. Two stroke two-wheeler (2W2S)
2. Four stroke two-wheeler (2W4S)
3. Two stroke three-wheeler (3W2S)
4. Four stroke three-wheeler (3W4S)
5. Four wheeler gasoline (4WG)
6. Four wheeler diesel (4WD)
7. Heavy Duty Diesel Low sulfur (HDDLS)
8. Heavy Duty Diesel High sulfur (HDDHS)

Pollutants for which Emission Factors have been determined

1. Carbon dioxide (CO₂)
2. Carbon monoxide (CO)
3. Oxides of Nitrogen (NO)

Computed Mass Emission Factors for Different Vehicles

Species		2W2S	2W4S	3W2S	3W4S	4WG	4WD	HDDLS	HDDHS
FC	g/km	11.0	9.7	22.1	25.9	84.3	92.7	195.2	195.2
	G/hr	254.9	225.2	511.2	599.8	1576.5	1733.2	3649.9	3649.9
CO ₂	g/km	26.6	28.3	60.3	78.5	223.6	208.3	515.1	515.2
	G/hr	617.0	655.6	1397.2	1817.1	4181.3	3896.4	9633.3	9634.8
CO	g/km	2.0	1.4	5.25	2.0	24.8	2.0	4.7	4.7
	G/hr	46.4	33	121.6	46.9	462.9	36.8	88.4	87.4
NO	g/km	0.8	1.4	1.2	2.0	3.3	116.9	354.3	405.3
	G/hr	20.5	32.9	28.1	46.0	62.4	2185.5	6626.6	7579.2

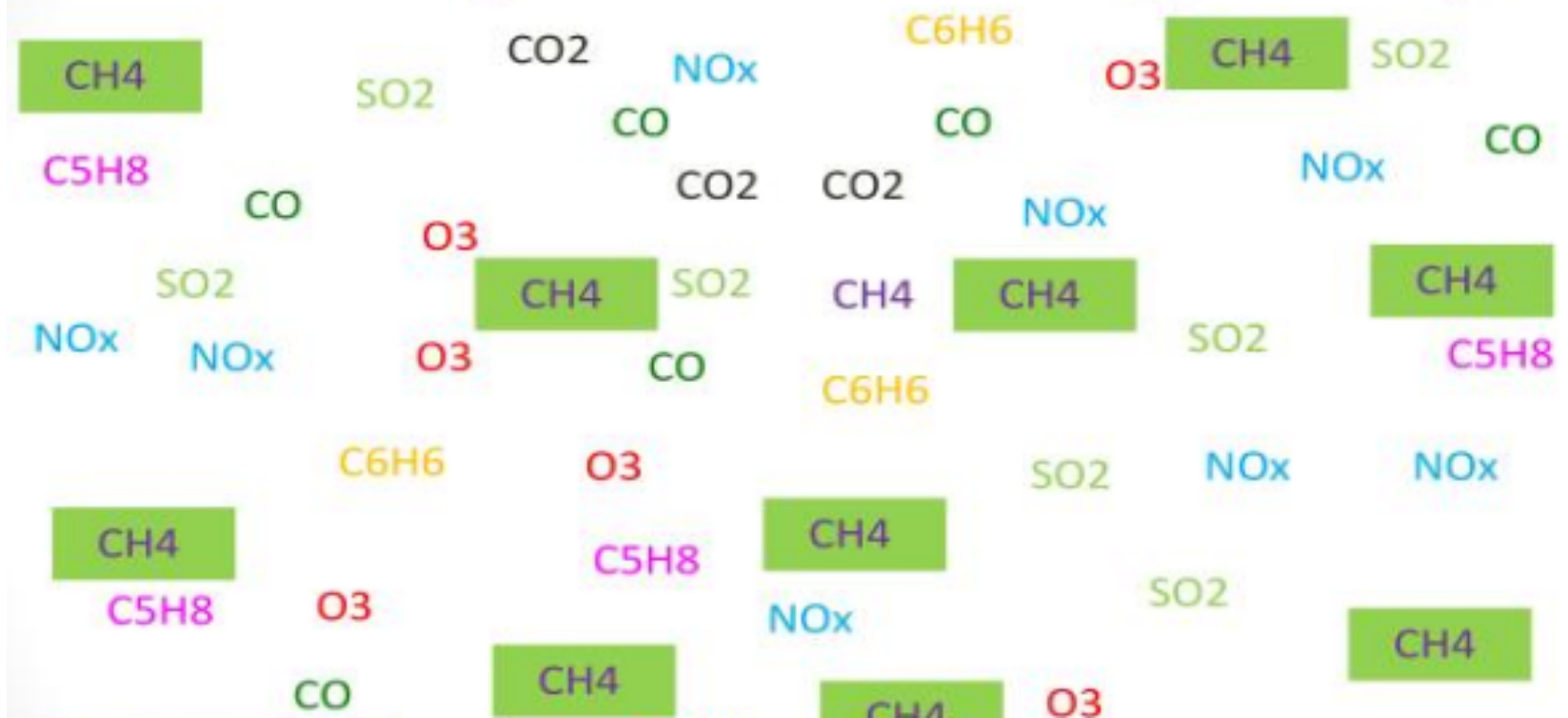
Area of Delhi: 36 km x 36 km

Number of total vehicles : 4.5 million

- | | | |
|----|--|-----------|
| 1. | Two stroke two-wheeler (2W2S): 10% | 25km /day |
| 2. | Four stroke two-wheeler (2W4S) : 10% | 25km /day |
| 3. | Two stroke three-wheeler (3W2S): 15% | 50km /day |
| 4. | Four stroke three-wheeler (3W4S):5% | 50km /day |
| 5. | Four wheeler gasoline (4WG): 30% | 25km /day |
| 6. | Four wheeler diesel (4WD): 10% | 25km /day |
| 7. | Heavy Duty Diesel Low sulfur (HDDLS): 10% | 10km /day |
| 8. | Heavy Duty Diesel High sulfur (HDDHS): 10% | 10km /day |

Calculate total **annual emissions from the transportation sector**

Challenge: How to measure one compound among many?



Category	Parameters	Instruments
Ground and surface properties	Temperature and water profile in the ground Ground heat flux	Soil moisture sensors
Surface layer meteorology	Temperature, humidity, pressure, wind speed, wind direction at 1m, 2m, 5m, 10m, turbulence and precipitation	All in one weather station, 10m vertical mast, IMD-SAFAR AWS network, sonic anemometers
Radiation fluxes	Downwelling shortwave (SW) direct, diffuse, and global long wave (LW) irradiance	Sky Radiometer, Net radiometer, pyranometers
Atmospheric profiles	Temperature, humidity, pressure, wind speed, wind direction	Microwave radiometer, Tethersonde instruments, Radio-sonde (IMD)
Aerosols and fog optical properties	Visibility, fog and cloud base height, absorption coefficient, scattering coefficient, aerosol optical depth	RVR, fog detector, Photoacoustic Extinctionmeter (PAX- aerosol absorption and scattering), Neflometer, sun-photometers, Aethalometer
Aerosols and fog microphysics	Aerosol particle size distribution, aerosol particle counter, Condensation Particle Counter/ Condensation droplet counter, CCN	SMPS, CPC/CDP, CCN counter, GRIMM
Aerosol/gas chemistry	Fog collector, Particle in Liquid Sampler (PILS), EC/OC analyser, VOCs, Aérosol filter collector, SO ₂ gas analyzer , PM ₁ , PM _{2.5} , PM ₁₀	Fog Collector, PILS, PTRMS EC/OC analyzer, SO ₂ analyzer, PM analyzers

In brief, tropospheric NO₂ column is derived in three main steps involving the calculation of

(1) slant column (using Differential Optical Absorption Spectroscopy (DOAS) approach in the 405–465 nm spectral window),

(2) tropospheric slant column (using modeling/ assimilation approach),

(3) tropospheric vertical column (using air mass factor—AMF).

